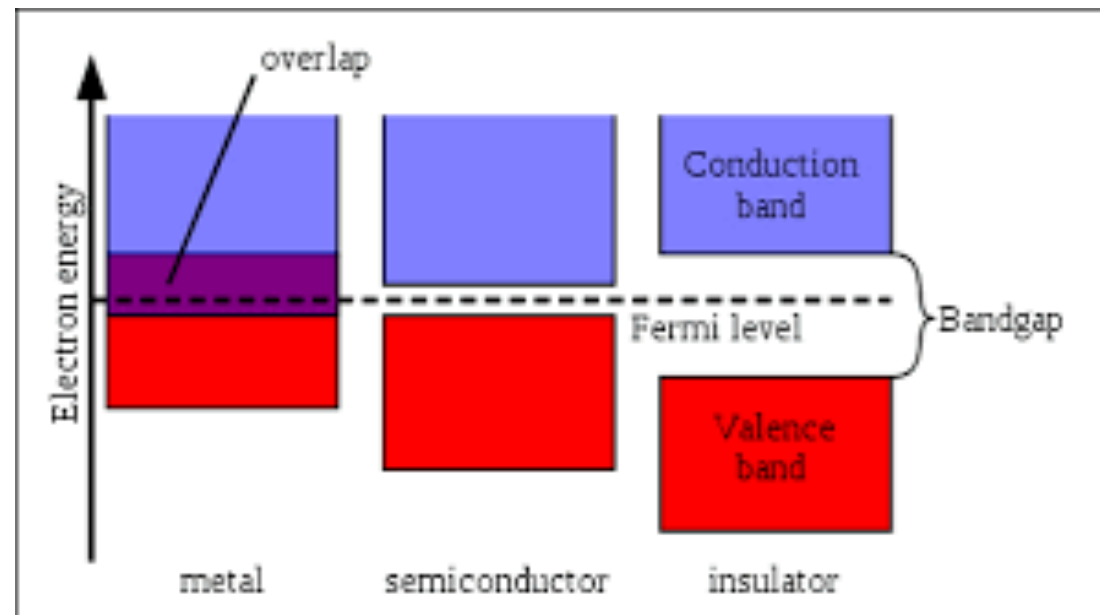
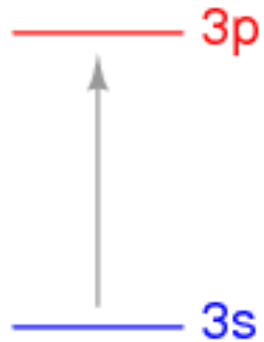


8. Estados electrónicos



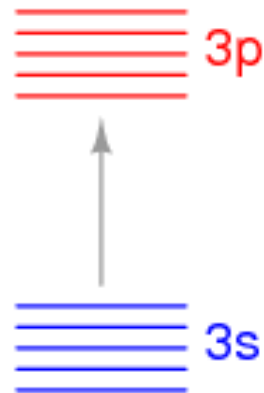
Overlap of atomic states

Significant leap required for an electron to move to the next higher level



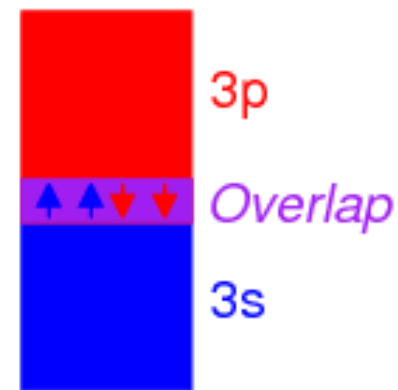
Single atom

Shorter leap required



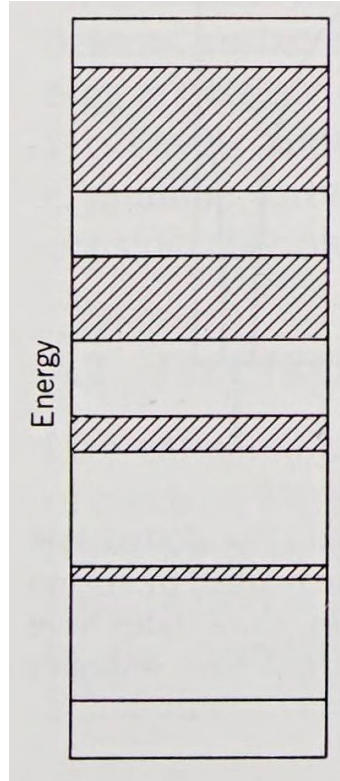
Five atoms
in close proximity

Overlap permits electrons to freely drift between bands

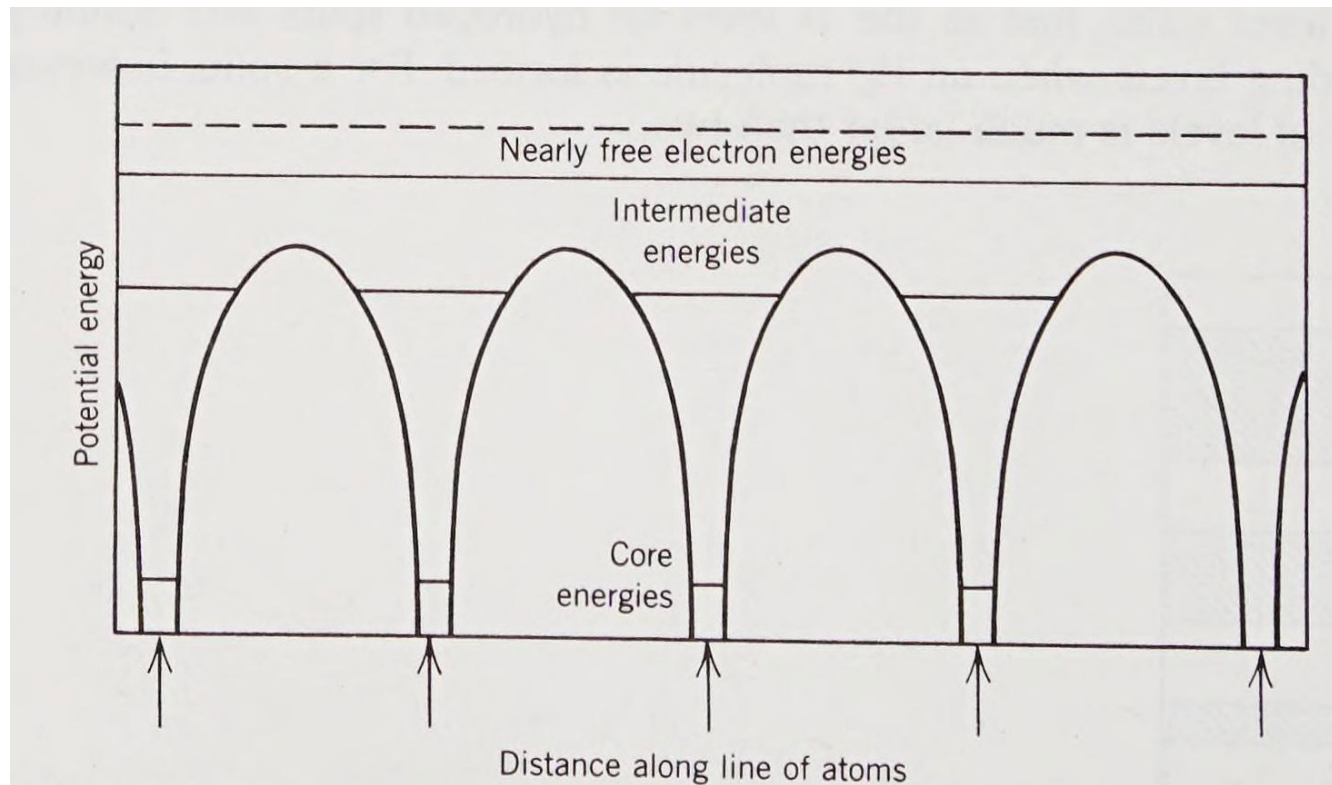


Multitudes of atoms
in close proximity

Bands



Electronic energy along a line of atoms

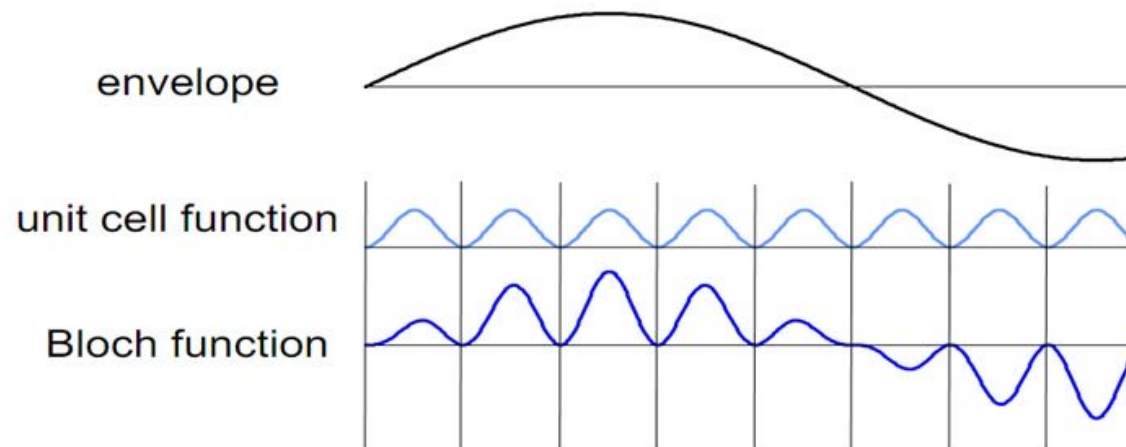


Bloch's theorem

we can rewrite the Bloch theorem equation $\psi(x + a) = \exp(ika)\psi(x)$ alternative form

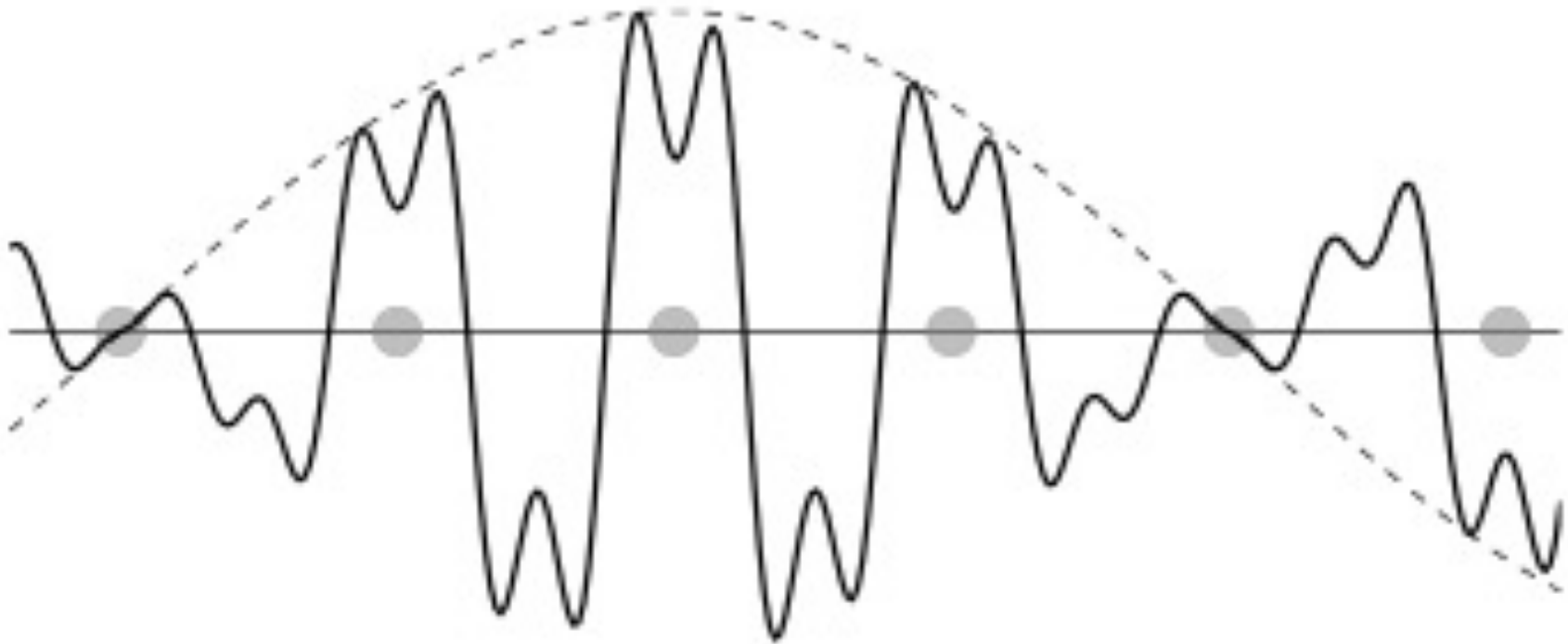
$$\psi(x) = u(x)\exp(ikx)$$

where $u(x)$ is periodic with the lattice periodicity.

