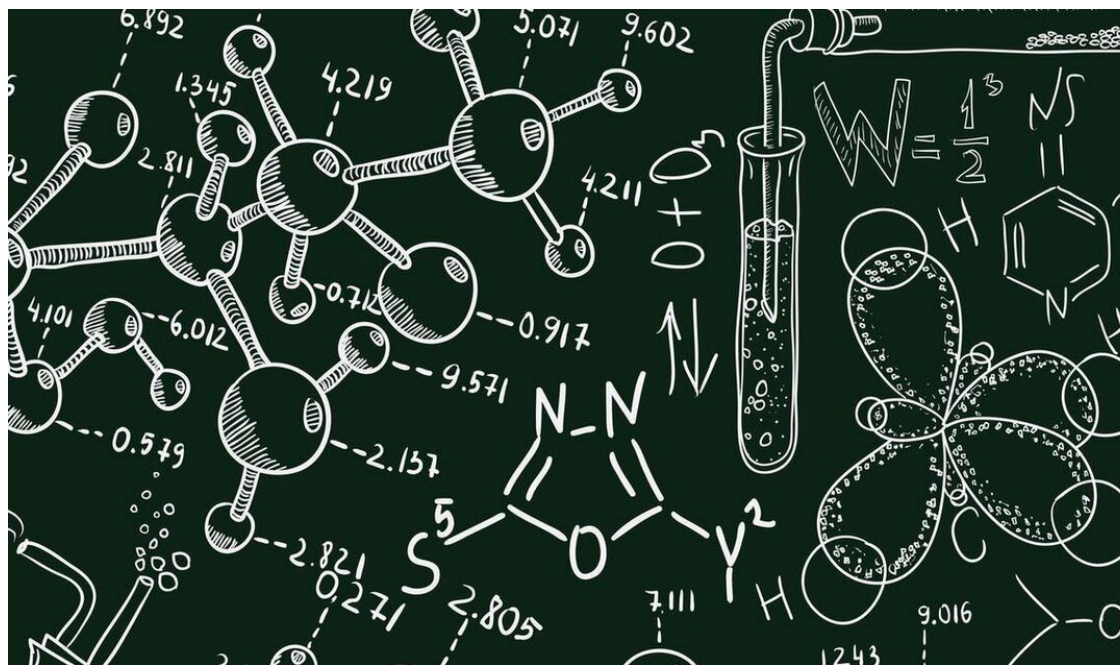
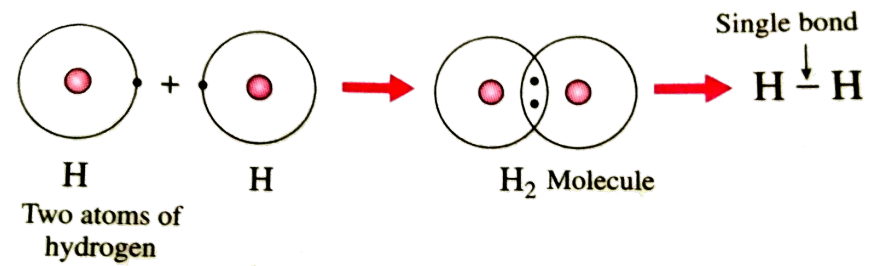
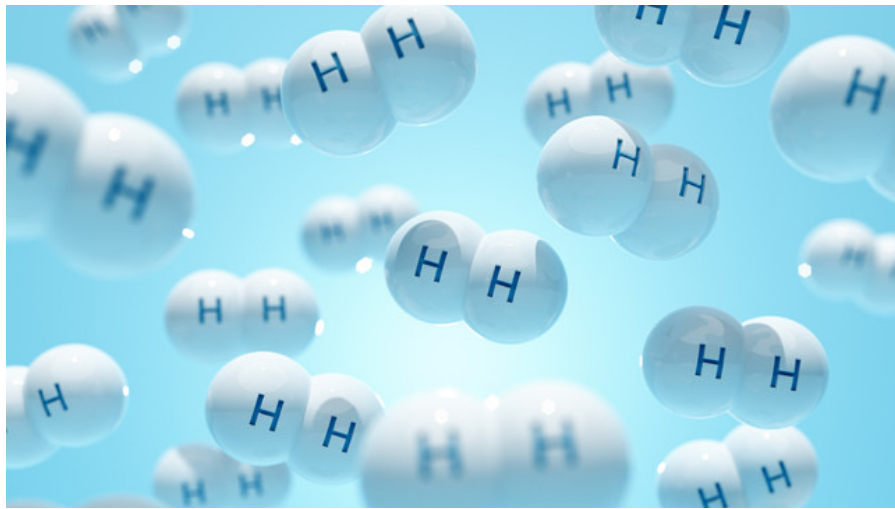


5. Bonding



Hydrogen molecule



Schroedinger's equation

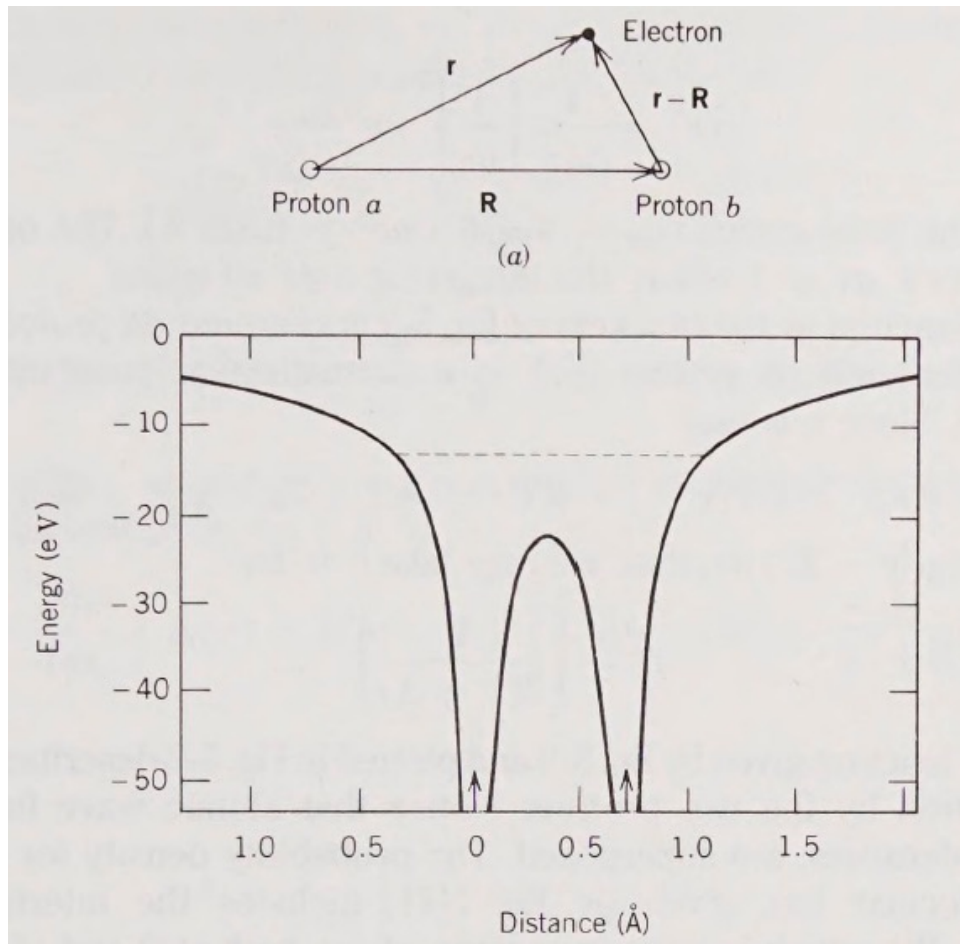
$$H\Psi = i\hbar \frac{\partial \Psi}{\partial t}$$

Time evolution

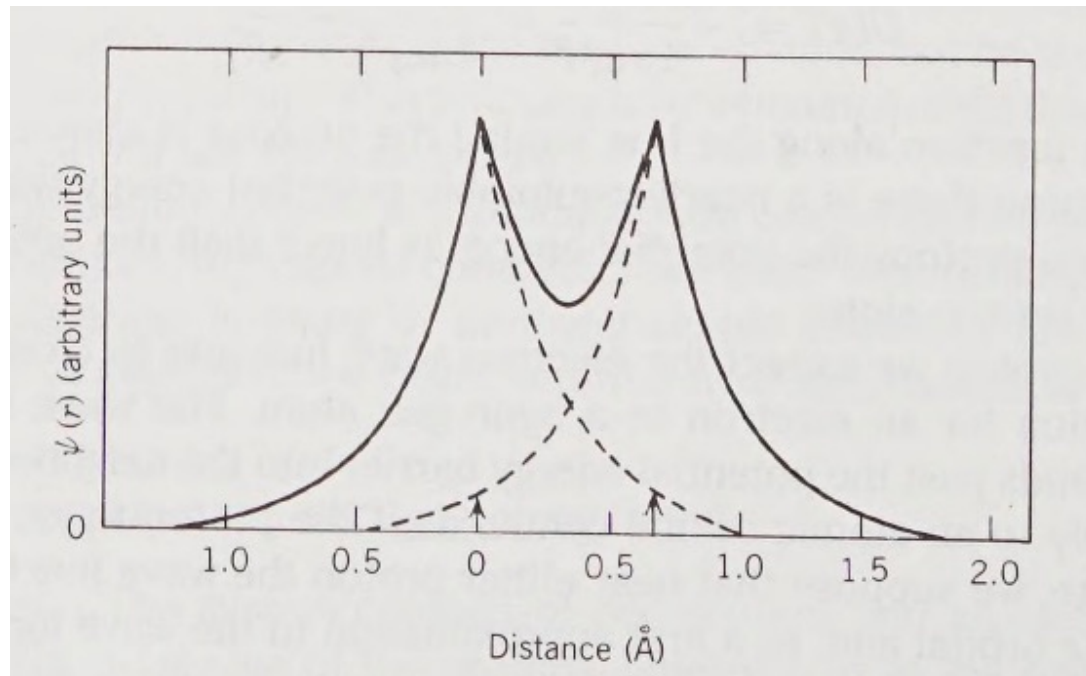
$$\frac{-\hbar^2}{2m} \frac{\partial^2 \Psi(x)}{\partial x^2} + U(x)\Psi(x) = E\Psi(x)$$

