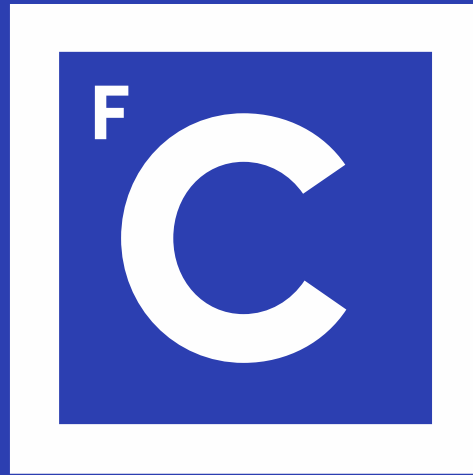




Ciências ULisboa

Faculdade
de Ciências
da Universidade
de Lisboa

Eng Energy & Environment



Environmental Impact and Life Cycle Analysis

Professor: Carla Silva (camsilva@ciencias.ulisboa.pt)

Assignments

Weekly basis



VS





Case study

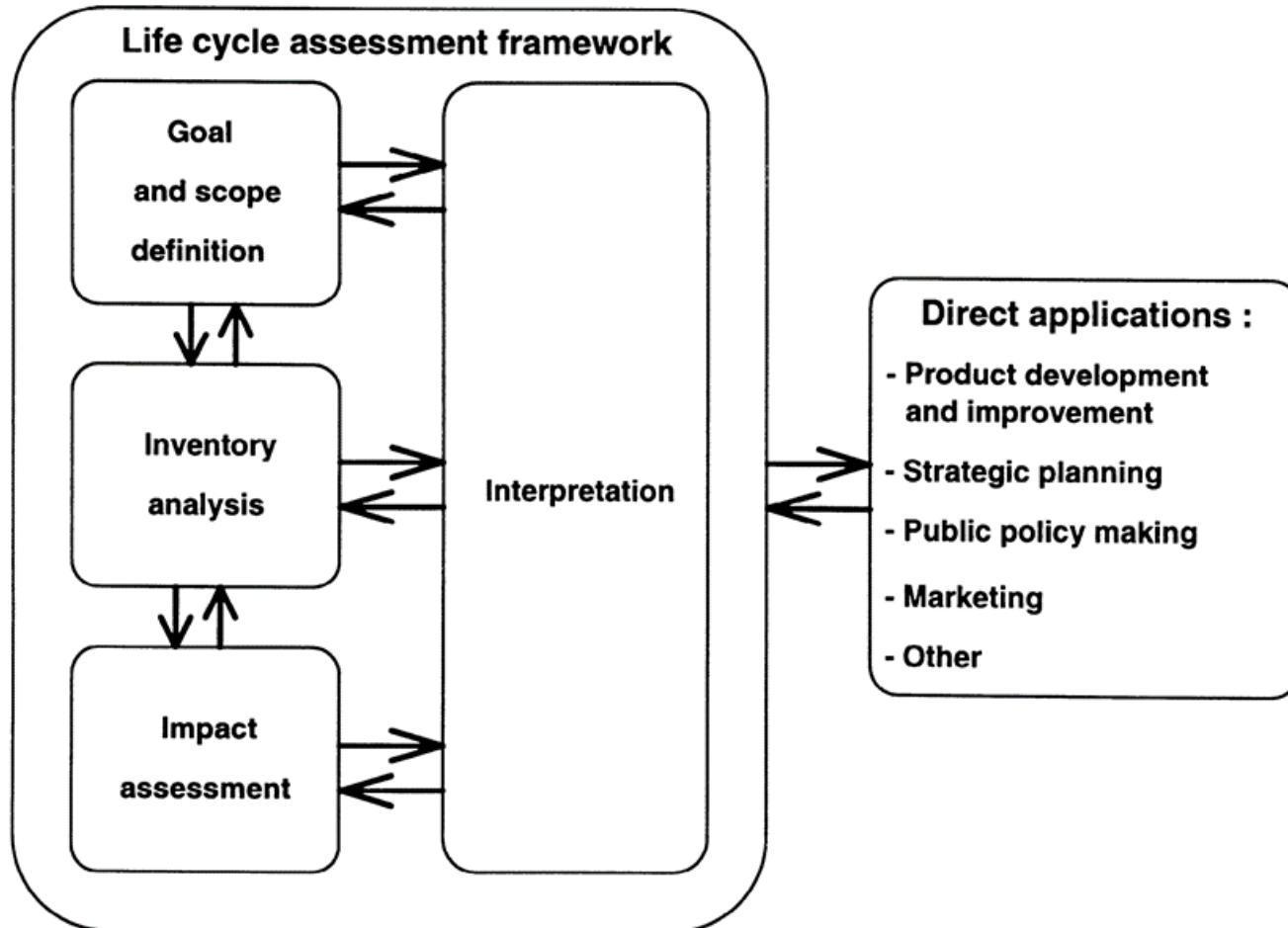


Figure 1 : Phases of an LCA

Âmbito/Scope: comparar lavar cabelo com líquido ou sólido considerando

FU-Functional unit/ unidade funcional: 1 lavagem

Materiais da embalagem (extração, produção, fim de vida)