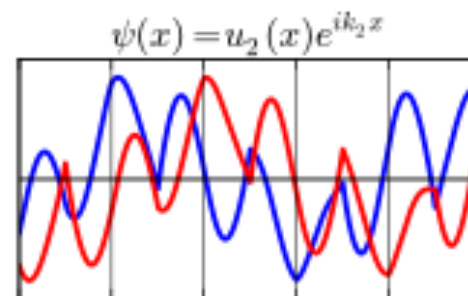
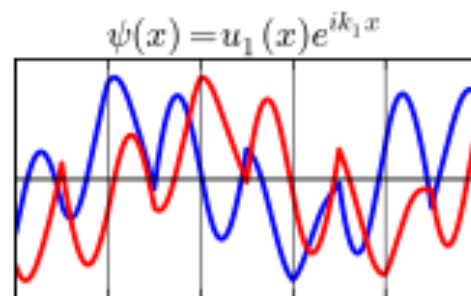
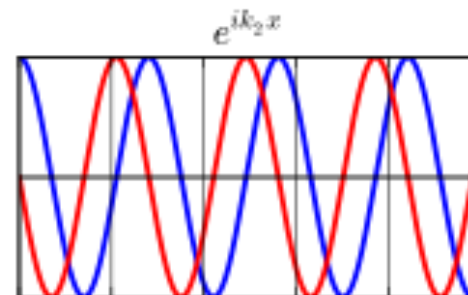
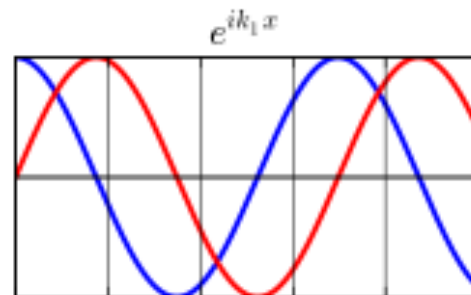
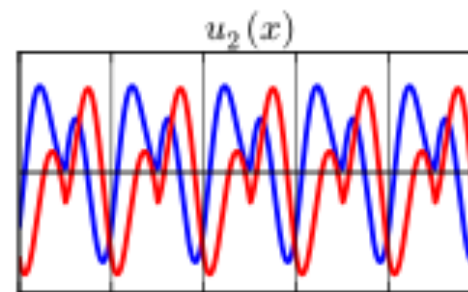
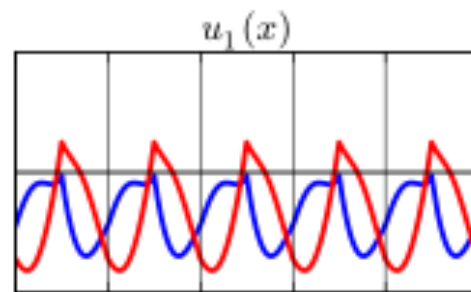


Concept of the Bloch functions. We can think of the $\exp(ikx)$ as being an example of an “envelope” function that multiplies the unit cell function $u(x)$

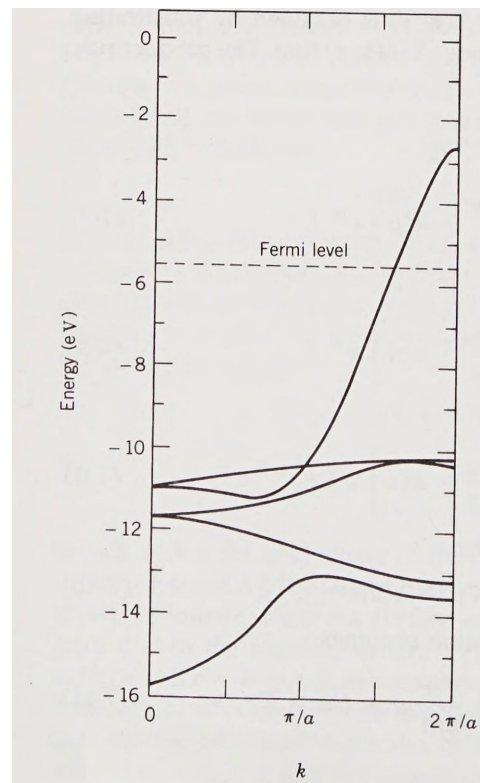
Bloch function



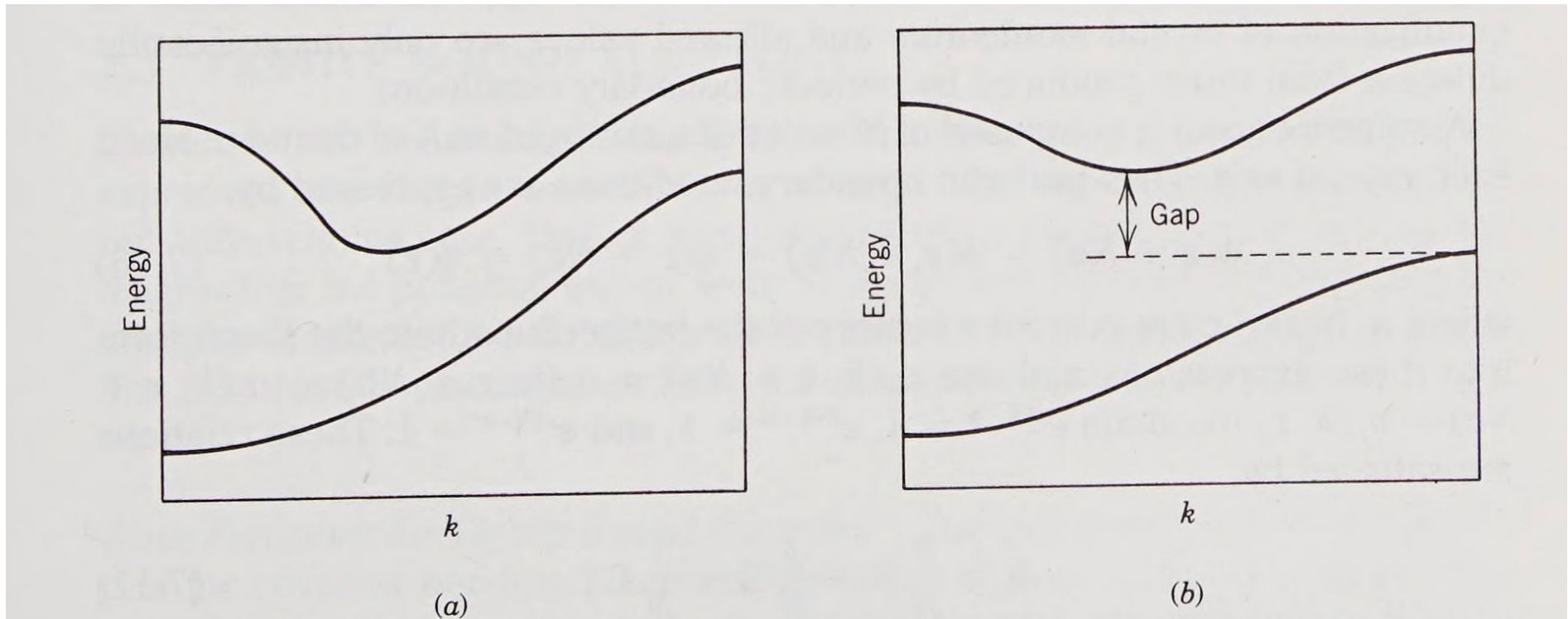
Bloch functions



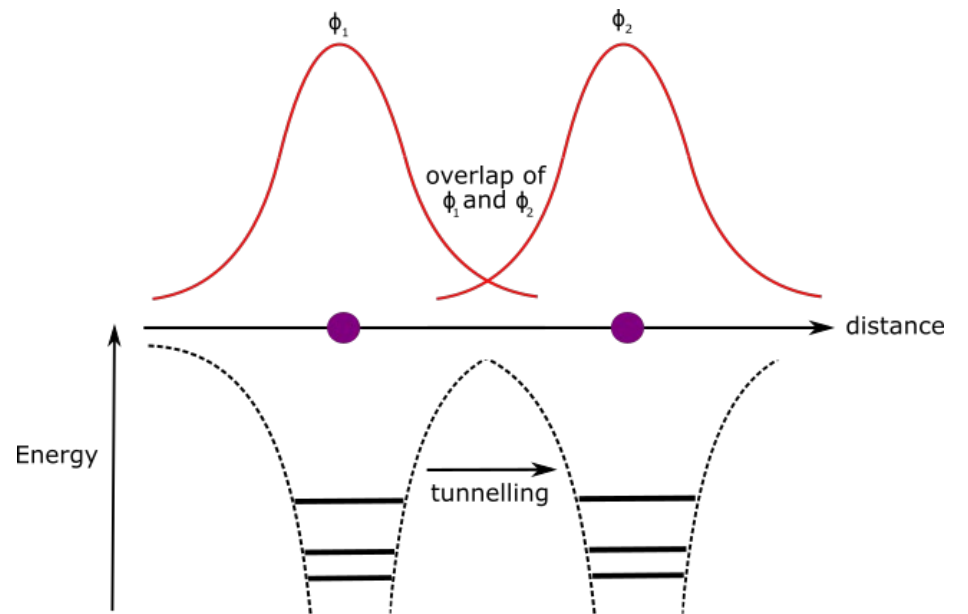
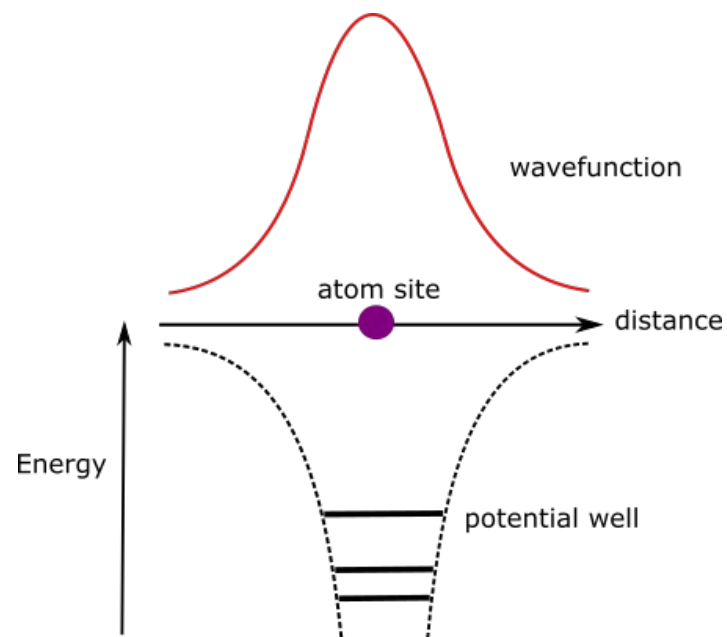
Bands of copper in the 100 direction



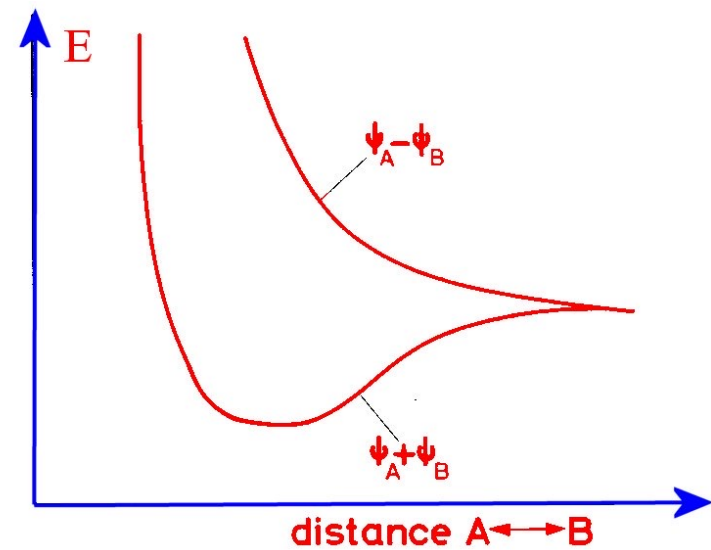
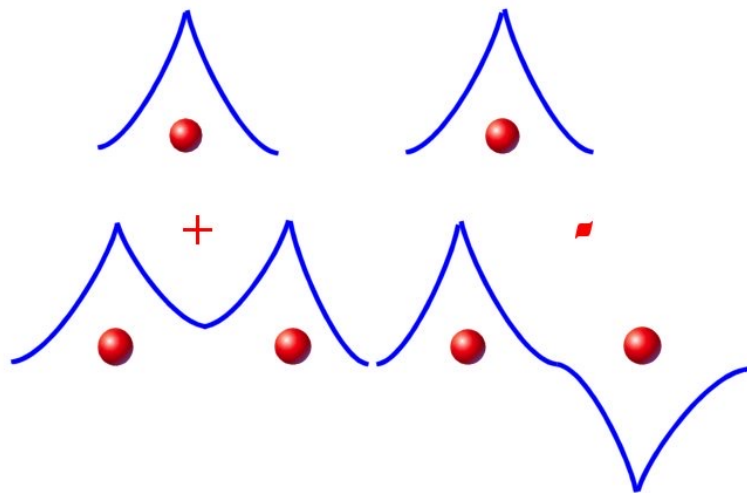
Band overlap