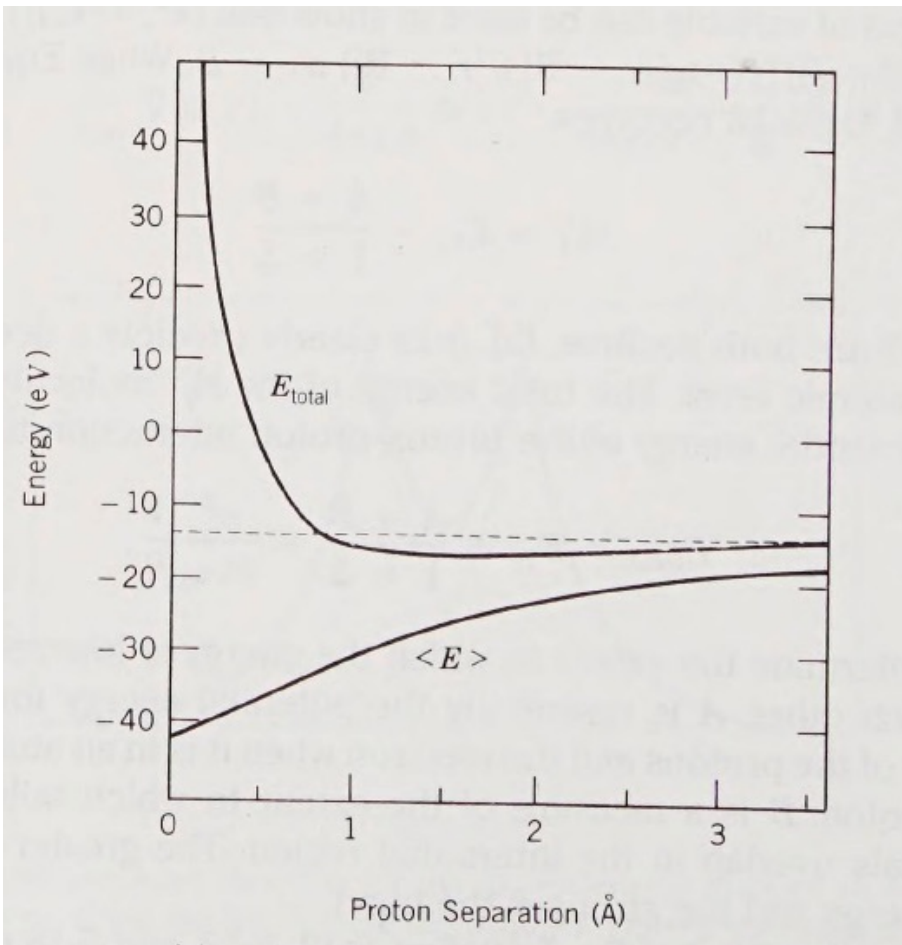
Time independent equation

Molecular H₂ ion



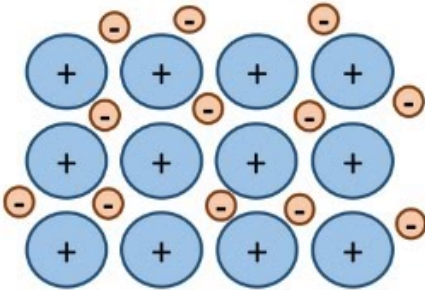
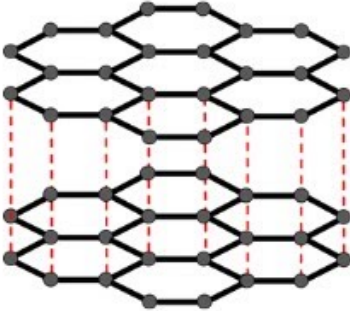
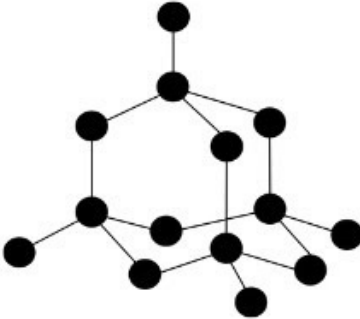
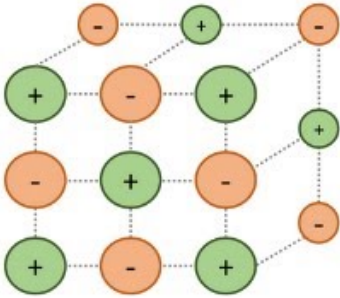
Bonding wave function



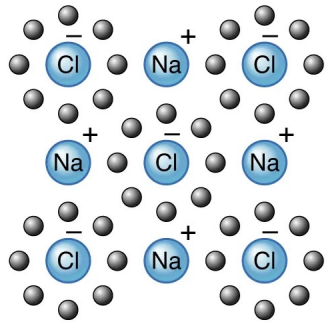


Energy

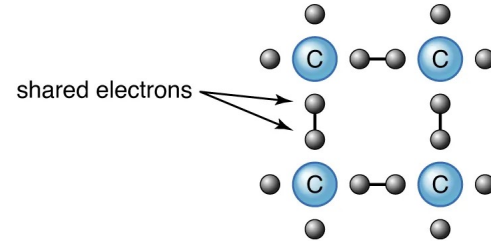
Ionic, covalent, van der Waals and metallic bonding



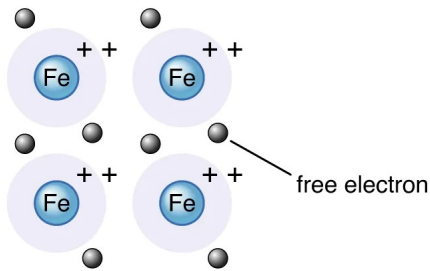
Bonding mechanisms



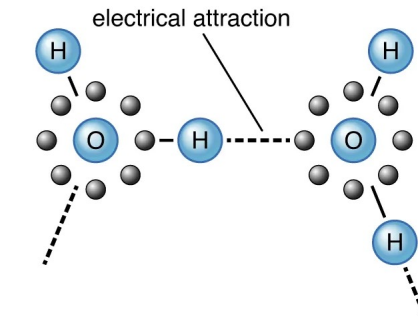
ionic bonding
electron transferred from Na to Cl



covalent bonding
atoms share electrons

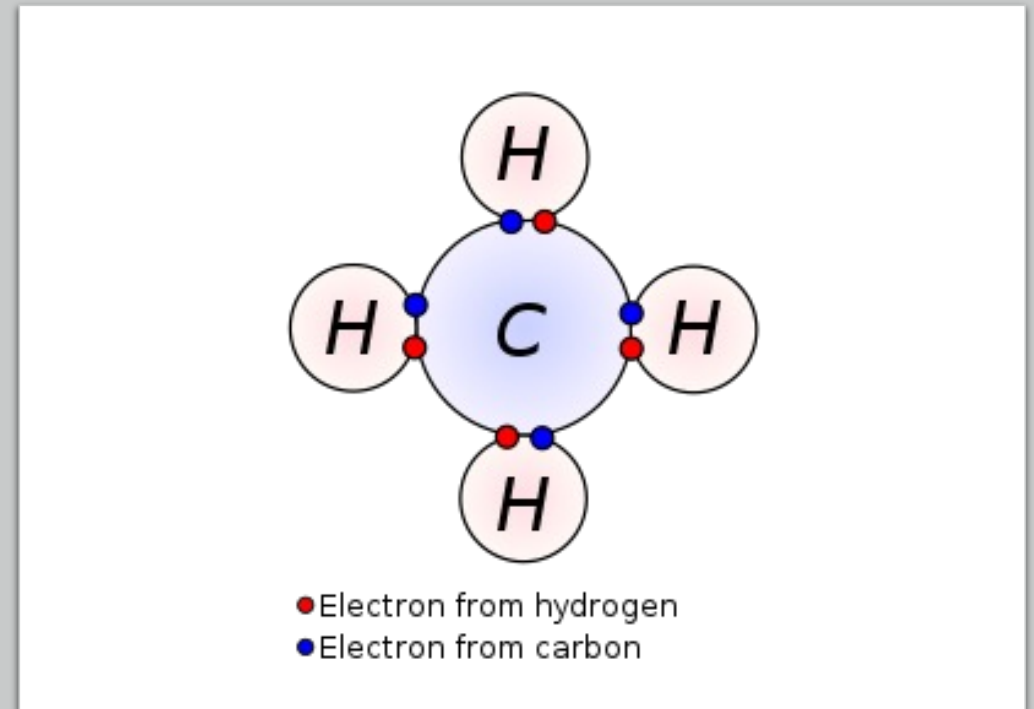
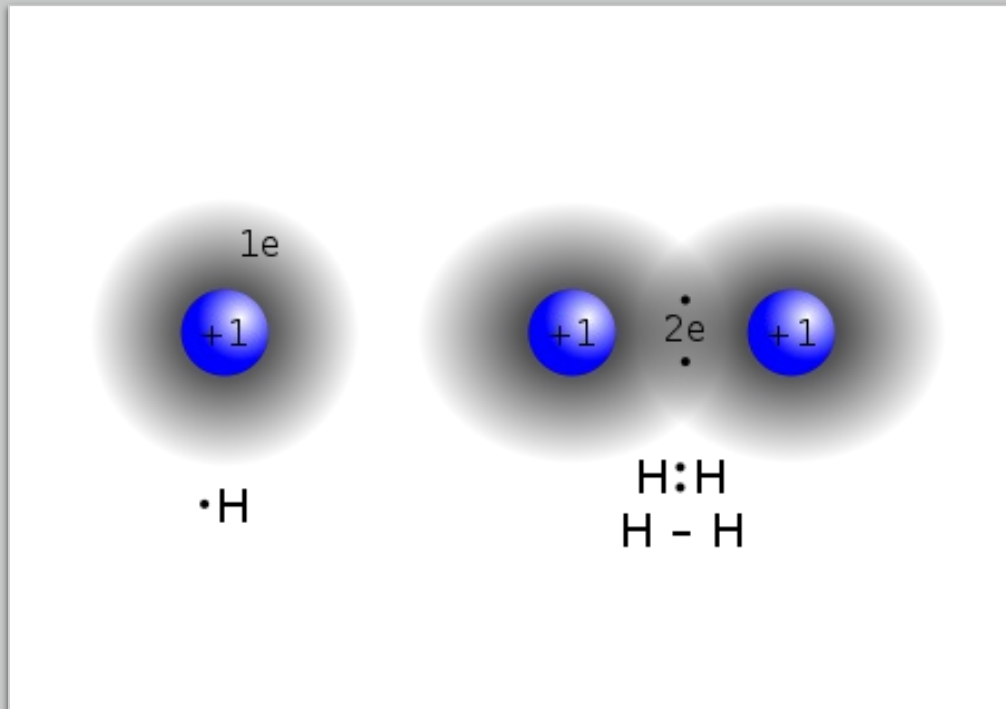