Água da lavagem (uso)

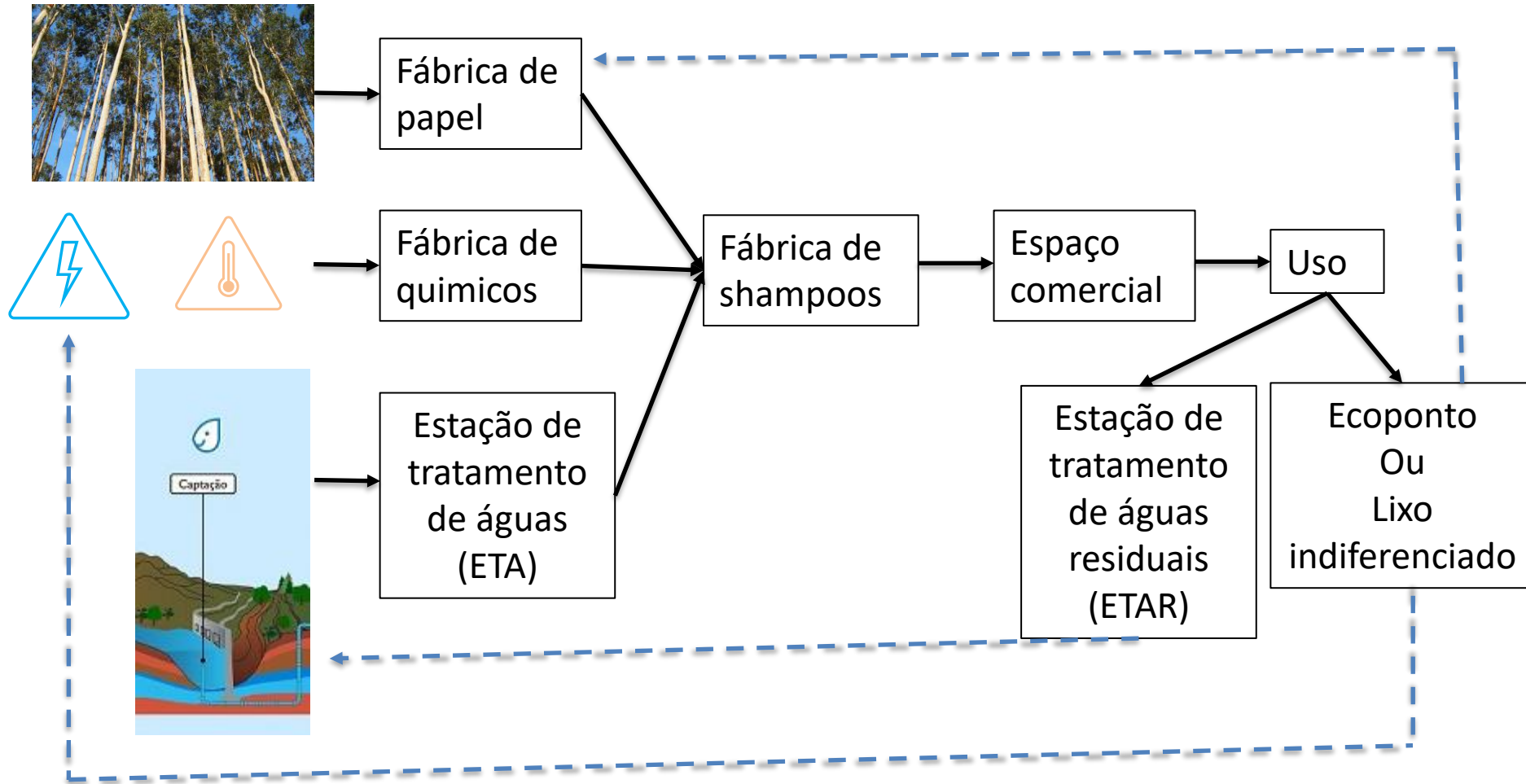
Gás natural e eletricidade (uso)

Químicos constituintes (extração, produção)

O que considerar??