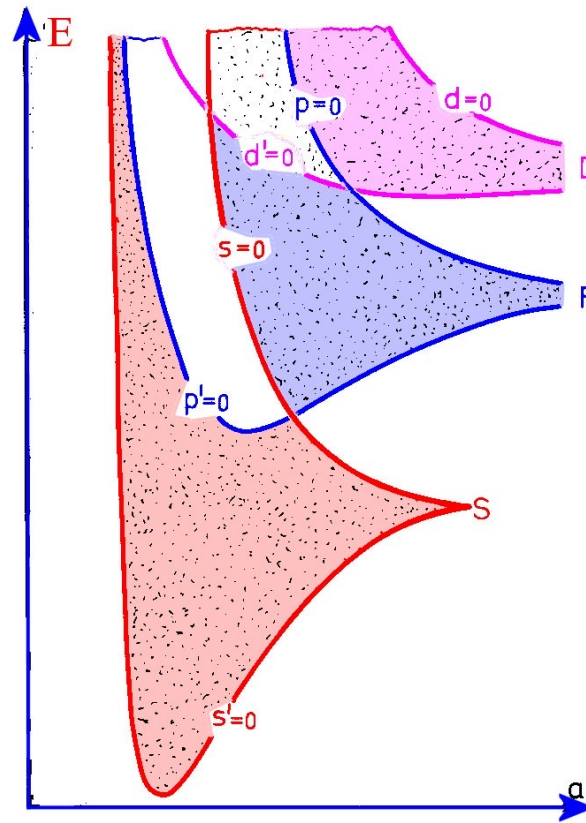
Recall: Covalent Bonding



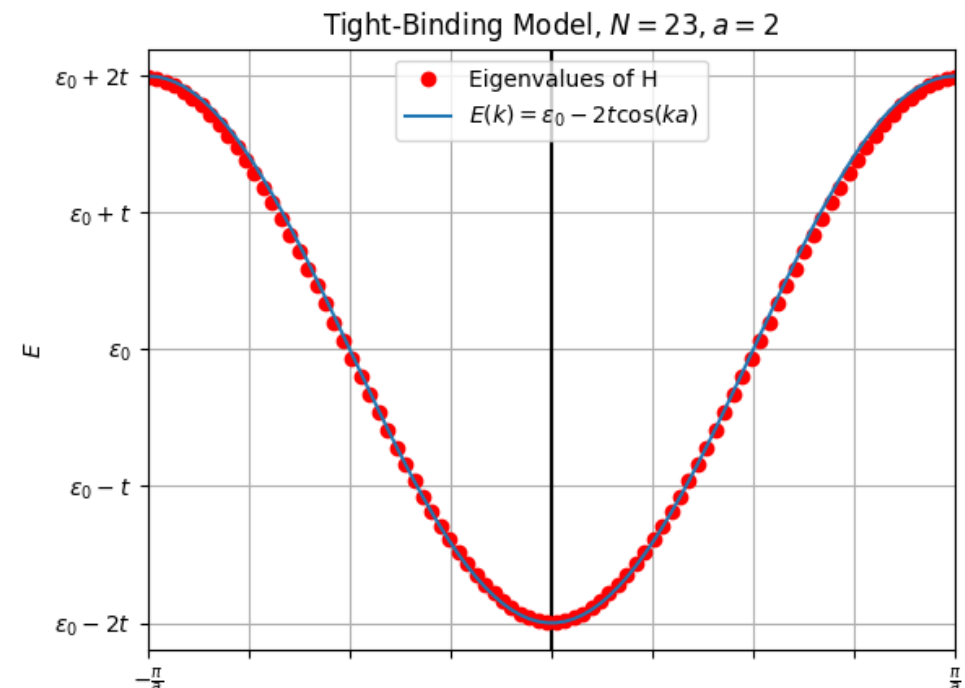
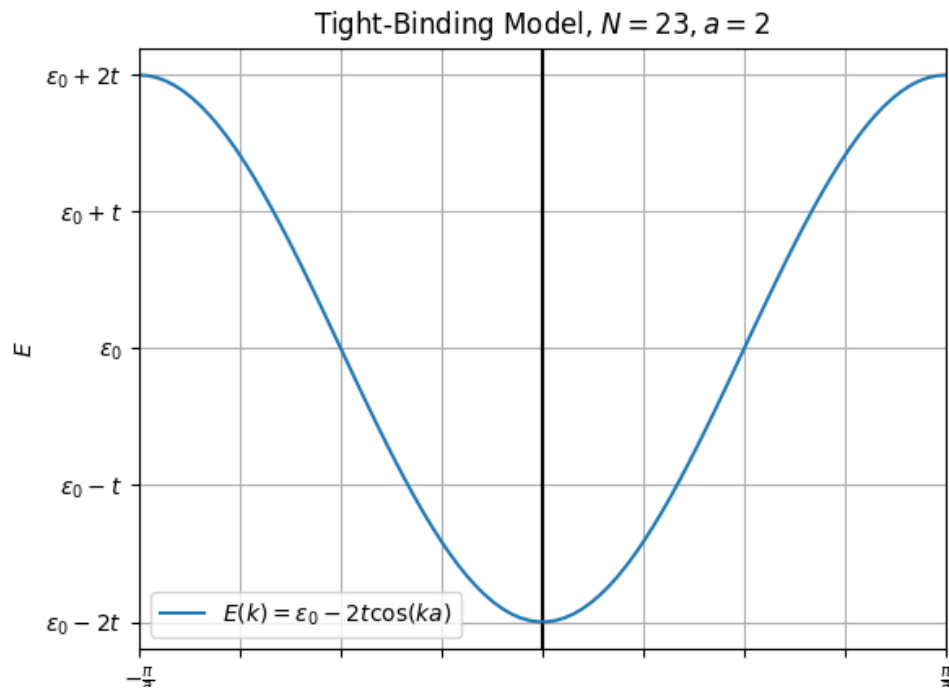
Bonding and anti-bonding states & energies



Many atoms: bands

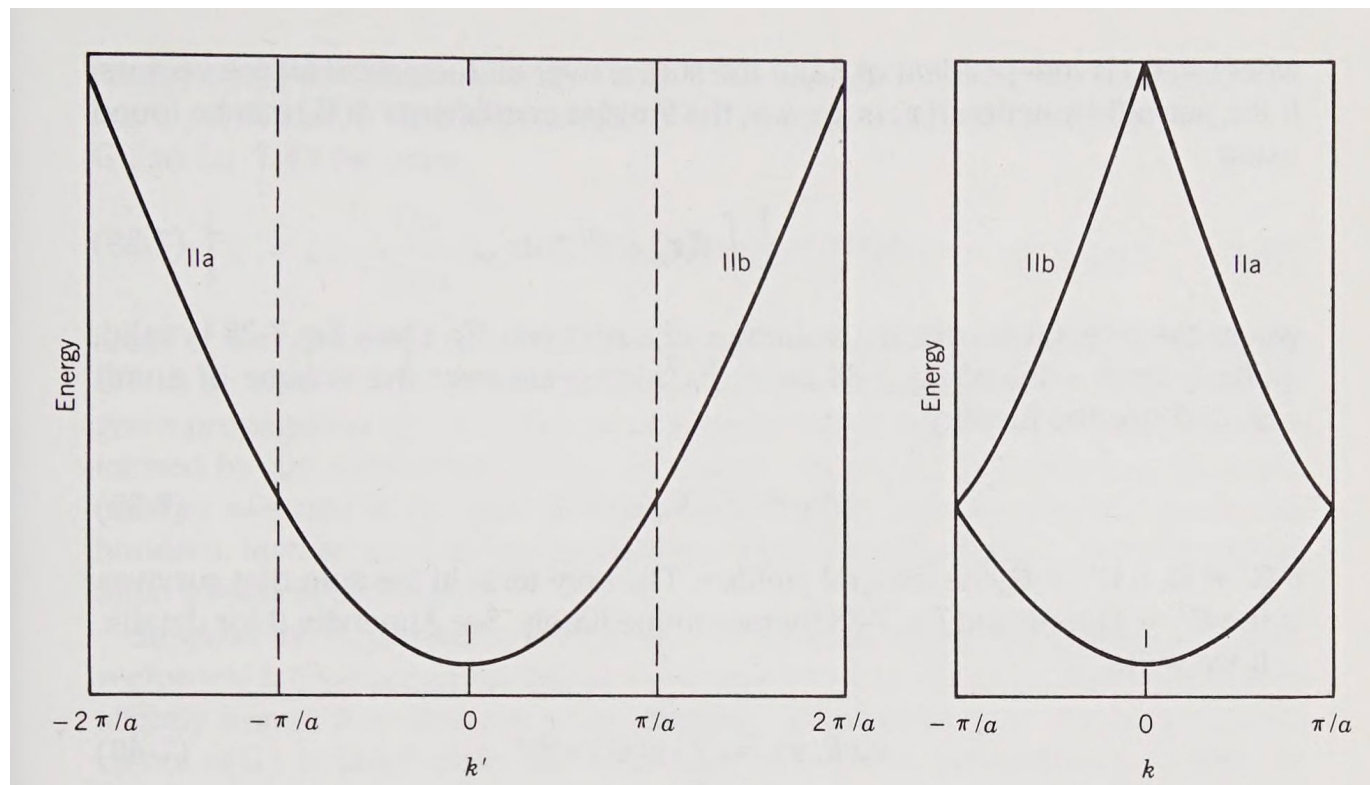


Exercise: tight binding model in 1d

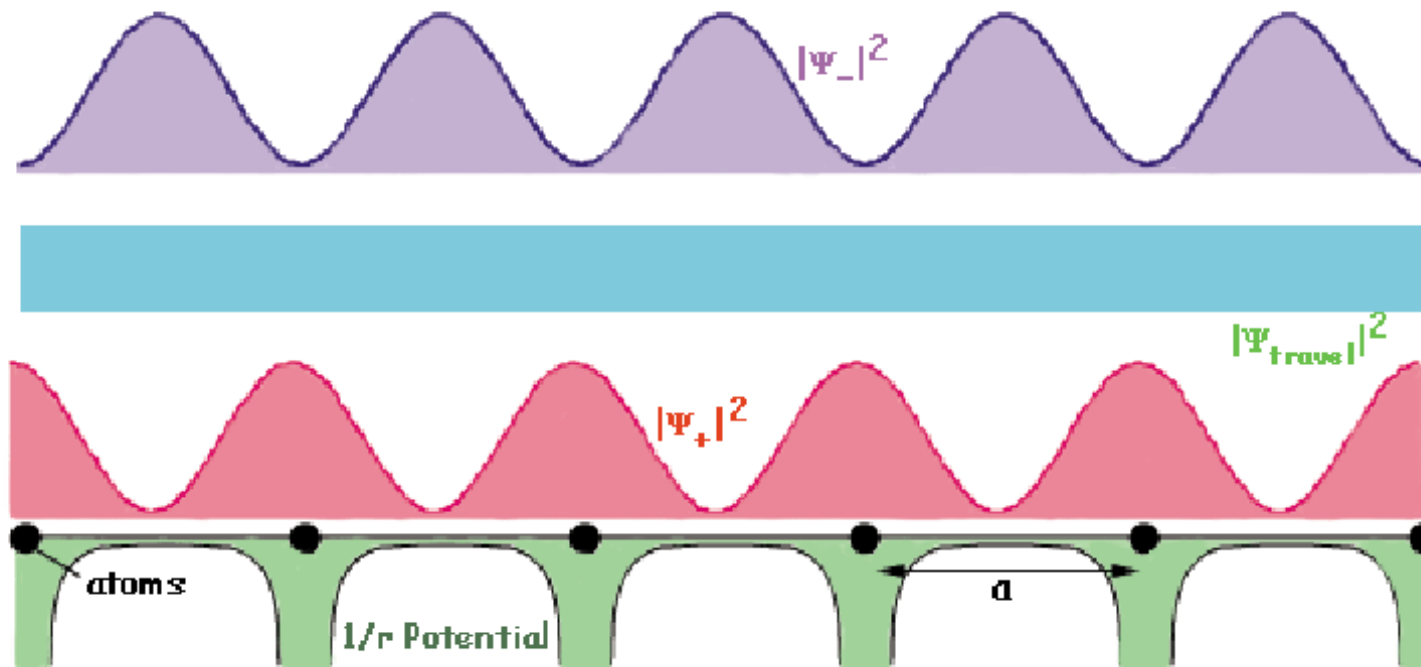


Simon, S. H. (2013). *The Oxford solid state basics*. Oxford, UK: Oxford Univ. Press.