metallic bonding
ions surrounded by free electrons

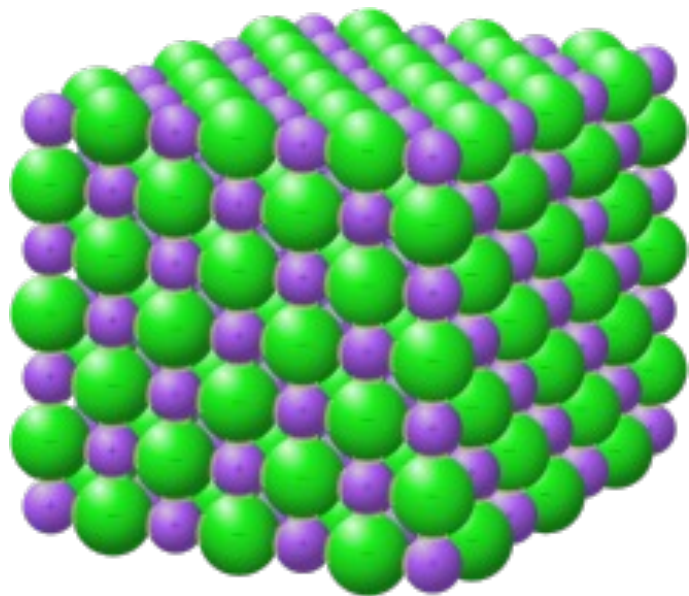


molecular bonding
weak electrical attraction binds molecules

Covalent bonding



Ionic bonding



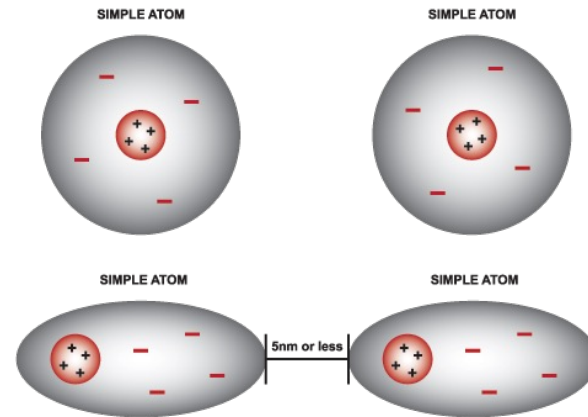
Van der Waals bonding



VAN DER WAALS' FORCES (VDW)
DIAGRAM

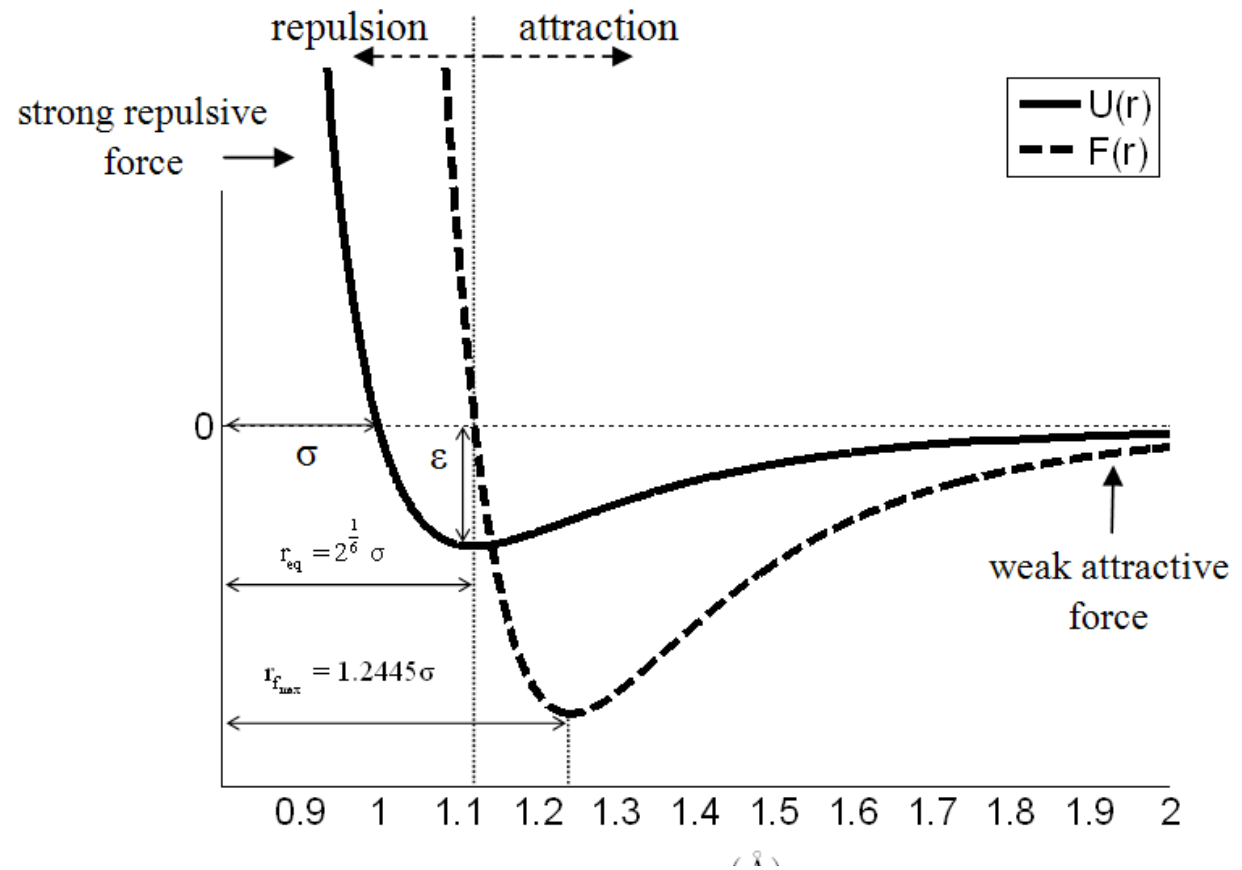
KEY

- + POSITIVE NUCLEUS
- NEGATIVE CHARGED ELECTRON CLOUD

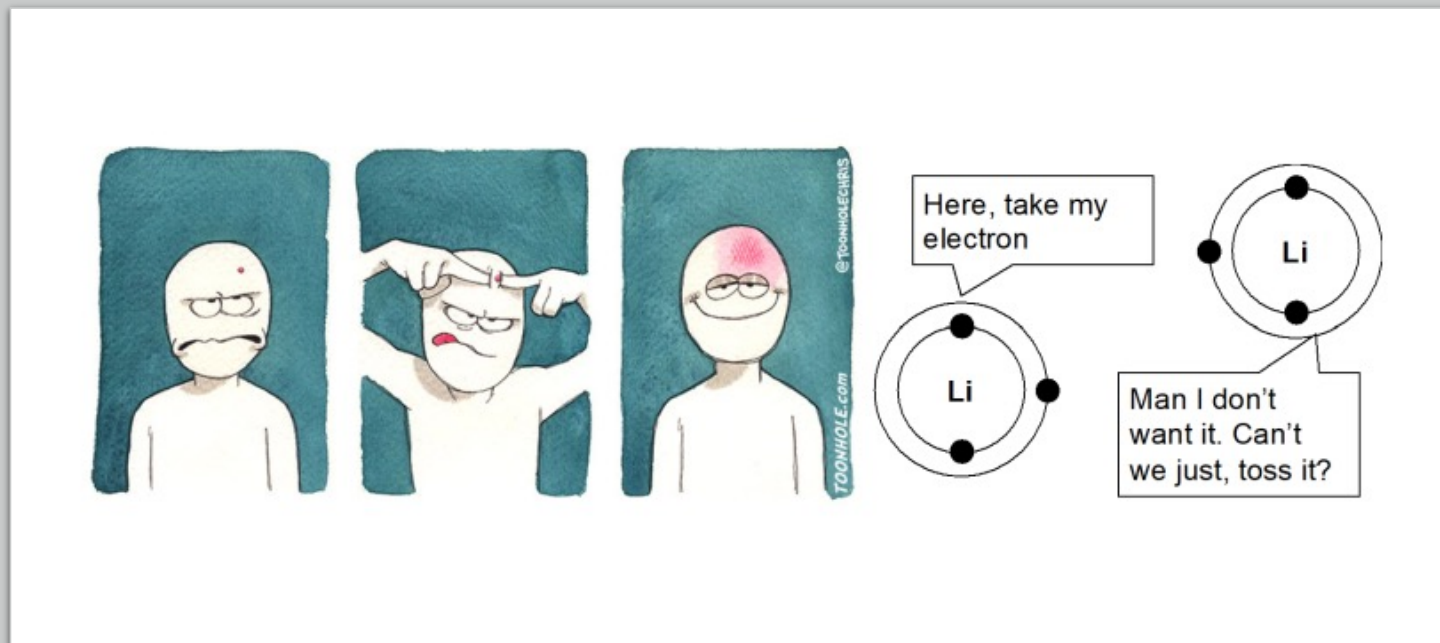


When two atoms come within 5 nanometers of each other, there will be a slight interaction between them, thus causing polarity and a slight attraction.