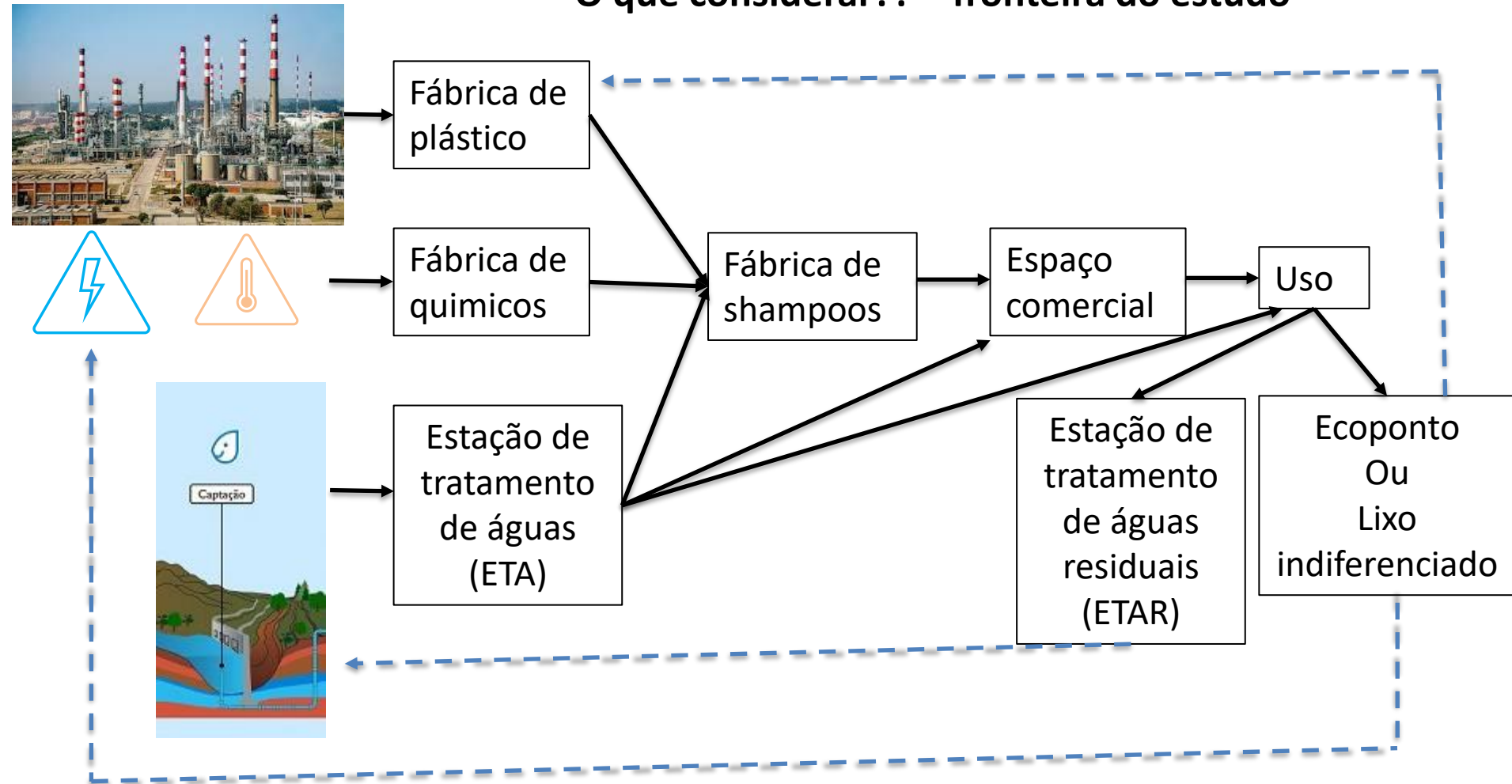
Case study

O que considerar?? – fronteira do estudo



Case study

O que considerar?? – fronteira do estudo



GHG emissions

EU ETS benchmark Heat

47.3 gCO_{2eq}/MJ

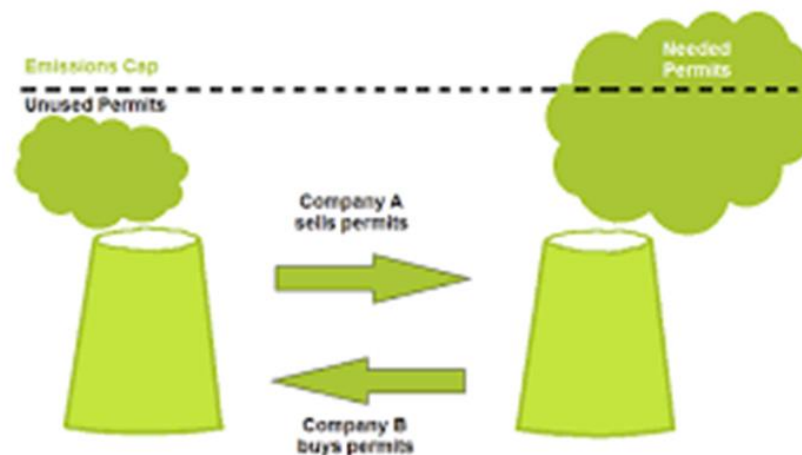


EU ETS benchmark Electricity 2030

150 gCO_{2eq}/kWh

EU ETS benchmark Electricity 2050

0 gCO_{2eq}/kWh



GWP_{100years}



Emission factor database gCO_{2eq}/ unit

Construction materials
“grey materials”