Free electron bands in 1d



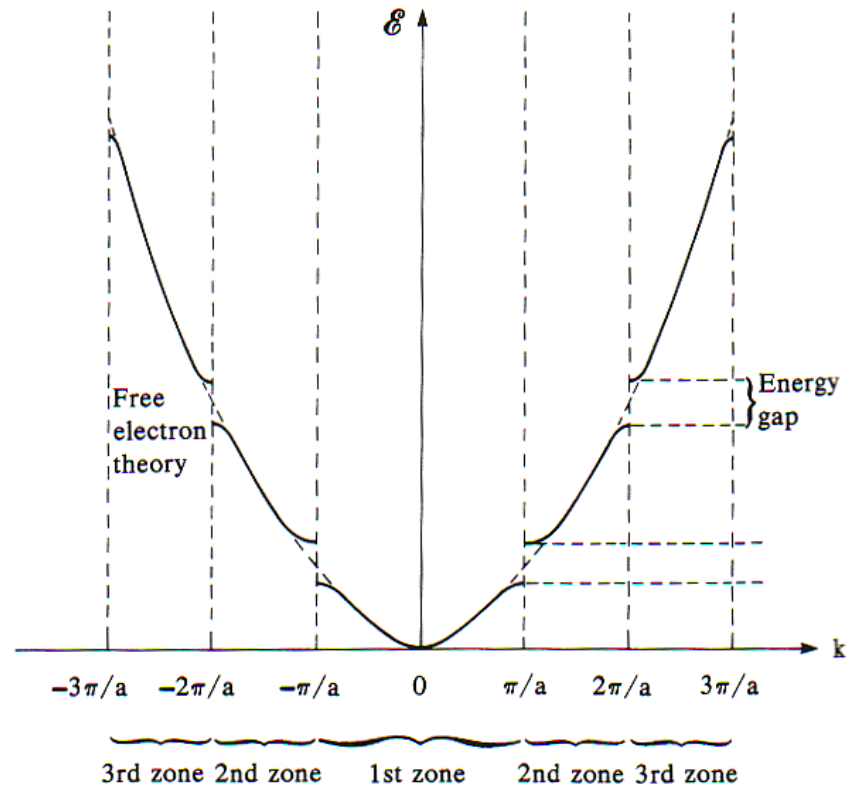
Bragg reflection of free electrons



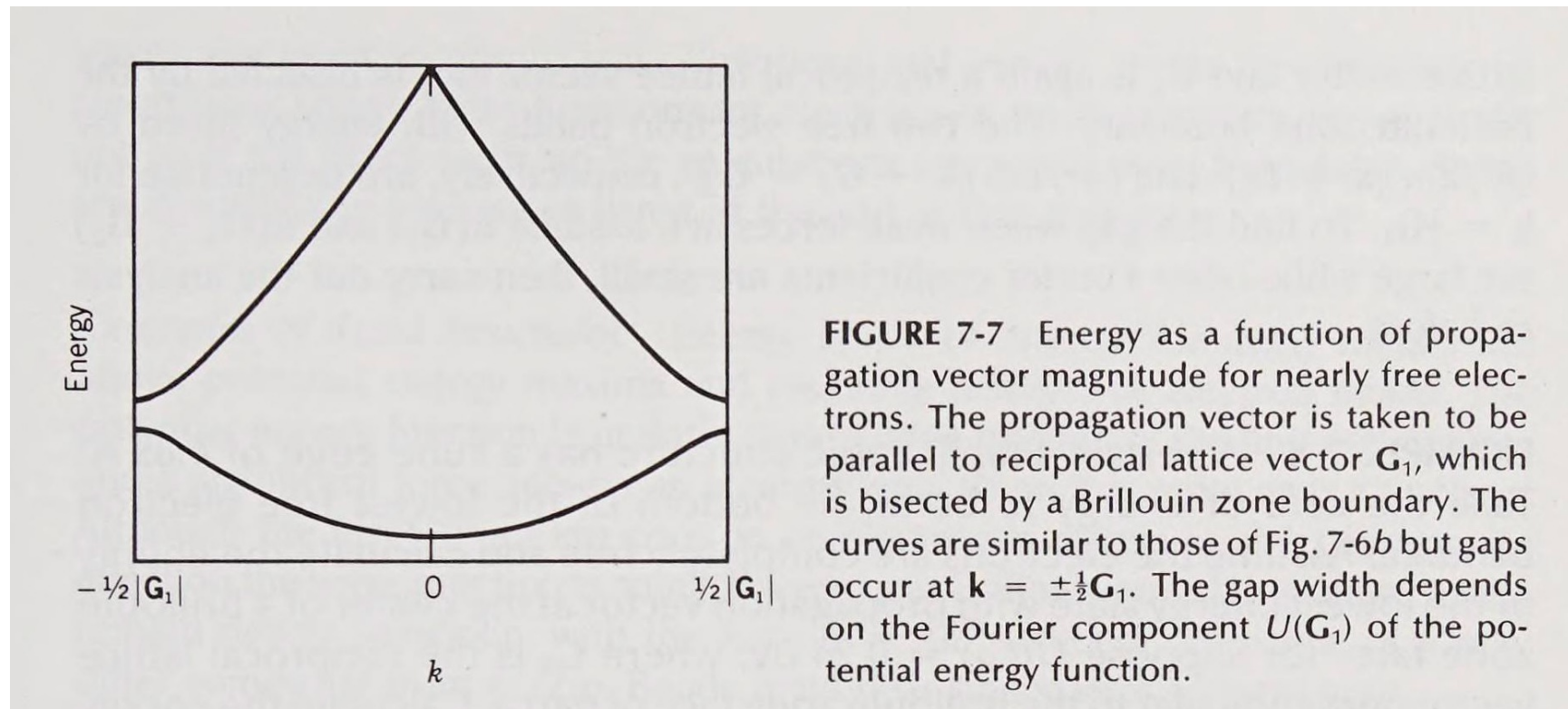
$$e^{ikx} - e^{-ikx} \\ \sim \sin(kx)$$

$$e^{ikx} + e^{-ikx} \\ \sim \cos(kx)$$

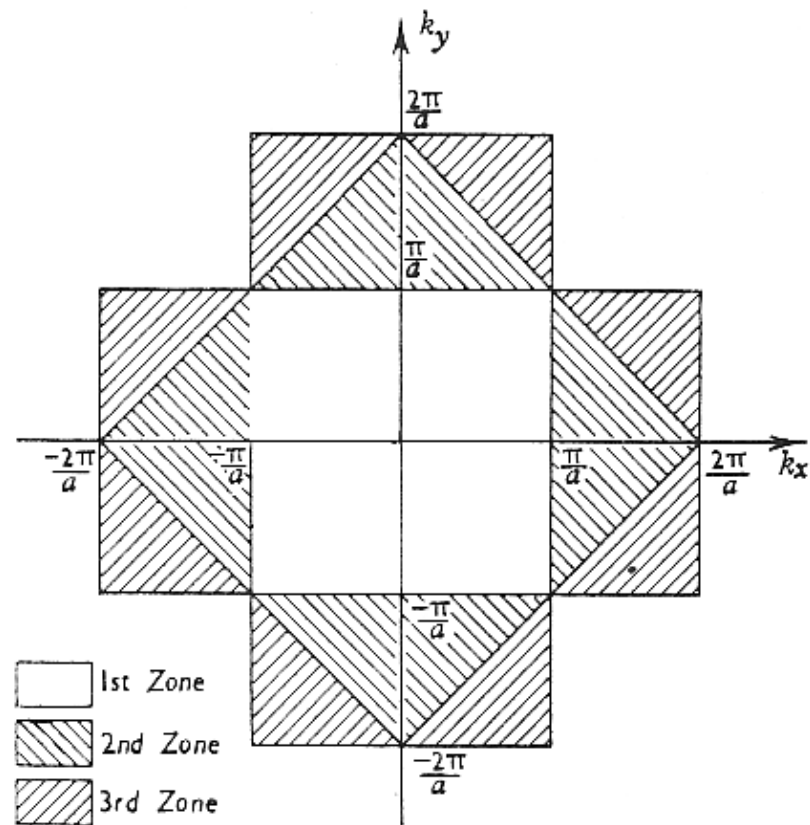
Band gaps of nearly free electrons in 1d



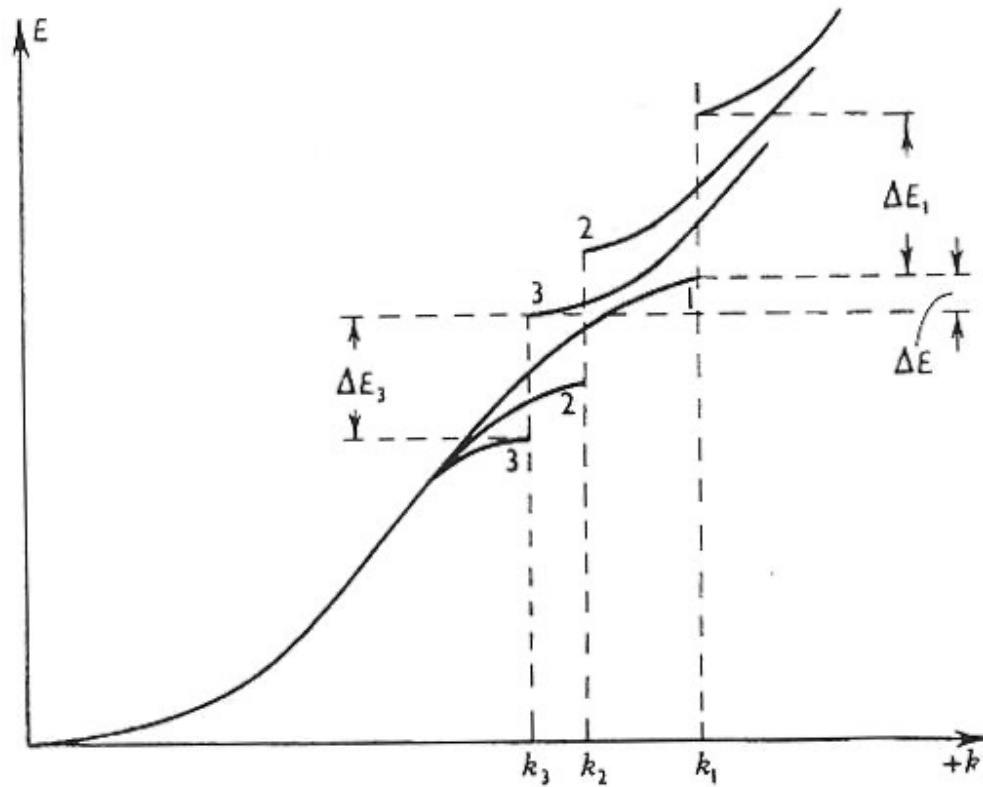
Nearly free electron bands in 1d



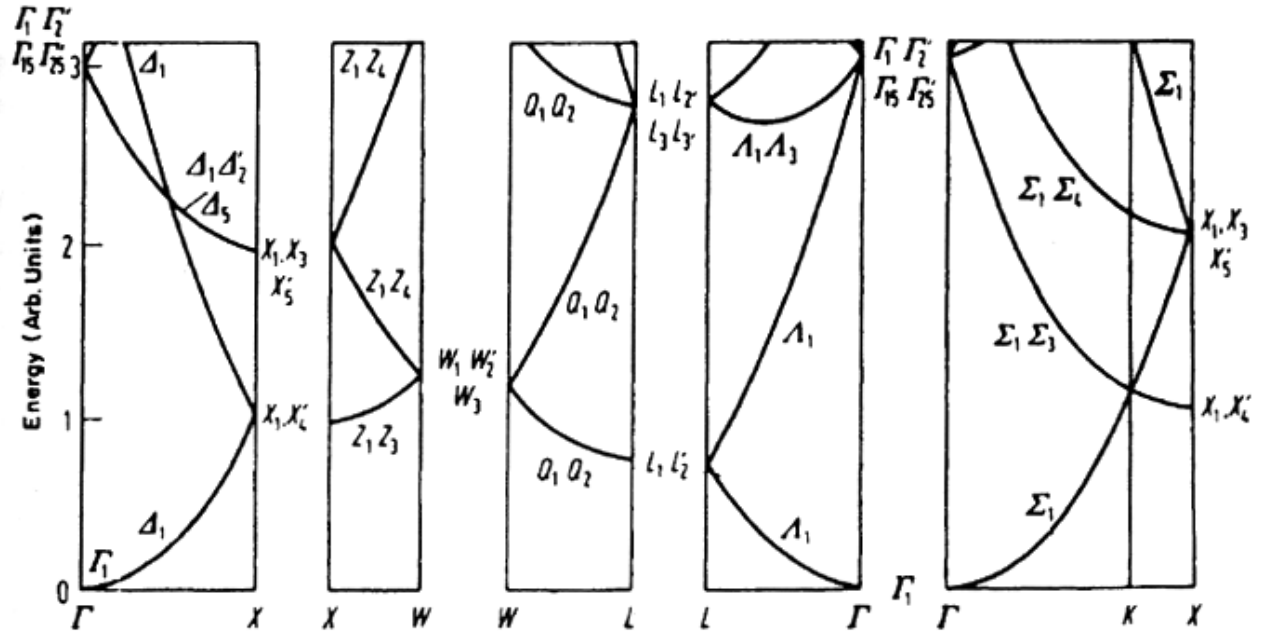
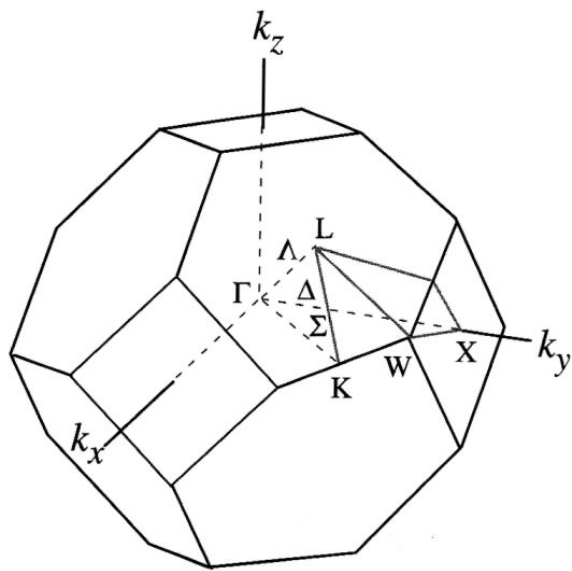
Brillouin zones in 2d