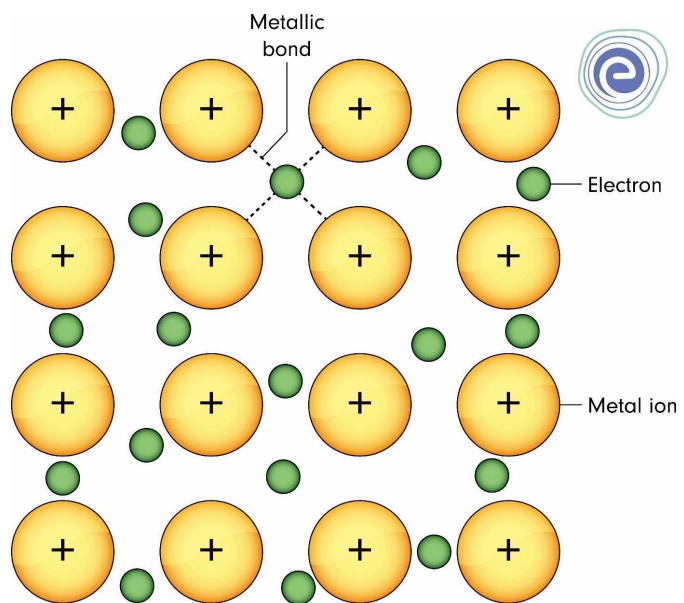
Lennard-Jones potential



Metallic bonding



Metallic bonding



Aluminum



Copper



Brass



Magnesium

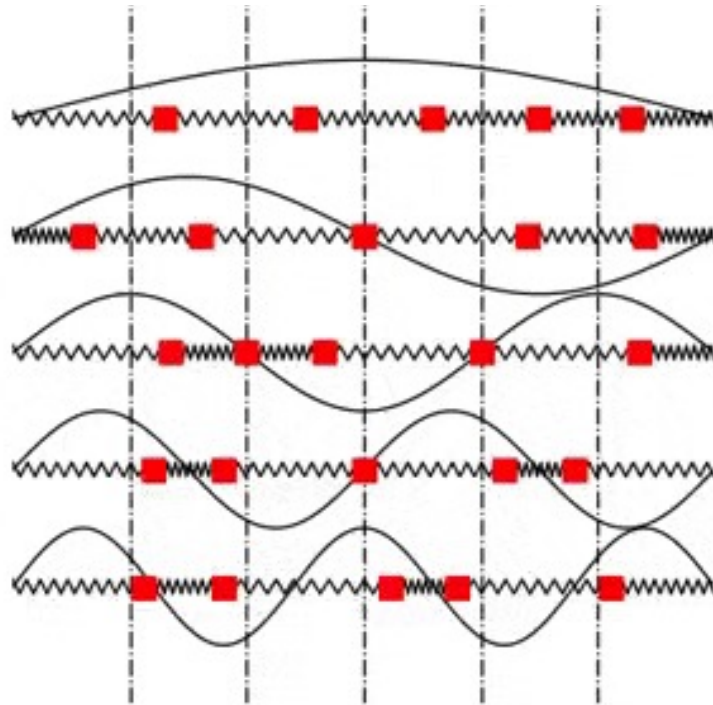


Titanium

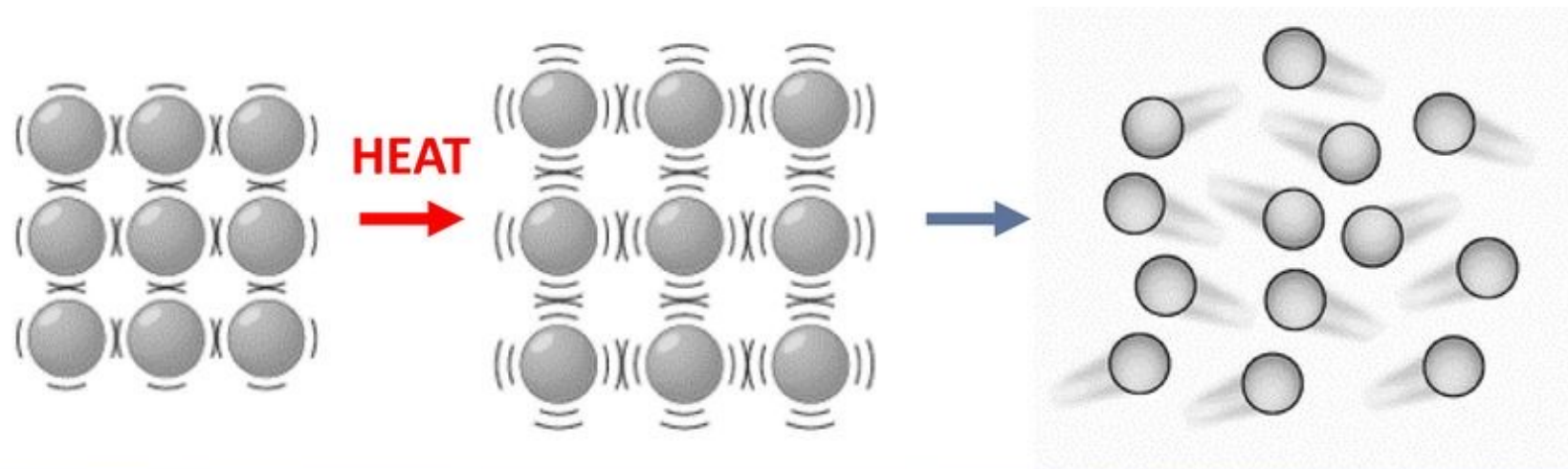


Tungsten

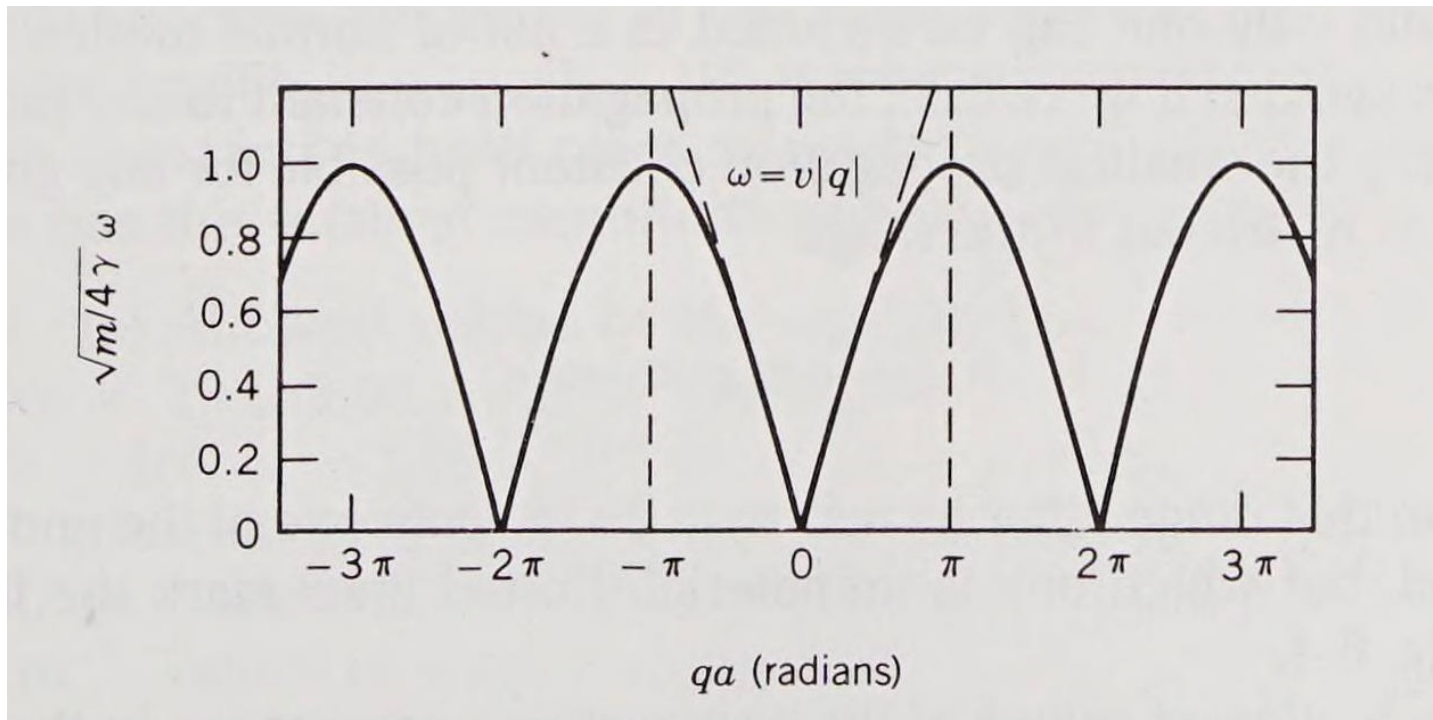