Chemicals

Fuels

Heat/vapour

Electricity



GREENHOUSE
GAS PROTOCOL



Department
for Environment
Food & Rural Affairs



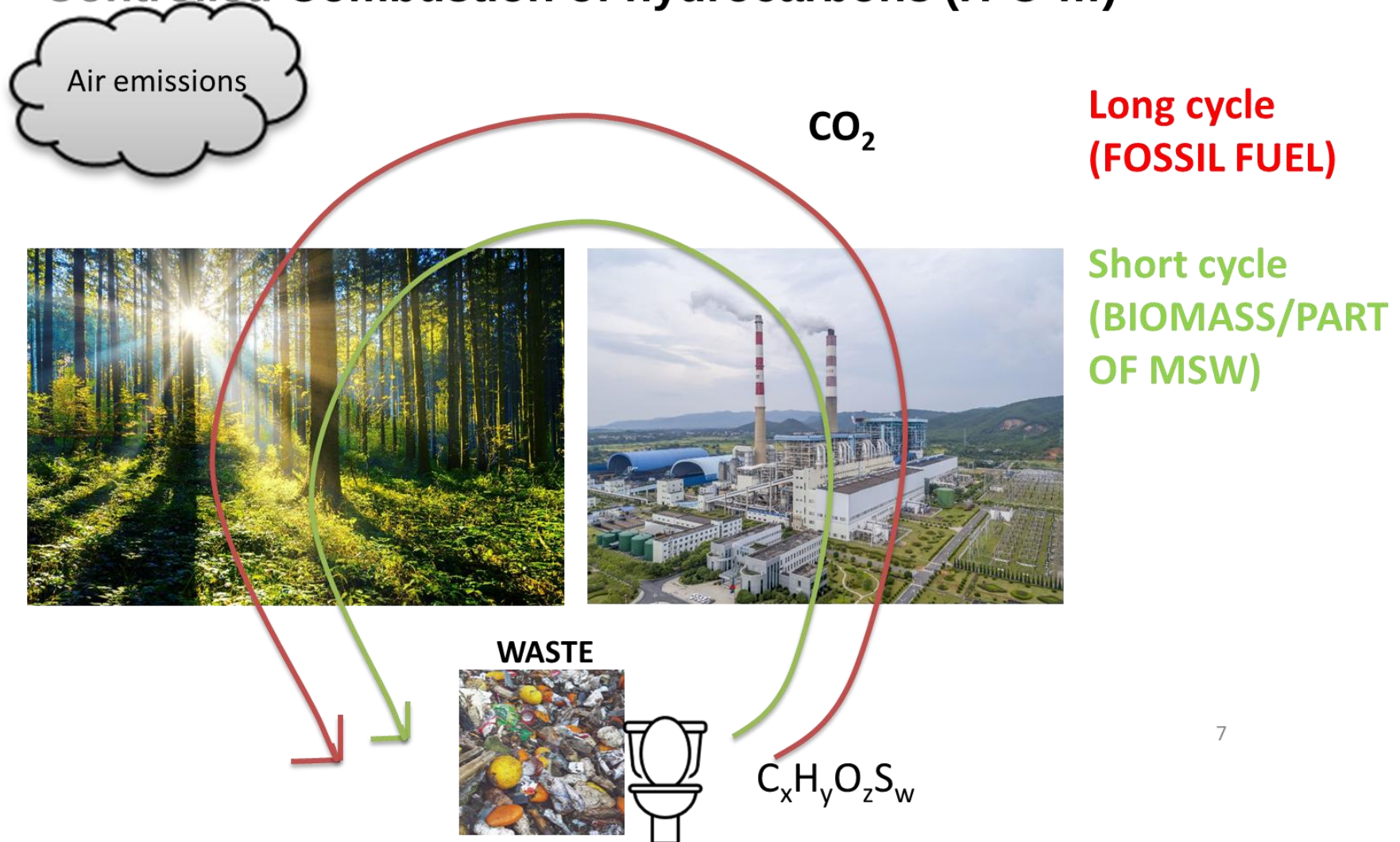
2006 IPCC Guidelines for National Greenhouse
Gas Inventories

39

Scientific literature DOI:.....