Nearly free electron bands in 2d



Free electron bands in 3d: Al



Nearly free electron bands In 3d: Al

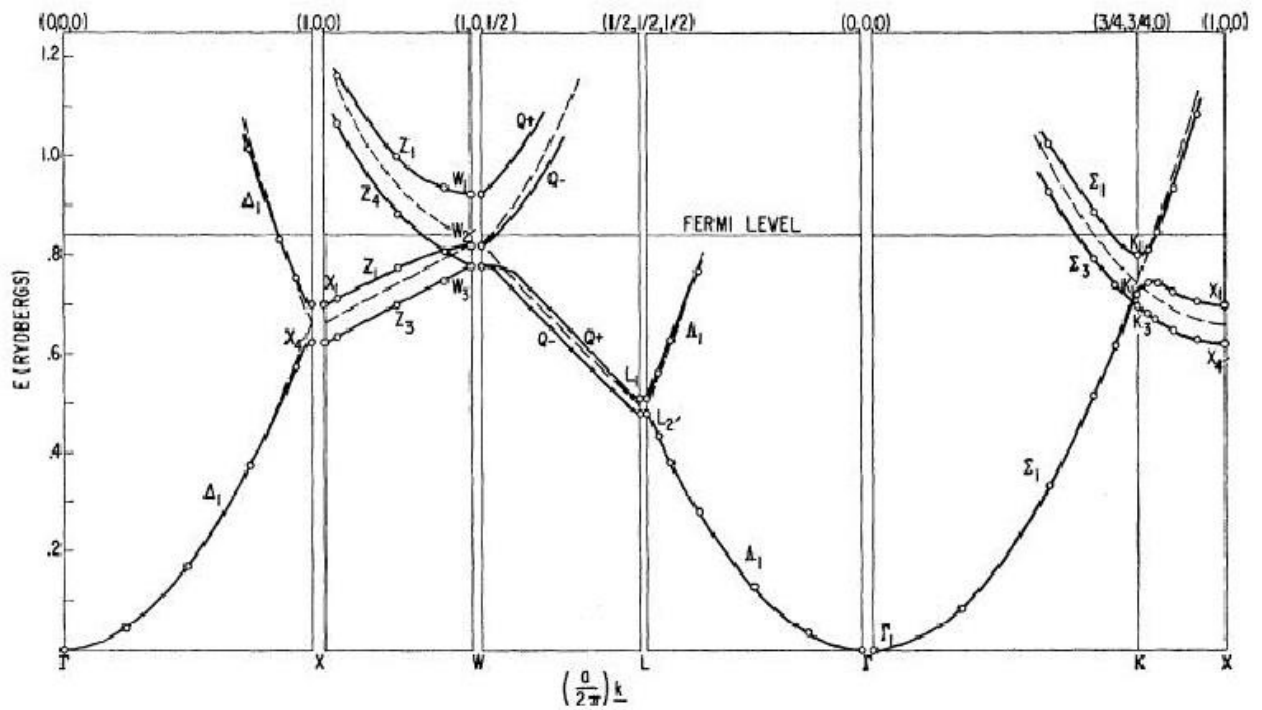
Γ - point: $(0,0,0)$

X- point: $\frac{2\pi}{a} (1,0,0)$

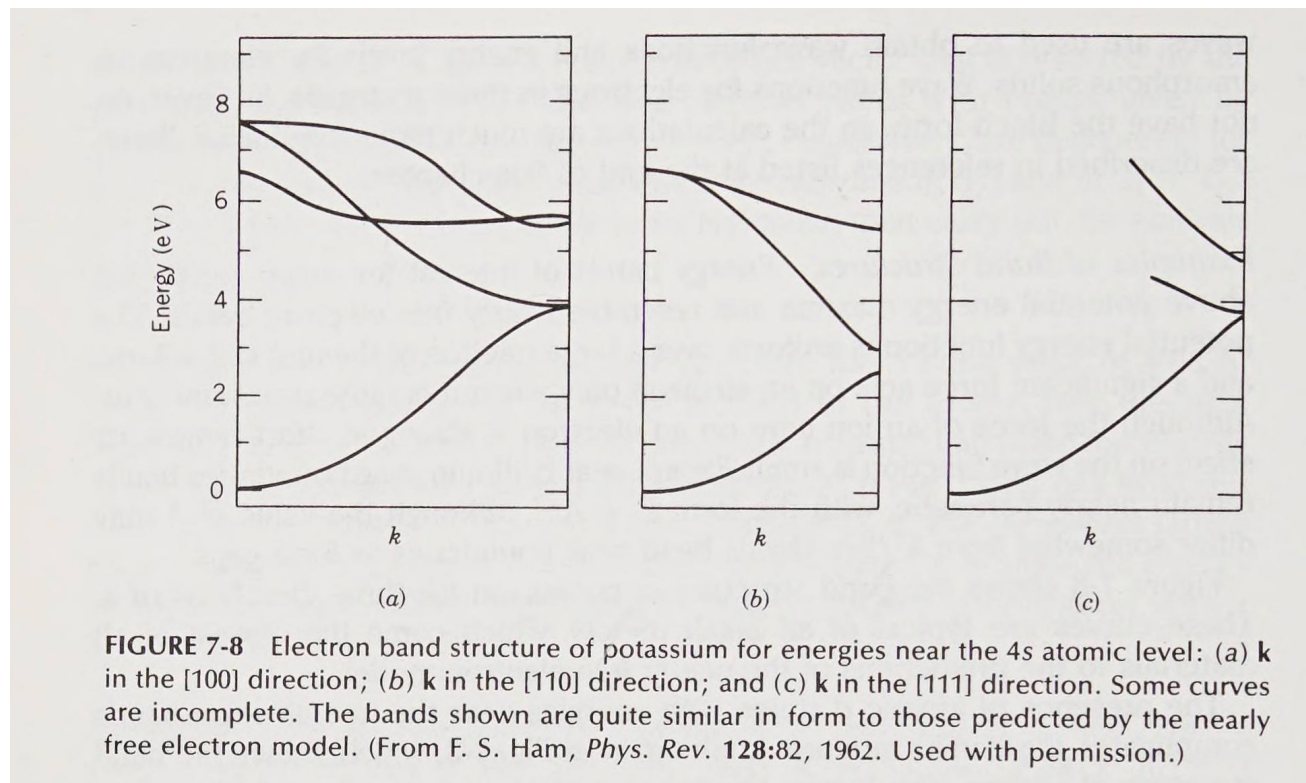
L- point: $\frac{\pi}{a} (1,1,1)$

K- point: $\frac{2\pi}{a} (\frac{3}{4}, \frac{3}{4}, 0)$

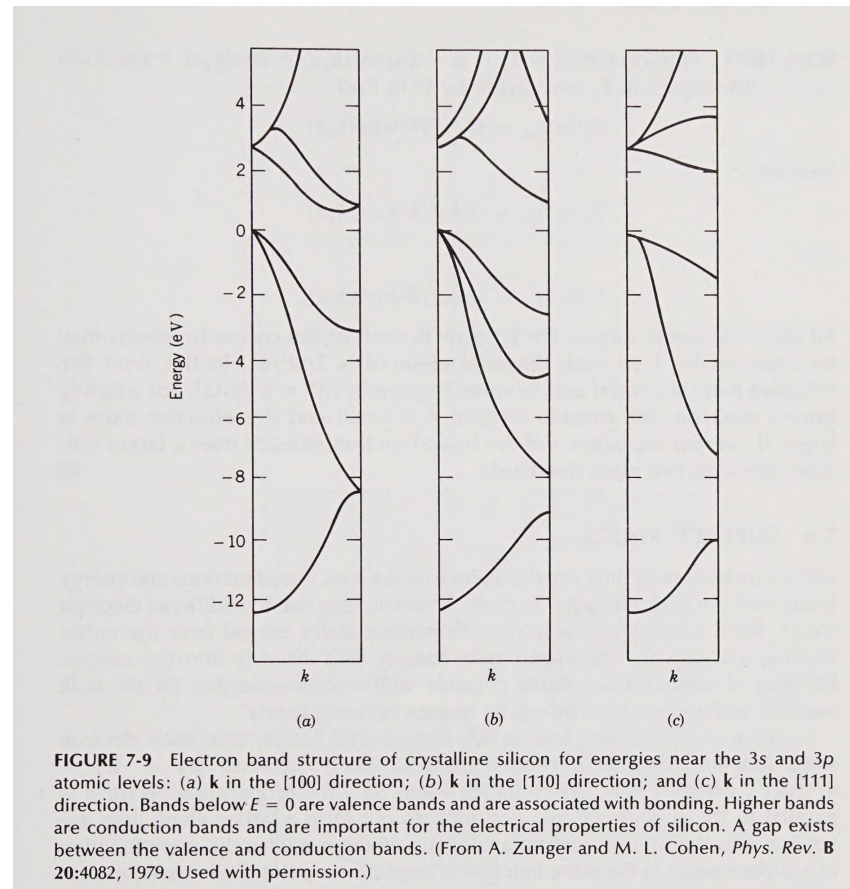
W- point: $\frac{2\pi}{a} (1, \frac{1}{2}, 0)$



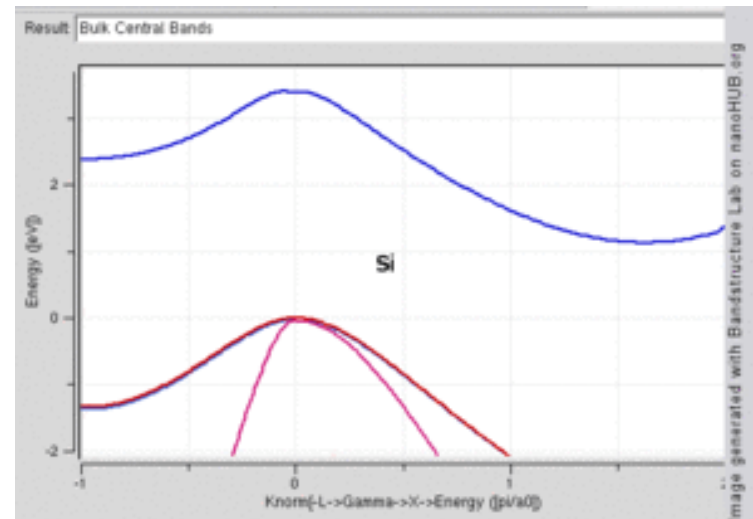
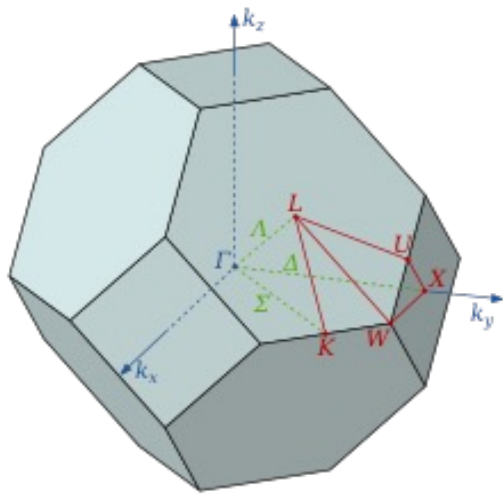
Bands of potassium in 3 directions