6. Vibrações atômicas



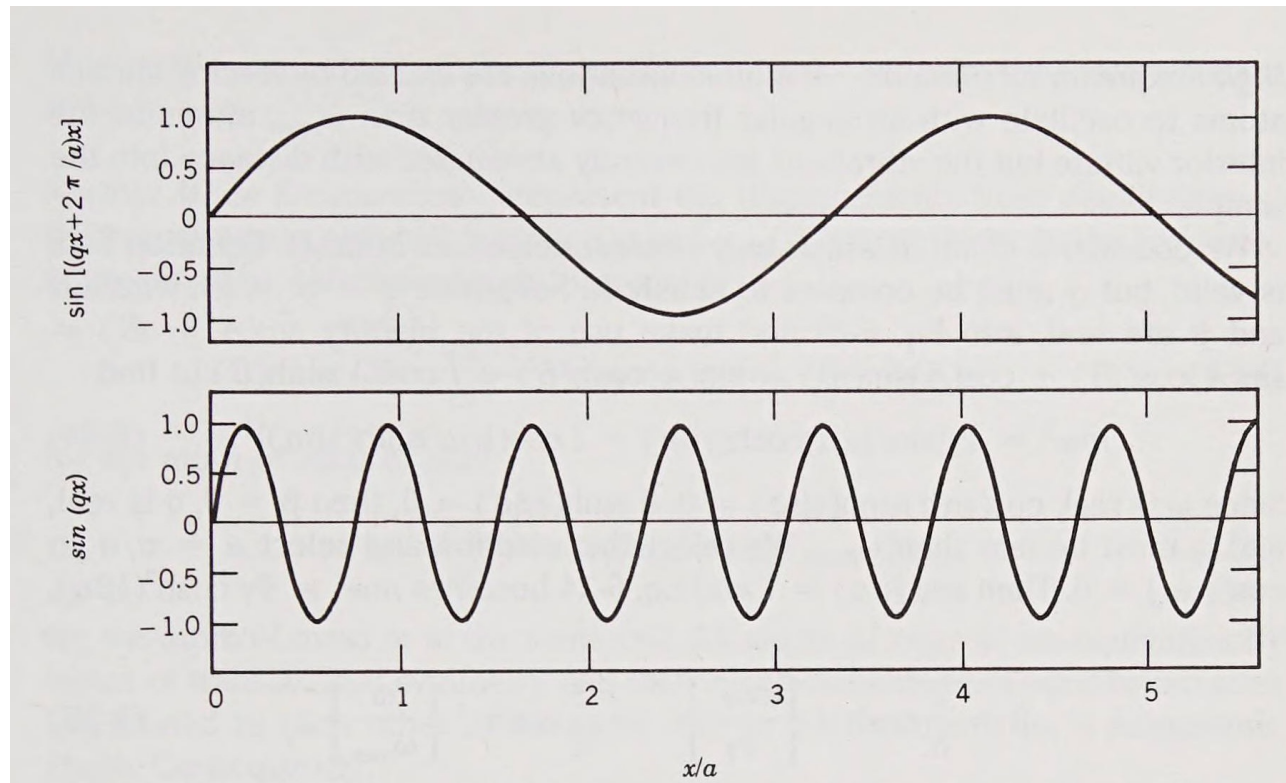
Vibrations



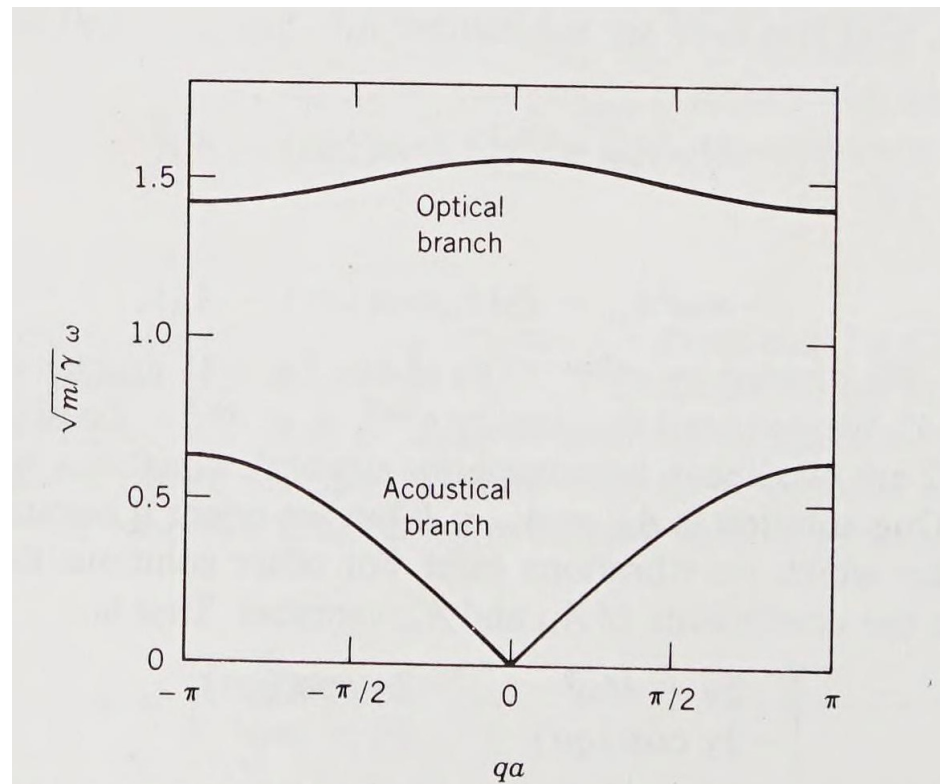
Dispersion relation: 1d lattice



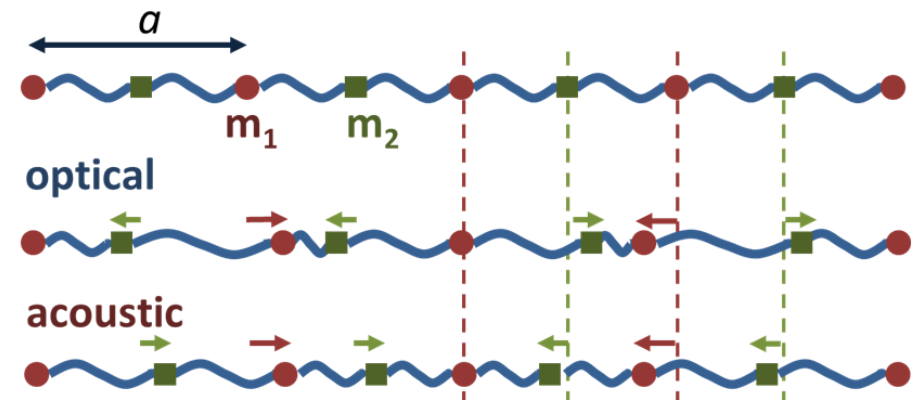
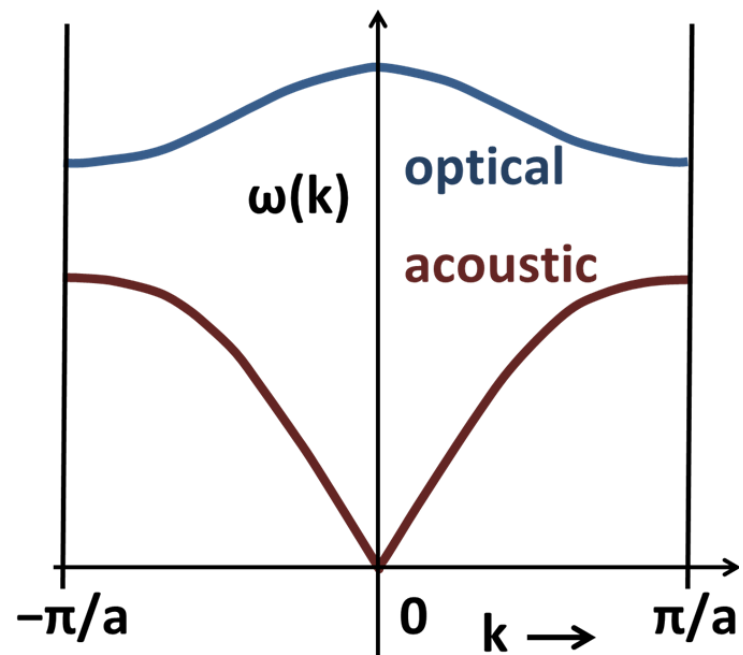
N normal modes: 1d lattice