Case study

Controlled Combustion of hydrocarbons (H-C-...)



7

$$mCO_{2eq} = m_{CO_2} * 1 + m_{CH_4} * EQ_{CH_4} + m_{N_2O} * EQ_{N_2O} + \dots$$

GWP_{100years}

AR = Assessment report IPCC

EQ = Equivalence

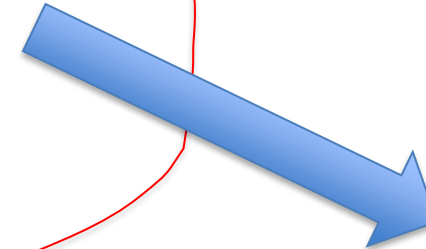
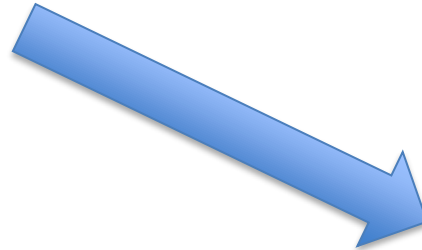
Substance	AR1 (1990)	AR2 (1995)	AR3 (2001)	AR4 (2007)	AR5 (2013)
Carbon dioxide, fossil (CO ₂)	1	1	1	1	1
Methane, fossil (CH ₄)	21	21	23	25	28
Methane, biogenic (CH ₄)	18.25	18.25	20.25	22.25	25.25
Dinitrogen monoxide (N ₂ O)	290	310	296	298	265
HCFC-141b	440	-	700	725	782
HFC-134a	1200	1300	1300	1430	1300
HCFC-22	1500	-	1700	1810	1760
HCFC-142b	1600	-	2400	2310	1980
CFC-11	3500	-	4600	4750	4660
CFC-12	7300	-	10600	10900	10200
Sulfur hexafluoride	-	23900	22200	22800	23500

Case study

Materiais da embalagem (extração, produção, fim de vida)

Água da lavagem (uso)

Químicos constituintes



GREENHOUSE
GAS PROTOCOL

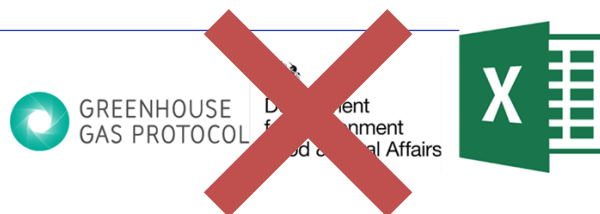


Department
for Environment
Food & Rural Affairs



Base de dados??

Químicos



Químico	Valor Cradle-to-Gate	Referência da base de dados
Sodium Lauryl Sulfate (SLS)	1.63 Ton CO ₂ eq/ton	Environmental Fact Sheet (#5) C12-14 and C12-15 Sodium Alkyl Sulphate (C12-14 mix AS) oleo/petrochemical anionic surfactant https://www.erasm.org/
Cocamidopropyl betaine	1.63 Ton CO ₂ eq/ton	Environmental Fact Sheet (#28) C8-18 Alkyl Amidopropyl Betaine (CAPB) oleo/petrochemical amphoteric surfactant https://www.erasm.org/
Cocamide Diethanolamine	-0.88 Ton CO ₂ eq/ton	Environmental Fact Sheet (#16) Cocamide Diethanolamine (CDEA) oleochemical non-ionic surfactant https://www.erasm.org/
Propylene glycol	4.67 Ton CO ₂ eq/ton	OpenLCA Energies 2020, 13, 5653; doi:10.3390/en13215653
Titanium dioxide	1.43 Ton CO ₂ eq/ton	2006 IPCC Guidelines for National Greenhouse Gas Inventories (Volume 3, Chapter 3, Table 3.9) https://www.ipcc-nggip.iges.or.jp/EFDB/find_ef.php?ipcc_code=2.B.6&ipcc_level=2
HCL	0.89 Ton CO ₂ eq/ton	Winipeg -canada
Water	0.149 kg CO ₂ eq/m ³	UK DEFRA
Tallow Oil	3.05 kg CO ₂ eq/kg	Life Cycle Analysis of Greenhouse Gas Emissions from Biosynthetic Base Oil (BBO) compared to Poly-Alpha Olefin (PAO) Base Oil Prepared for Biosynthetic Technologies Prepared by Dustin Mulvaney, Ph.D., EcoShift Consulting February 3, 2014