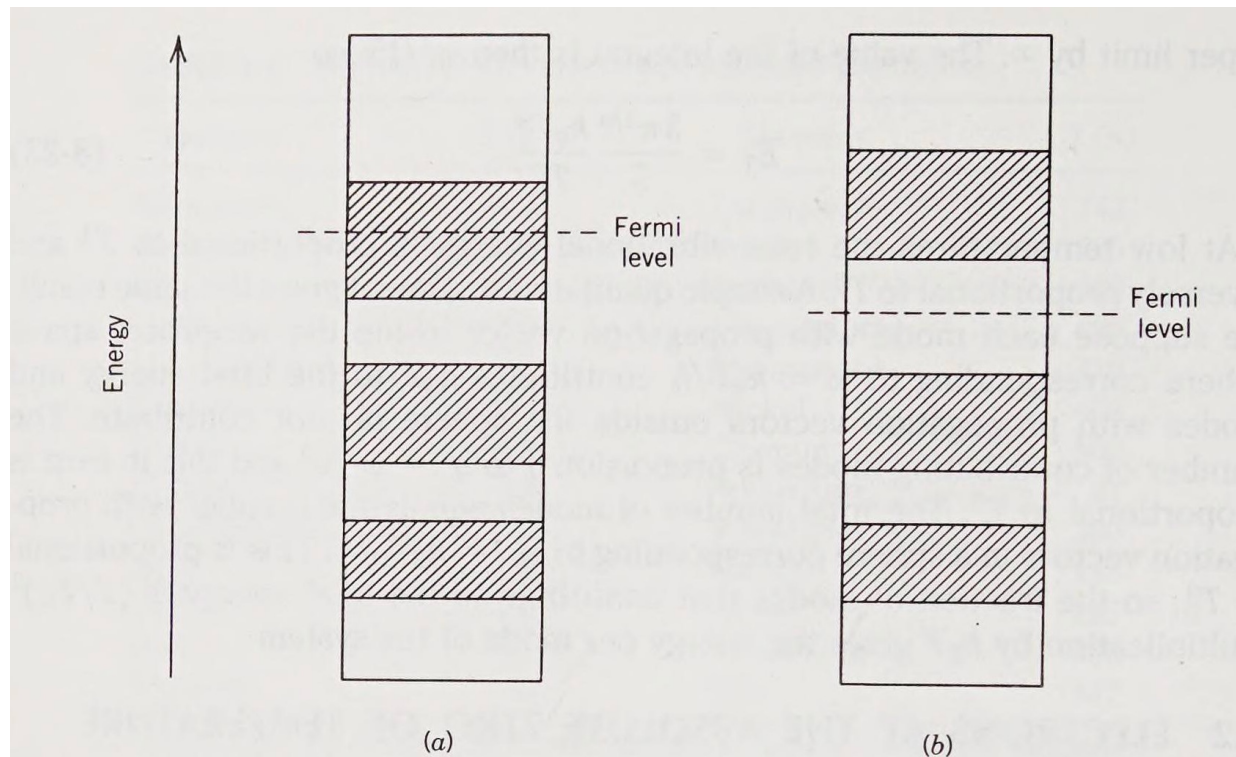
Bands of silicon in 3 directions



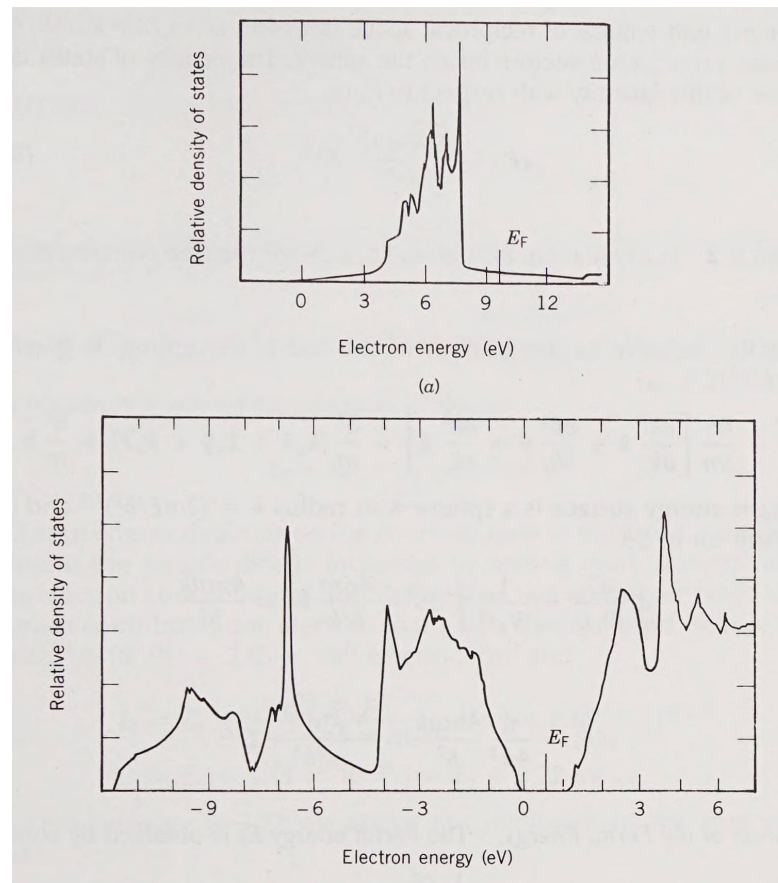
9. Termodinâmica de elétrons



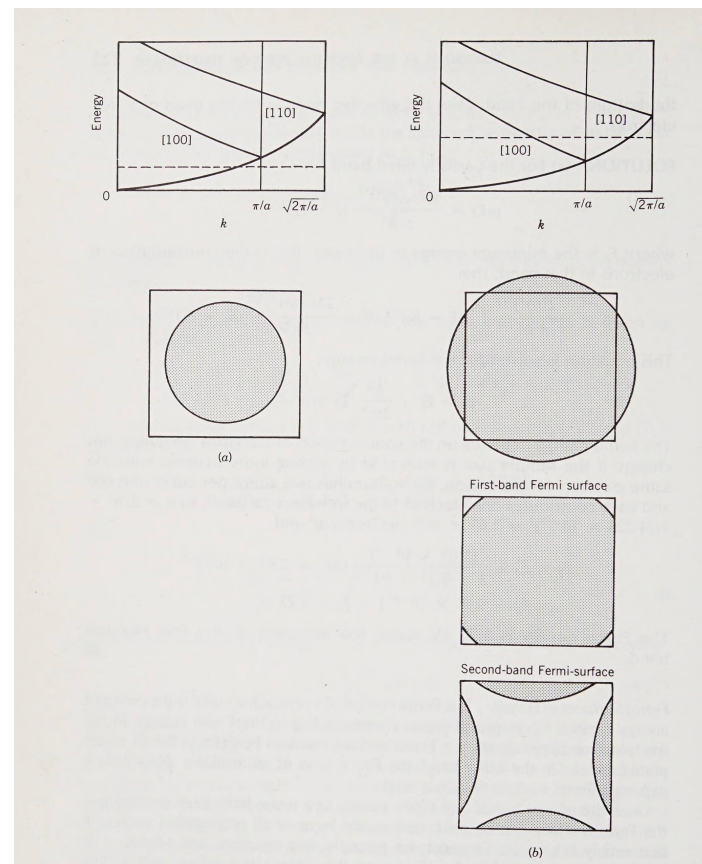
Fermi level



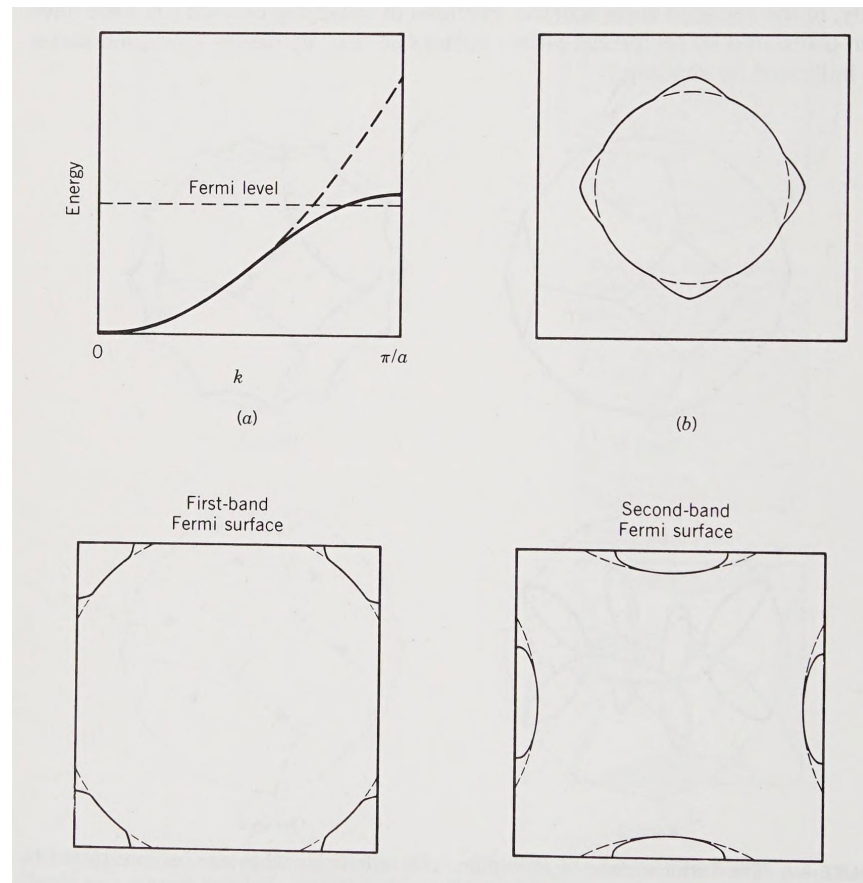
Density of states



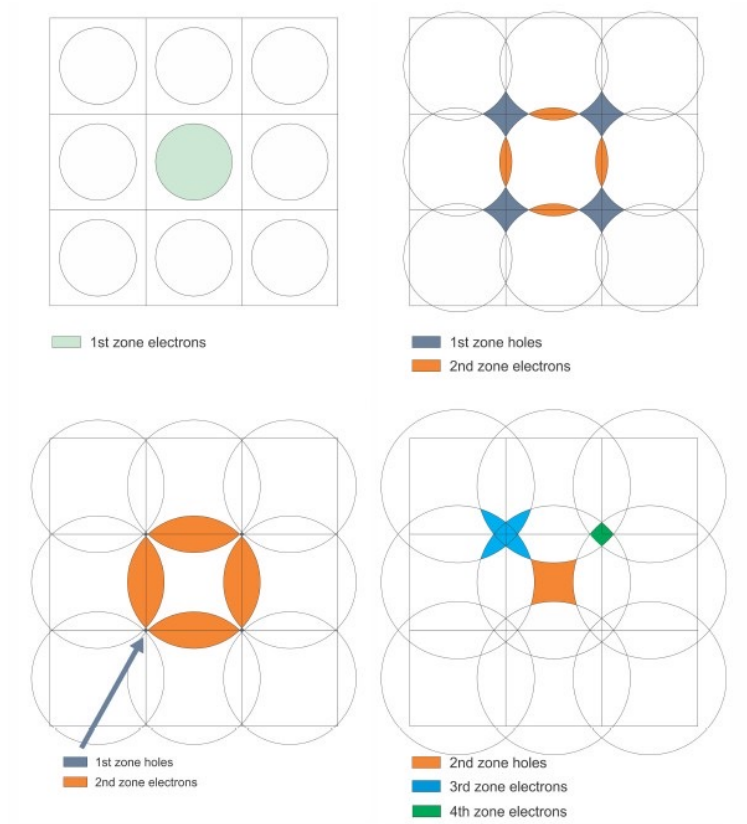
Fermi surface of free electrons in 2d (square)



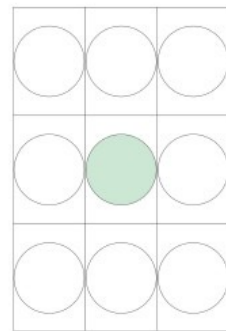
Fermi surface of nearly free electrons in 2d (square)



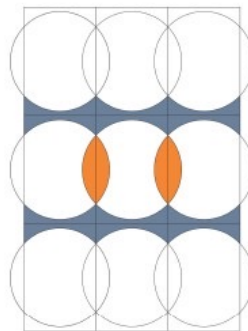
BZ of the square lattice: electrons & holes



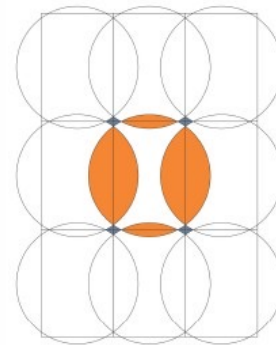
BZ of the rectangular lattice: electrons & holes