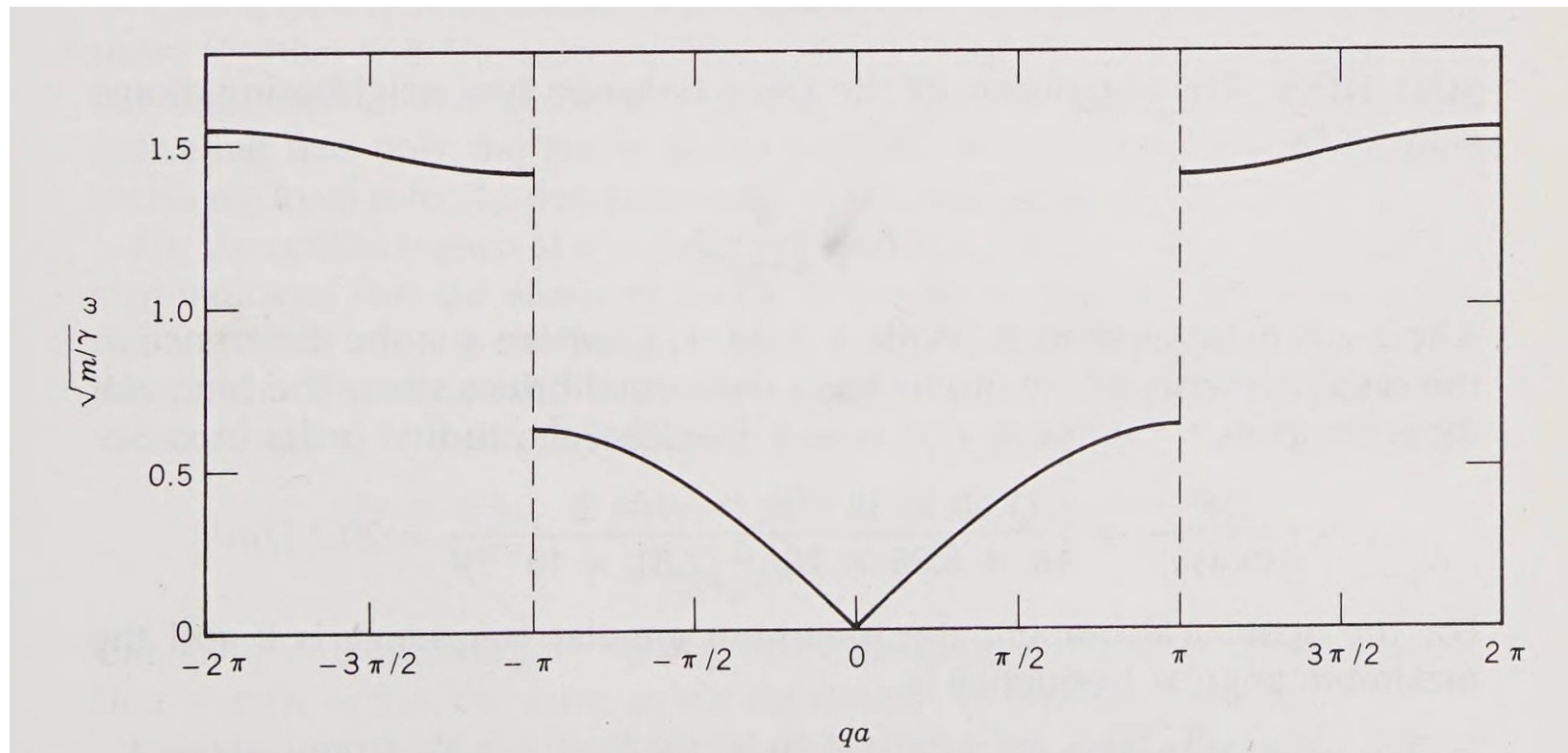
1d lattice with basis of two atoms



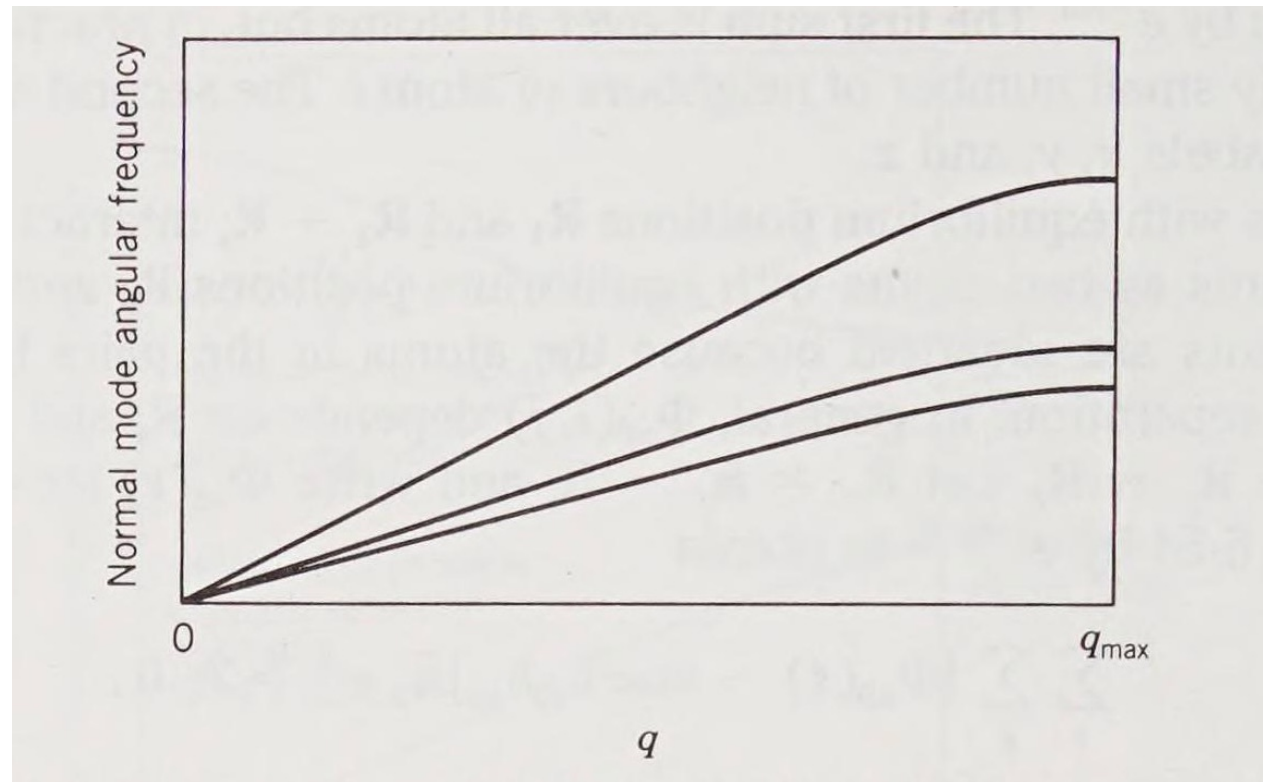
1d lattice with basis of two atoms



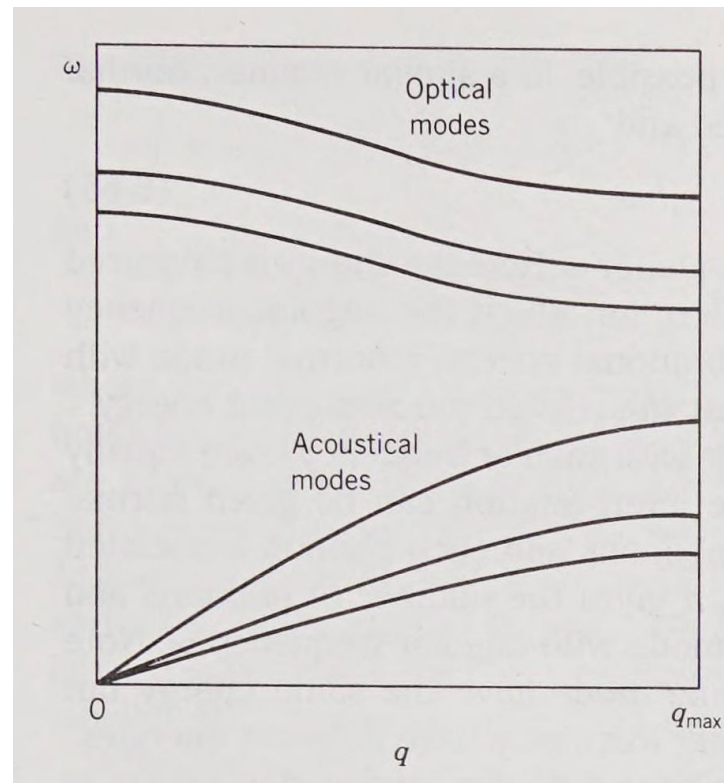
1d lattice with basis of two atoms: extended zone



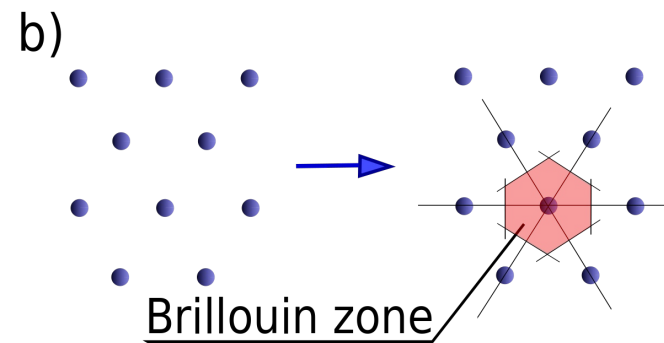
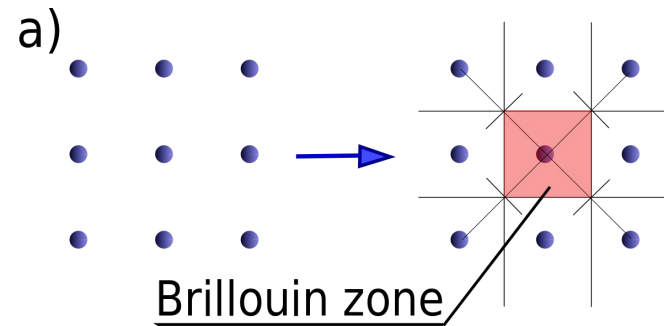
3d lattice



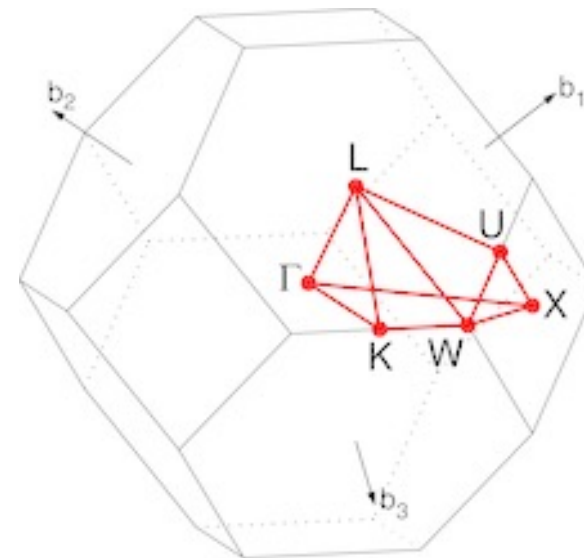
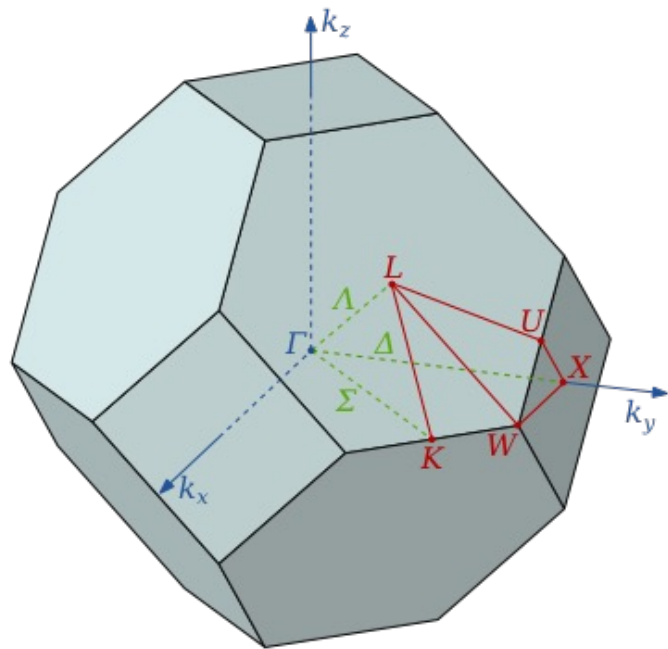
3d lattice with basis of two atoms



Brillouin zone



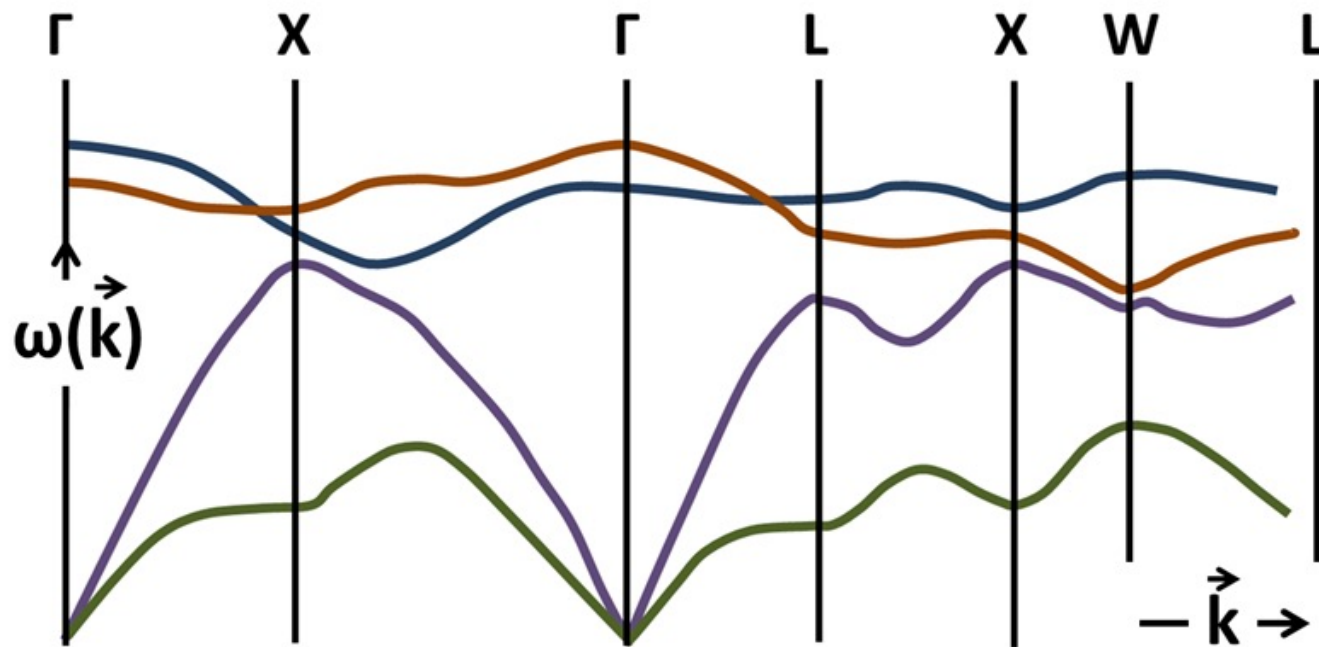
Brioullin zone



FCC path: Γ -X-W-K- Γ -L-U-W-L-K|U-X

[Setyawan & Curtarolo, DOI: 10.1016/j.commatsci.2010.05.010]