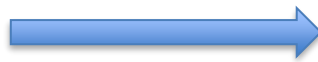
Peso embalagem: 6 g

Contéudo: 60g

1.67 g/lavagem

$$\frac{60 \text{ g}}{1.67 \text{ g/lavagem}} = 36 \text{ lavagens}$$

6 g de material de
embalagem



0.17 g de material de
embalagem por lavagem

Material embalagem: e.g. mistura papel e cartão

Fator de emissão:

$881.19 \text{ kg CO}_{2\text{eq}}/\text{Ton (0\%reciclado)} + 1041.804 \text{ kgCO}_{2\text{eq}}/\text{Ton (aterro)} = 1923 \text{ kg CO}_{2\text{eq}}/\text{Ton}$

$731.28 \text{ kg CO}_{2\text{eq}}/\text{Ton (100\% reciclado)} + 21.294 \text{ kg CO}_{2\text{eq}}/\text{Ton (reciclagem)} = 753 \text{ kg CO}_{2\text{eq}}/\text{Ton}$

UK Government GHG Conversion Factors for Company Reporting

Material use

[Index](#)

Activity	Material	Unit	Primary material production	Re-used	Open-loop source	Closed-loop source
			kg CO ₂ e	kg CO ₂ e	kg CO ₂ e	kg CO ₂ e
Paper	Paper and board: board	tonnes	821.23			718.54
	Paper and board: mixed	tonnes	881.19			731.28
	Paper and board: paper	tonnes	919.4			739.4

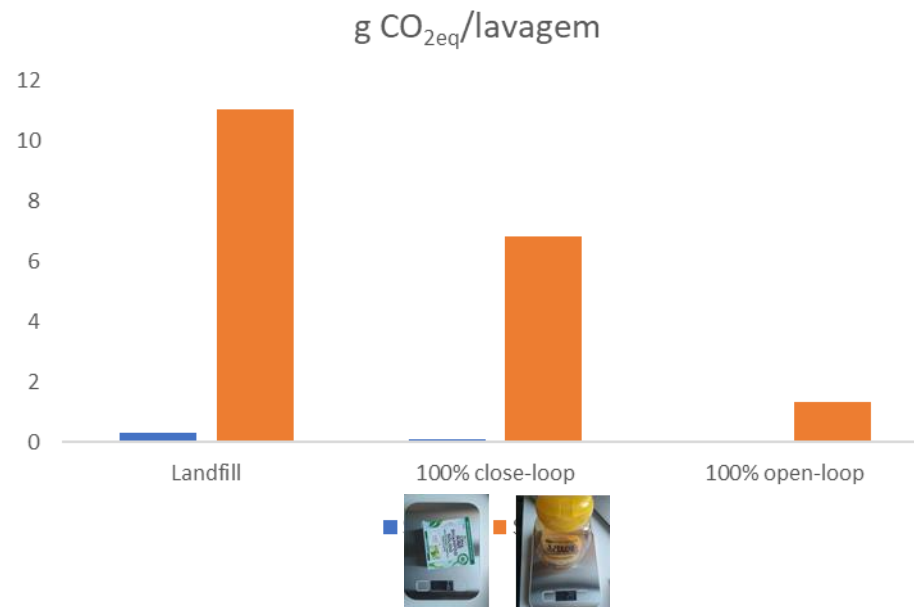
UK Government GHG Conversion Factors for Company Reporting

Waste disposal

[Index](#)

Activity	Waste type	Unit	Re-use	Open-loop	Closed-loop	Combustion	Composting	Landfill	Anaerobic digestion
			kg CO ₂ e	kg CO ₂ e	kg CO ₂ e	kg CO ₂ e	kg CO ₂ e	kg CO ₂ e	kg CO ₂ e
Paper	Paper and board: board	tonnes			21.294	21.294	8.951	1,041.804	
	Paper and board: mixed	tonnes			21.294	21.294	8.951	1,041.804	
	Paper and board: paper	tonnes			21.294	21.294	8.951	1,041.804	





Case study

Uso de água: **por defeito 15 l agua/lavagem**

ETA

A	B	C	D	E	F	G	H	I	J	K
UK Government GHG Conversion Factors for Company Reporting										
Water supply										
Index										
Company J multiplies the water used (cubic metres (m ³)) by the appropriate year's conversion factor called 'water supply' to produce its emissions.										
Activity	Type	Unit	kg CO ₂ e							
Water supply	Water supply	cubic metres	0.149							
		million litres	149.0							
For information about how the conversion factors have been derived, please refer to the Methodology paper' that accompanies the conversion factors.										

ETAR

A

B

C

D

E

F

G

UK Government GHG Conversion Factors for Company Reporting

Water treatment

Index

Example of calculating emissions from water treatment

Company J report its emissions from mains water treatment, a Scope 3 emissions source. It gathers data from its utility bill and the company's own records. Company J multiplies the volume of water disposed of via the drains (in cubic metres (m³)) by the appropriate year's conversion factor to calculate its emissions from mains water treatment.