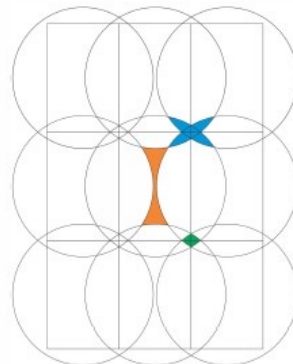
1st zone electrons



1st zone open orbits
2nd zone electrons

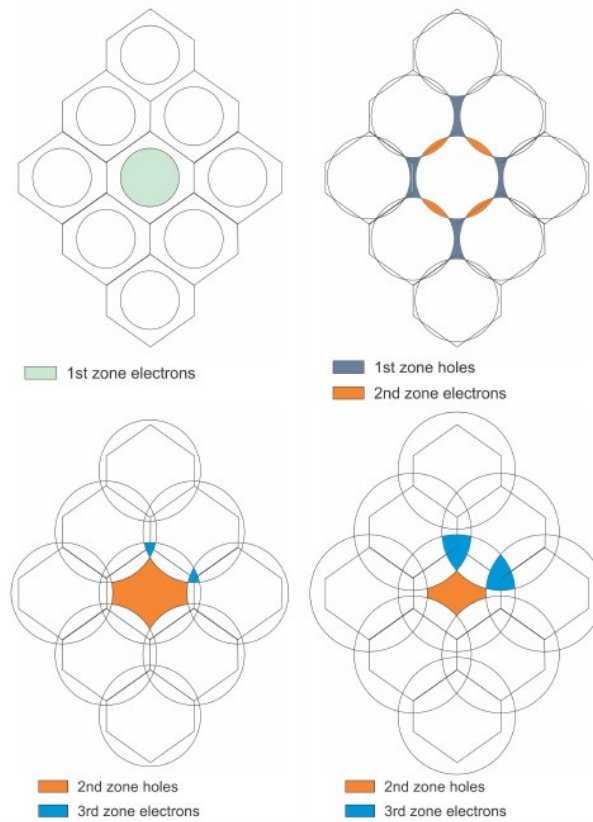


1st zone holes
2nd zone electrons

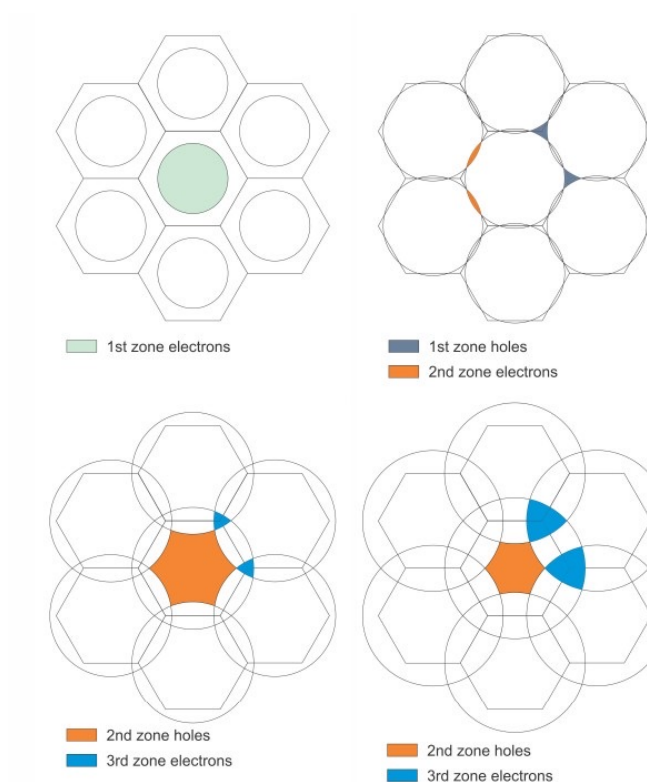


2nd zone holes
3rd zone electrons
4th zone electrons

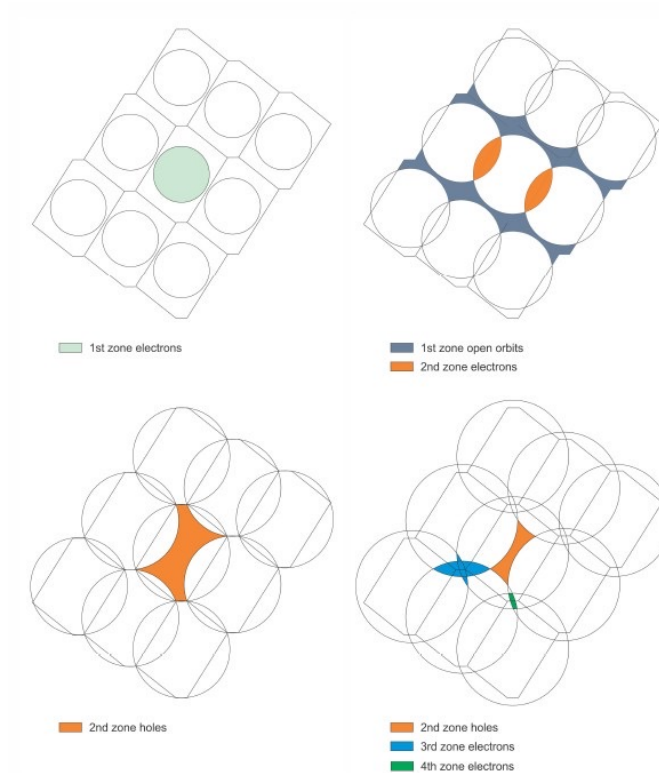
BZ of the face centred rectangular lattice: electrons & holes



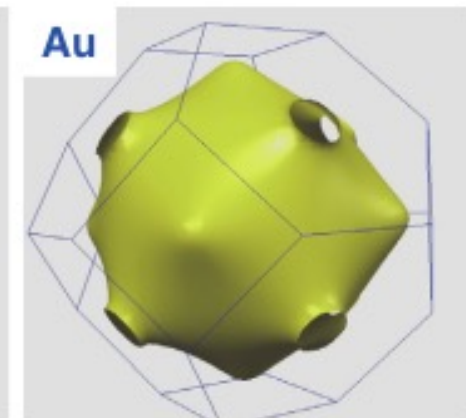
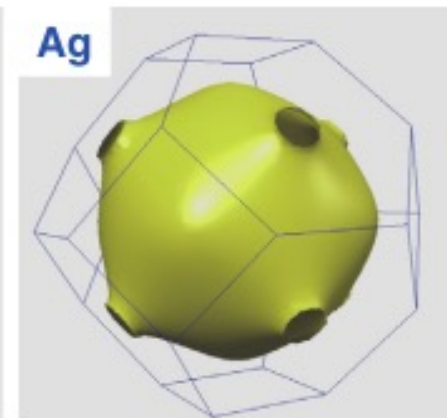
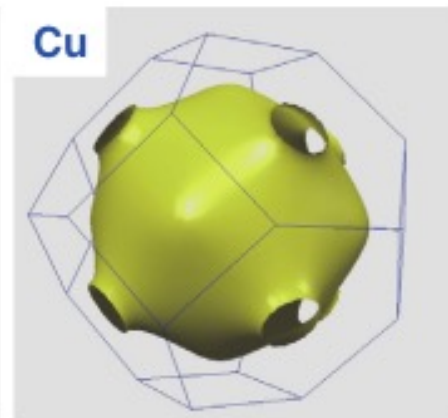
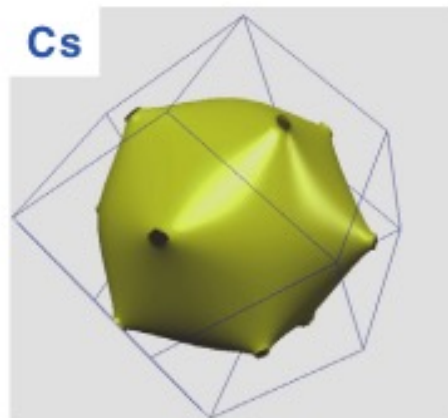
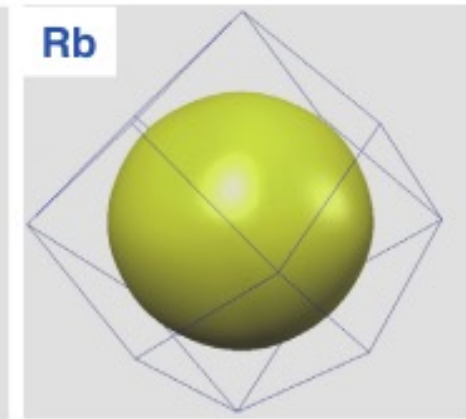
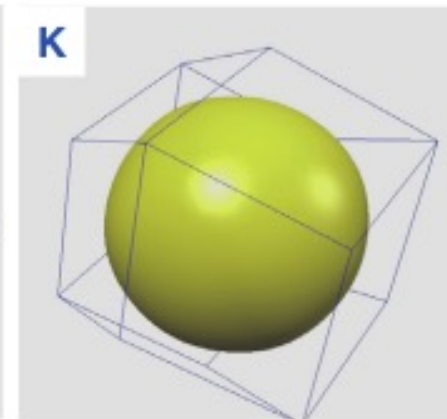
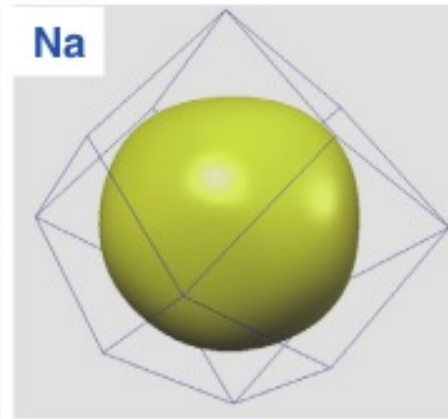
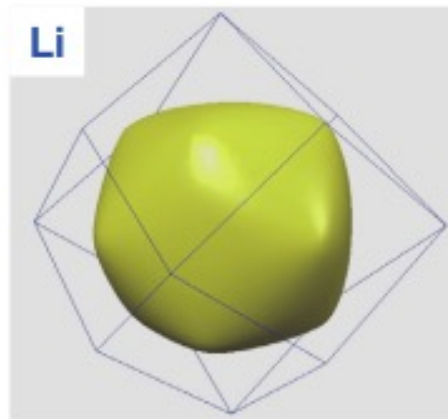
BZ of the hexagonal lattice: electrons & holes



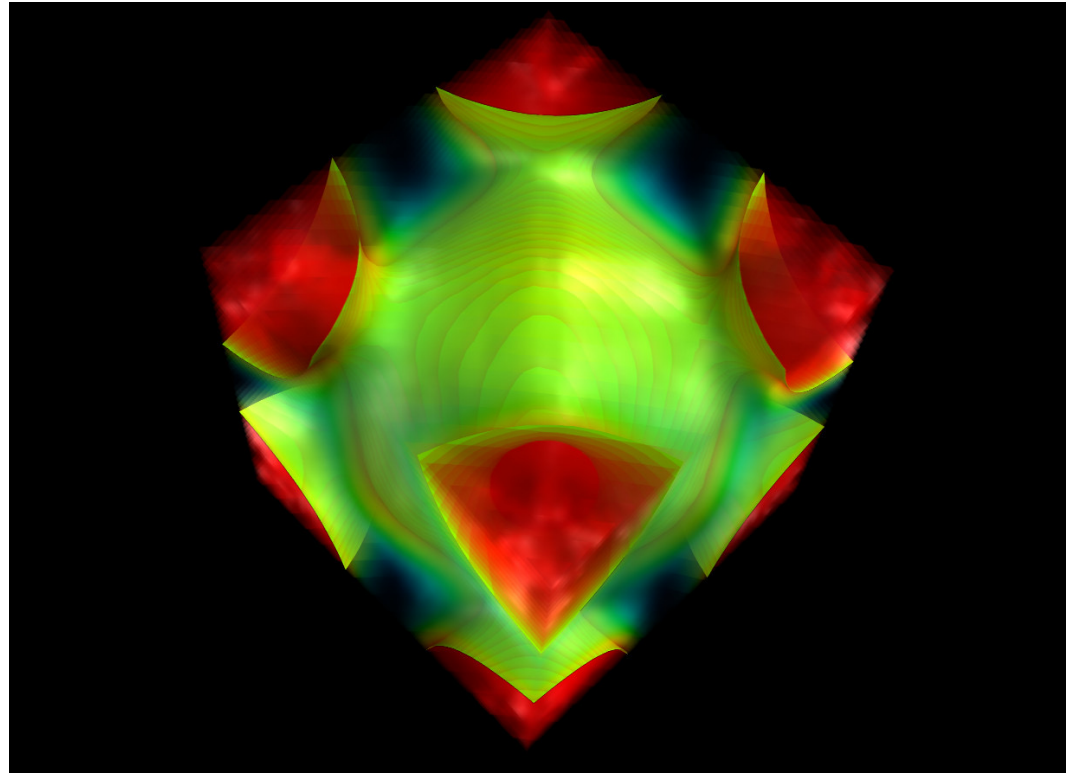
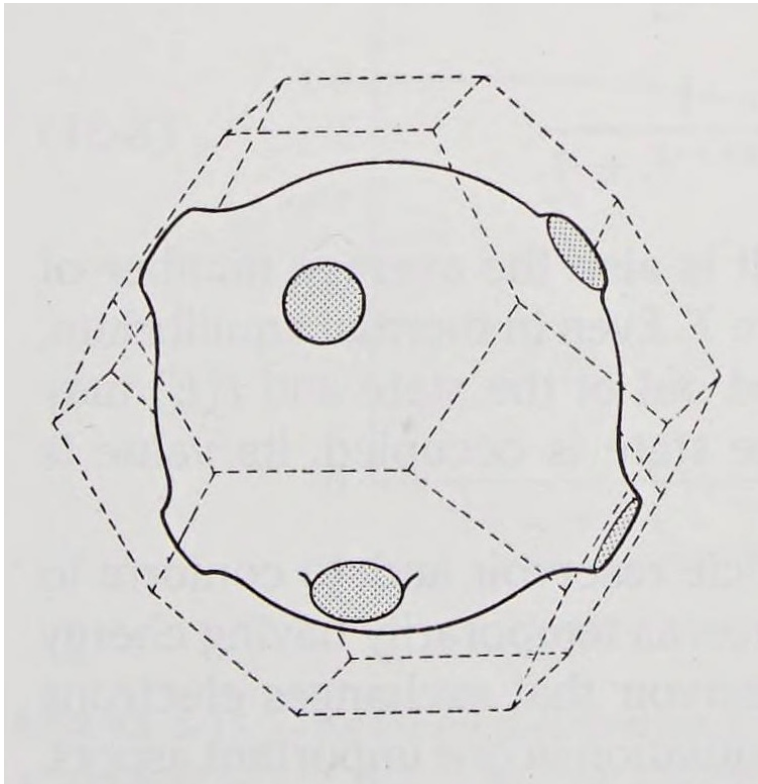
BZ of the oblique lattice: electrons & holes



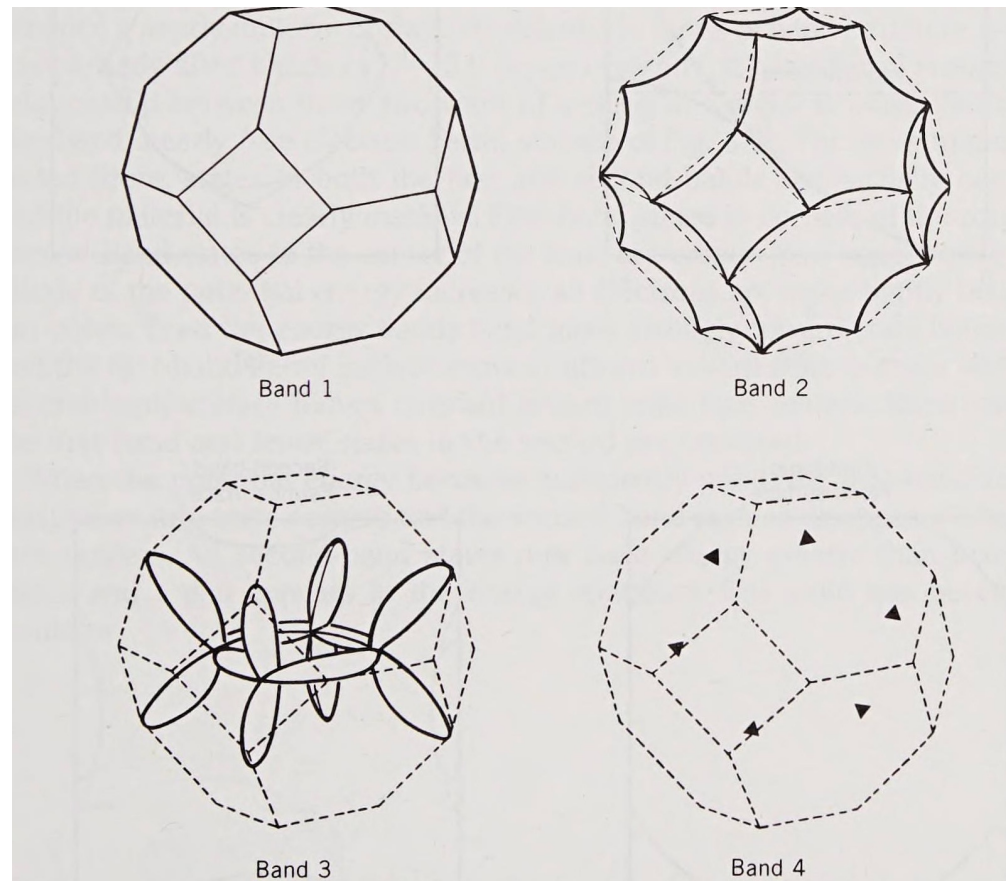
Fermi surface of alkali (bcc) and noble (fcc) metals of valence 1