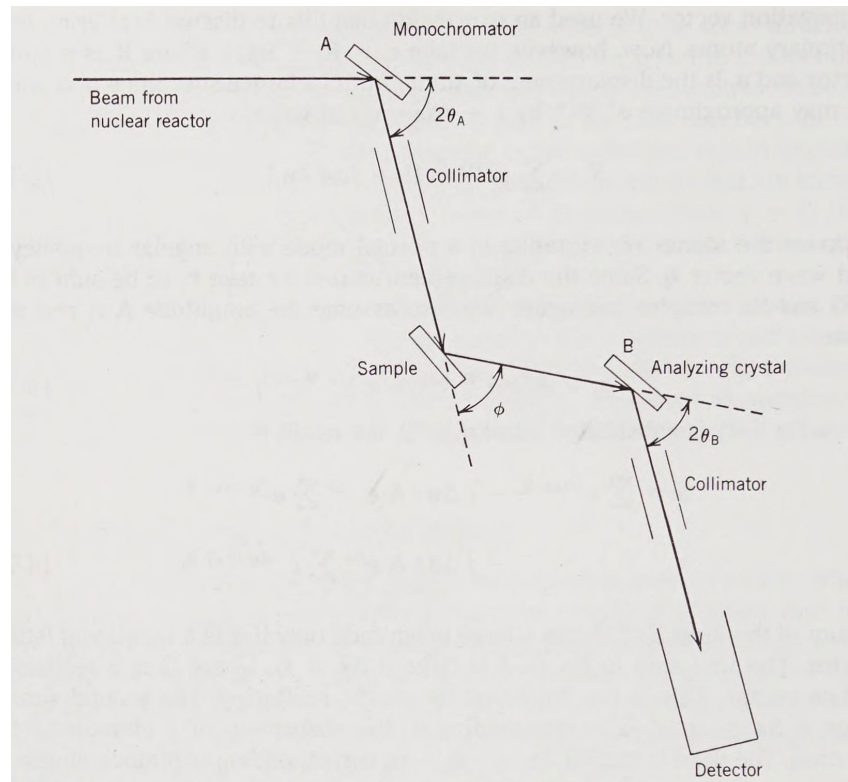
3d lattice with basis of two atoms



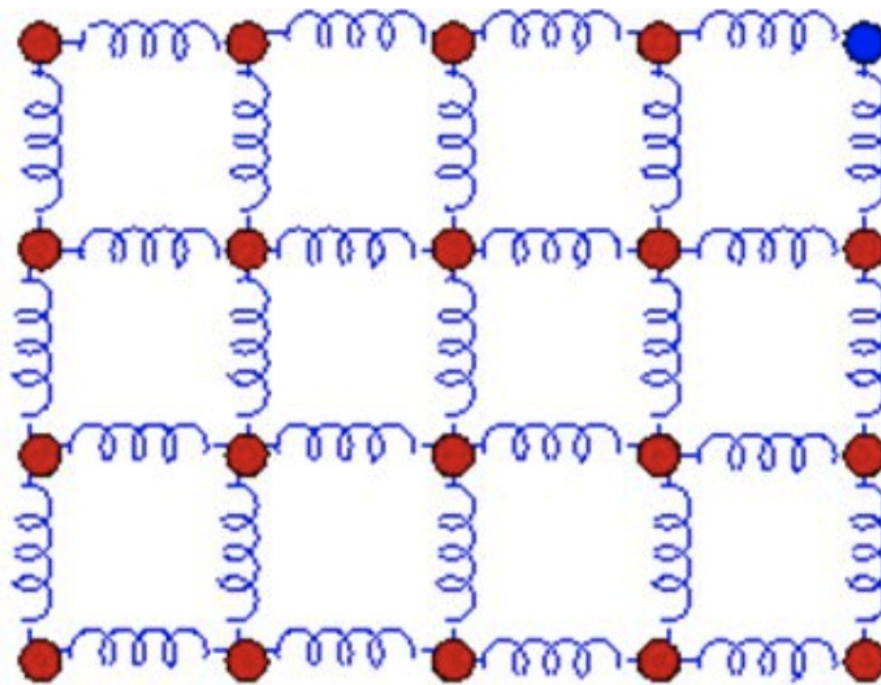
Crystal momentum



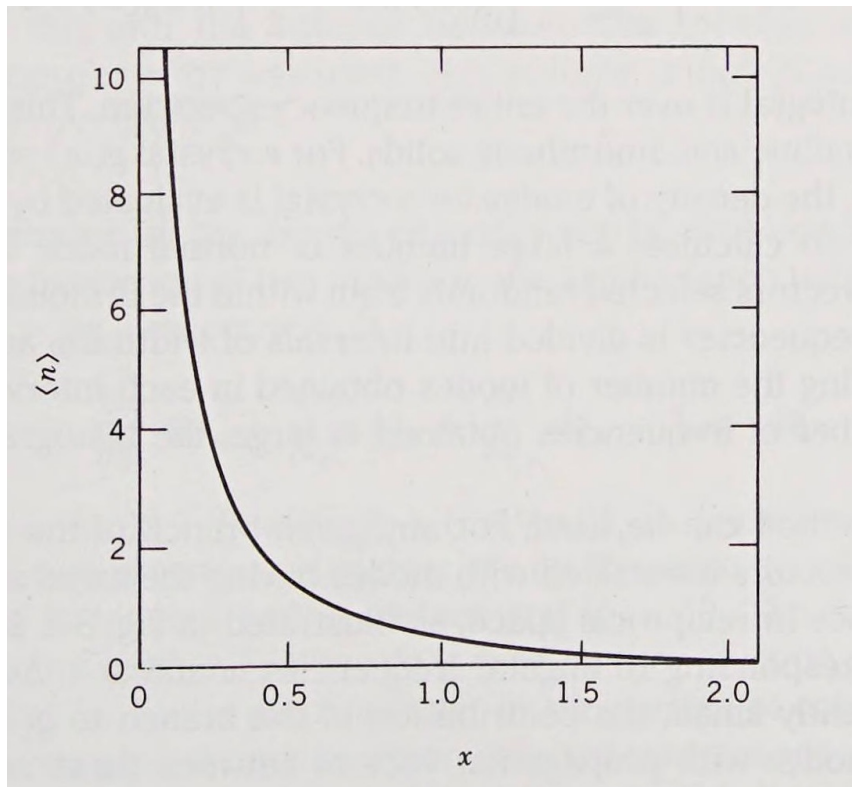
Triple axis neutron spectrometer



7. Termodinâmica de fonões



Number of phonons as a function of $x = \beta \hbar \omega$



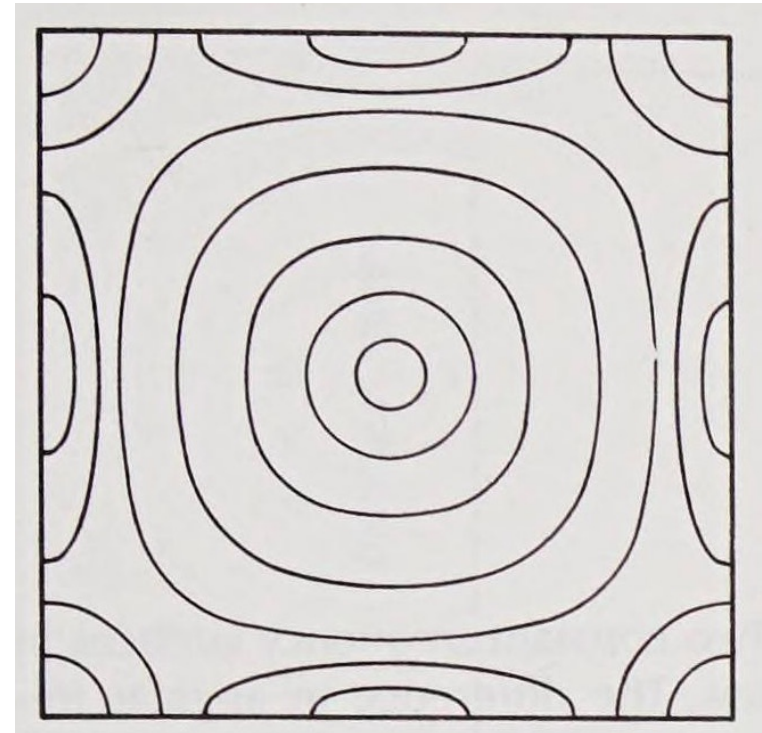
$$\langle n \rangle \longrightarrow e^{-\beta \hbar \omega}$$

$$\langle E \rangle \longrightarrow \hbar \omega e^{-\beta \hbar \omega} + \frac{1}{2} \hbar \omega,$$