Activity	Type	Unit	kg CO ₂ e
Water treatment	Water treatment	cubic metres	0.272
		million litres	272.0

For information about how the conversion factors have been derived, please refer to the 'Methodology paper' that accompanies this document.

WTT- UK & overseas elec

WTT- heat and steam

Water supply

Water treatment

Material use

GREENHOUSE
GAS PROTOCOL

 Department
for Environment
Food & Rural Affairs



Case study

Uso de água: **por defeito 15 l água/lavagem**

$$\text{ETA } 15\text{L} \cdot 10^{-6} \cdot 149 \cdot 10^3 = 2.235 \text{ g CO}_{2\text{eq}}$$

$$\text{ETAR } 15\text{L} \cdot 10^{-6} \cdot 272 \cdot 10^3 = 4.08 \text{ g CO}_{2\text{eq}}$$



Case study

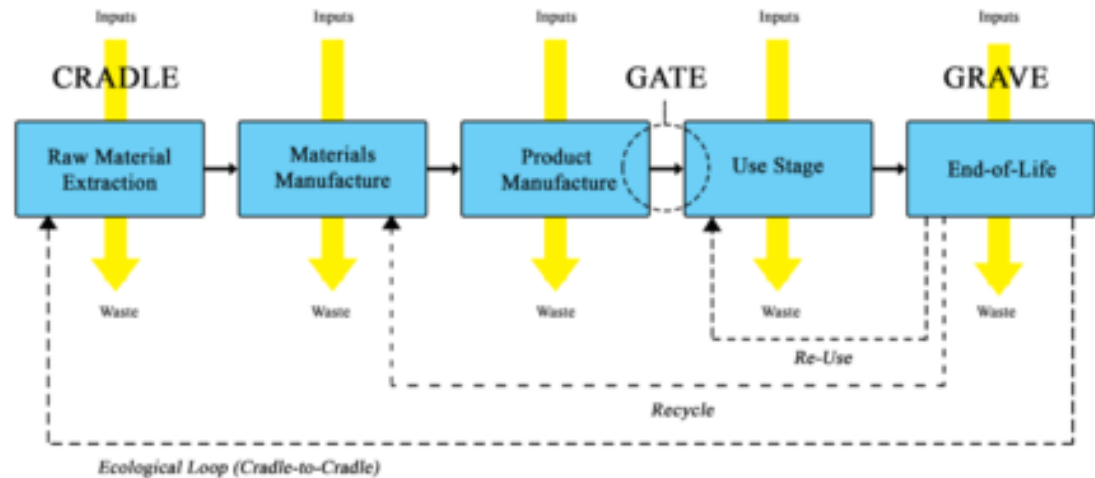
Uso de água: **por defeito 15 l água/lavagem**



Efic esquentador 90%
Determinação aquecimento água de 20°C para 45 °C: $mcp\Delta T$, cp 4.18 KJ/(kg.K)
Fator combustão gás natural: 0.203
gCO₂eq/kWh (ver DEFRA, separador FUELS)

Queima de gás natural

Uso de gás natural:



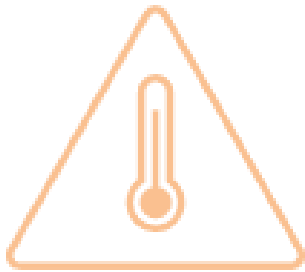
UK Government GHG Conversion Factors for Company Reporting

WTT- fuels

[Index](#)

Gaseous fuels	LPG	tonnes	347.0093
		litres	0.18383
		kWh (Net CV)	0.02719
		kWh (Gross CV)	0.02532
	Natural gas	tonnes	434.42892
		cubic metres	0.34593
		kWh (Net CV)	0.03474
		kWh (Gross CV)	0.03135
		tonnes	434.42892

Uso de gás natural para aquecimento de água:

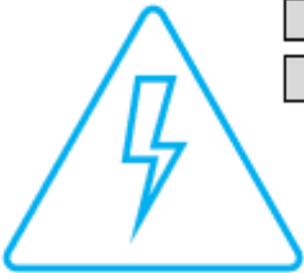


Combustão no local (casa)	$0.203 \text{ kgCO}_{2\text{eq}}/\text{kWh} * 0.48\text{kWh} = 0.097 \text{ kgCO}_{2\text{eq}}$
Extração e transporte para a rede de gás natural	$0.035 * 0.48 = 0.0168 \text{ kgCO}_{2\text{eq}}$
Total gás natural	$0.114 \text{ kgCO}_{2\text{eq}}$

$$\frac{mcp\Delta T}{0.9} = \frac{15 * 4.18 * 25}{0.9} = 1742 \text{ kJ} = 0.48 \text{ kWh/lavagem}$$