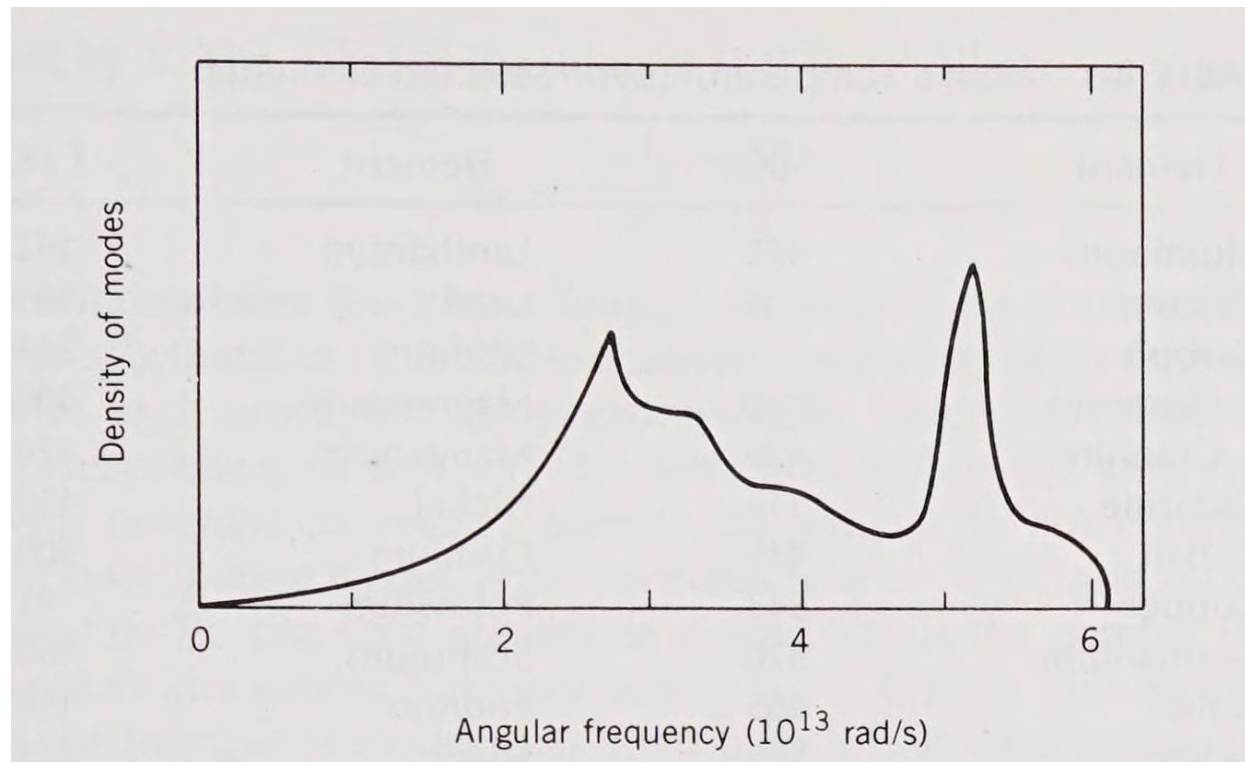
$$\langle n \rangle \longrightarrow \frac{1}{\beta \hbar \omega} = \frac{k_B T}{\hbar \omega}$$

Constant frequency surface

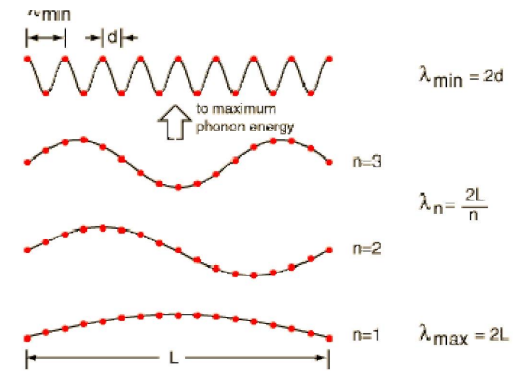
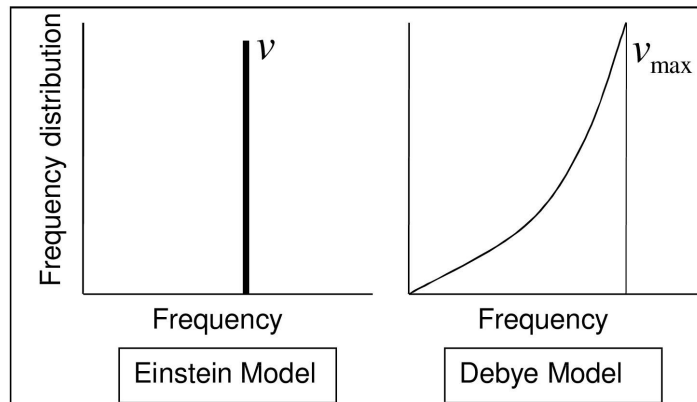
FIGURE 8-2 A schematic representation of constant frequency surfaces in reciprocal space. A Brillouin zone is shown and each contour is defined by propagation vectors such that $\omega(\mathbf{q}) = \text{constant}$. Different surfaces are associated with different values of ω .



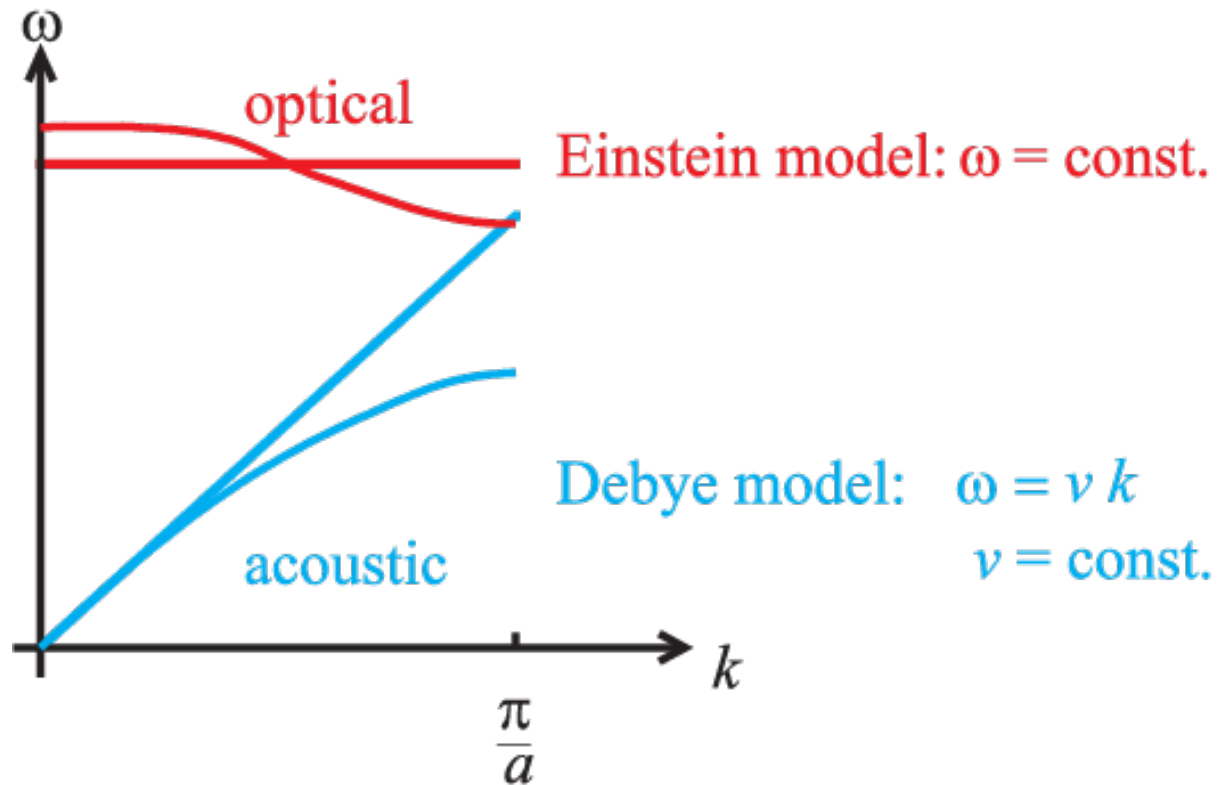
Density of modes



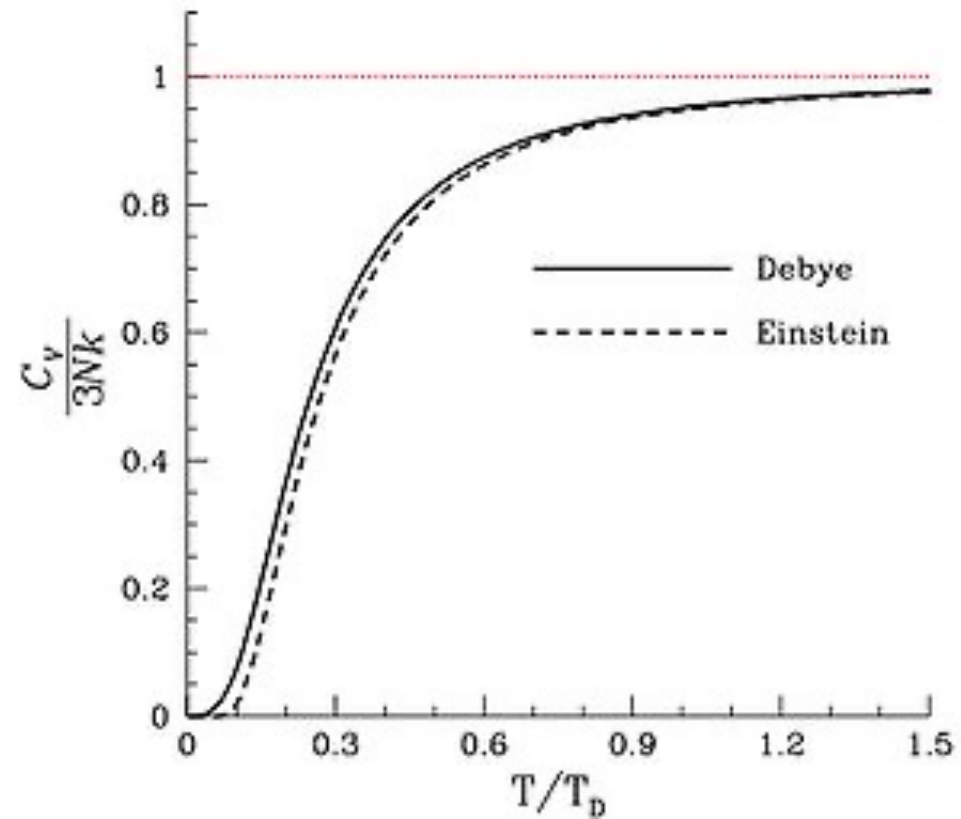
Models of Einstein and Debye



Models of Einstein and Debye



Models of Einstein and Debye



Model of Debye for various metals

Debye Model of Heat Capacity of Solids