Uso de eletricidade por lavagem: 10 minutos

TIPOS DE LÂMPADAS				
				
CONSUMO	COMUM	HALÓGENA	CFL	LED
	40 W	28 W	8 W	4 W
	60 W	42 W	12 W	6 W
	75 W	53 W	15 W	8 W
	100 W	70 W	20 W	10 W
DURABILIDADE	1 ano	1-3 anos	6-10 anos	15-25 anos
ECONOMIA	x	até 30%	até 80%	até 95%



Uso de eletricidade por lavagem: 10 minutos

Combustão no local produção https://www.eea.europa.eu/data-and-maps/indicators/overview-of-the-electricity-production-4/assessment	198.4 gCO _{2eq} /kWh
Extração e transporte gás natural e carvão UKDEFRA	52 gCO _{2eq} /kWh
Perdas transmissão 6%	$1 \text{ kWh} = x \cdot (1 - 0.06)$ $X = 1.06 \text{ kWh}$
Total eletricidade	$(198.4 + 52) \cdot 1.06 = 265.4 \text{ gCO}_{2eq}/\text{kWh}$

Case study

Quimicos



Function	Ingredient	CAS	DID-list N°	Concentration (wt%)
Anionic surfactant	Sodium laureth sulfate	68891-38-3	8	13.00
Amphoteric surfactant	Cocamidopropyl betaine	61789-40-0	61	8.00
Non-ionic surfactants	Cocamide MEA	68140-00-1	50	1.25
Viscosity controlling agent	Propylene glycol	57-55-6	174	1.00
Preservative	Sodium benzoate	532-32-1	95	0.30
pH-adjustor	Chlorhydric acid	7647-01-0		0.80
Fragrance	alpha-hexyl cinnamaldehyde	101-86-0	142	0.50
	beta-pinene	127-91-3		
	Dihydromyrcenol	2436-90-0		
	Hexyl salicylate	115-95-7		
	Patchouli oil	84238-39-1		
Additional ingredients for additional functions (e.g. hair conditioning agent, hypo-irritancy agent)	Dimethicone	63148-62-9	110	1.00
Additional ingredients for additional functions (e.g. hair conditioning agent, hypo-irritancy agent)	Polyquaternium-10	68610-92-4		0.40
Additional ingredient for aspect (pearlescent / opacifying agent)	Glycol distearate	627-83-8	185	0.50
Solvent	Water			73.25

Case study

Quimicos



Function	Ingredients for base case	Ingredients for worst case	Percentage (%)	Amount (g) in 100 g of product
Saponified oils (92%)	Tallow	Tallow		57
	Coconut oil fatty acids	Coconut oil fatty acids	92%	14
	Stearic acid	Stearic acid		14
Emulsifying / humectant	Glycerine	Propylene glycol	6%	5.52
Perfuming	Perfume	Benzyl alcohol	1%	1.38
Colorant	Colorants	Colorants	0,1%	0.092
Chelating agent	EDTA	EDTA	0,2%	0.184
Bleaching agent	Titanium dioxide	Titanium dioxide	0,1%	0.092
Water	Water	Water	8%	8

Desafio #4:

Juntar materiais com químicos e uso em gráficos de barra com cada fase identificada

Responder e justificar

Qual a contribuição dos químicos, embalagem e uso de água para o impacte/impacto total de CO₂eq/lavagem? Qual a mais importante no uso (eletricidade, gás natural, ETA ou ETAR?)

Qual o cenário de reciclagem e fim de vida preferível?

Qual o shampoo que permite ter menor pegada carbónica?

Limitações do estudo?

Desafio #4:

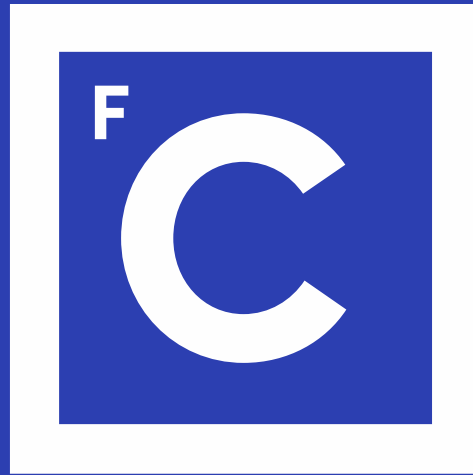
Apresentação dia 16 de Novembro sobre o caso de estudo e conclusões

14h30 Joana

14h50 Ivan

15h10 Henrique

Thanks



Ciências ULisboa

Faculdade
de Ciências
da Universidade
de Lisboa