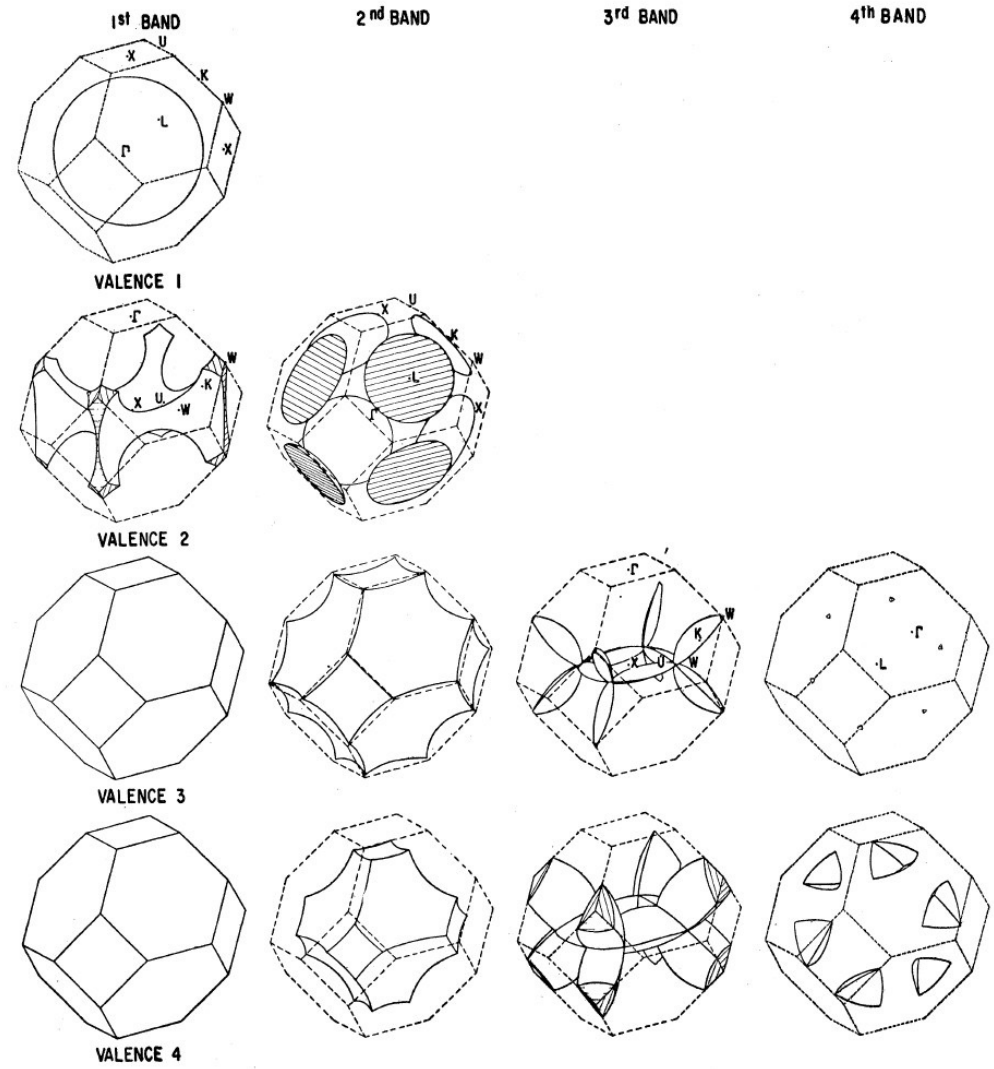
Fermi surface of Cu (fcc valence 1)



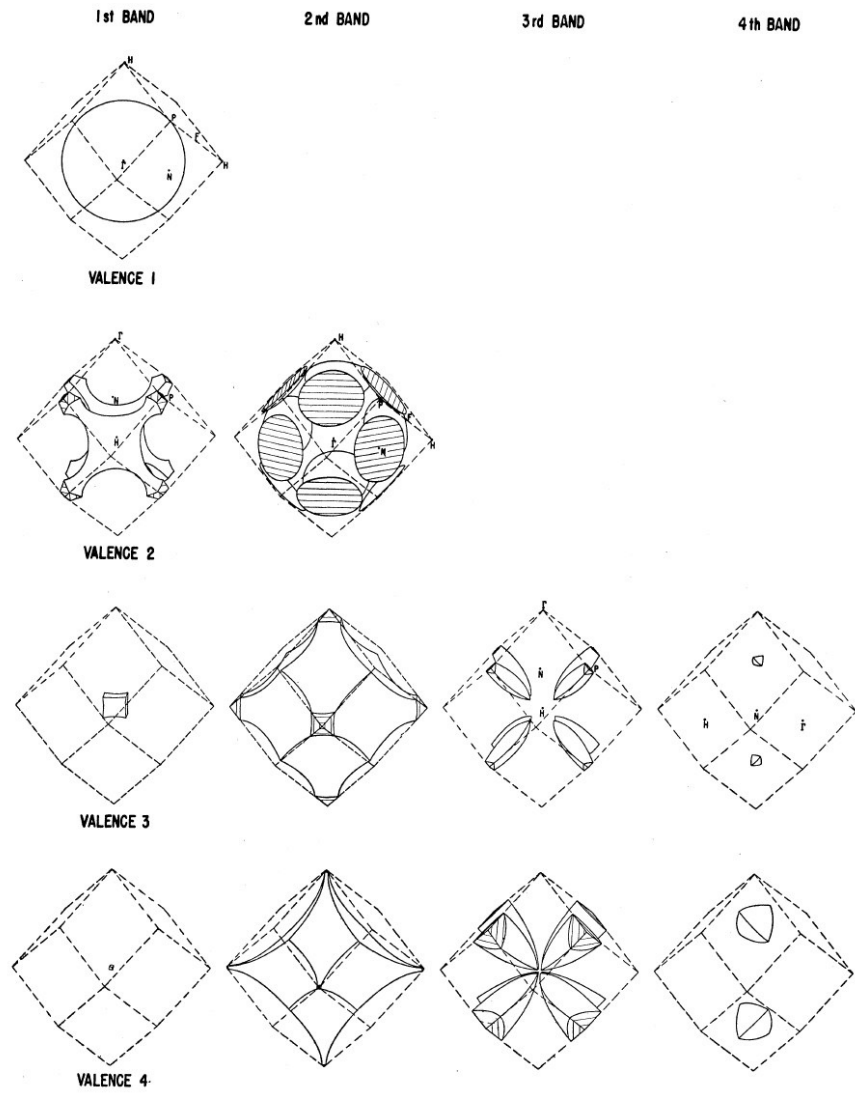
Fermi surface in 3d of Al (fcc valence 3)



Fcc lattice

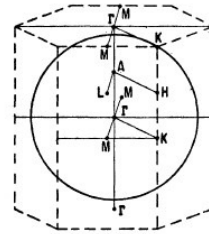


Bcc lattice

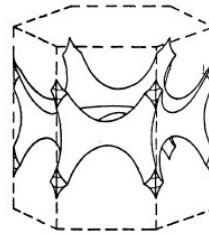


Hexagonal lattice

1st & 2nd BANDS

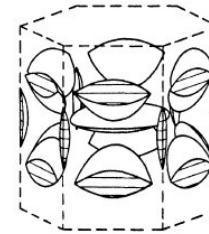


VALENCE 1

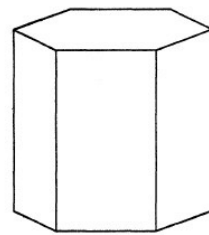


VALENCE 2

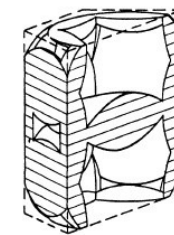
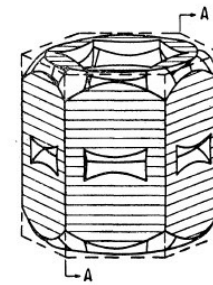
3rd & 4th BANDS



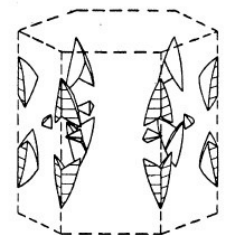
5th & 6th BANDS



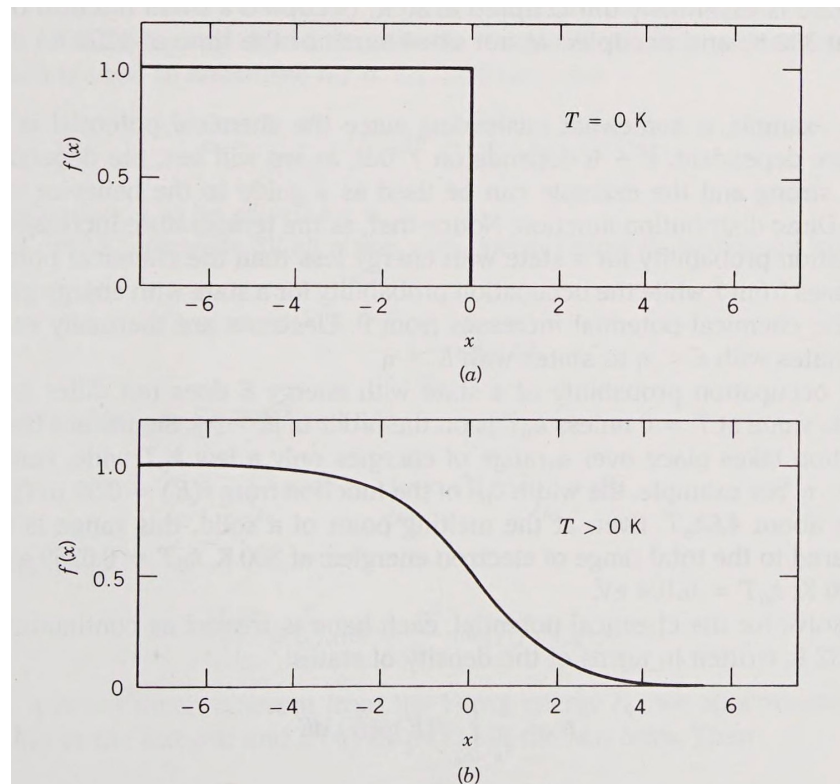
VALENCE 3



SECTION 'A-A'



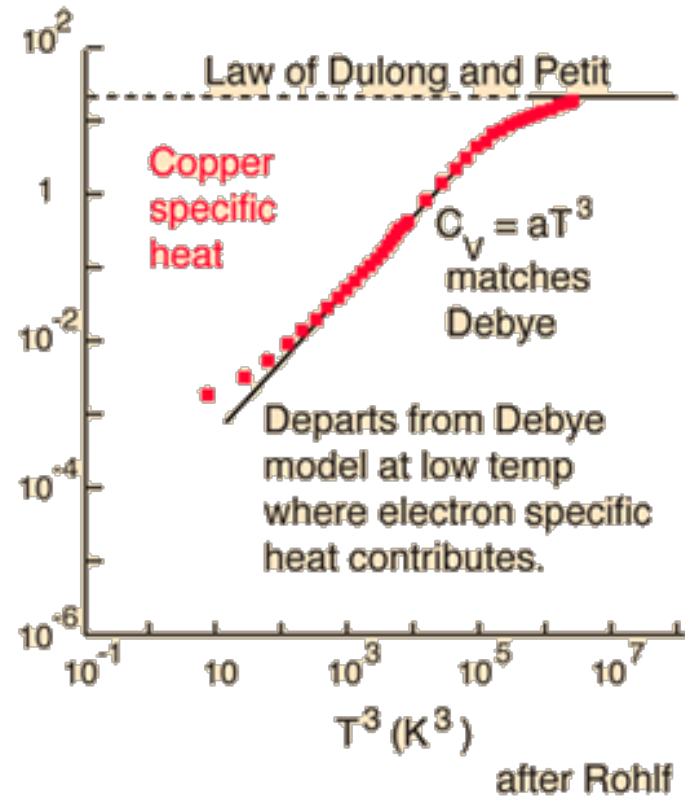
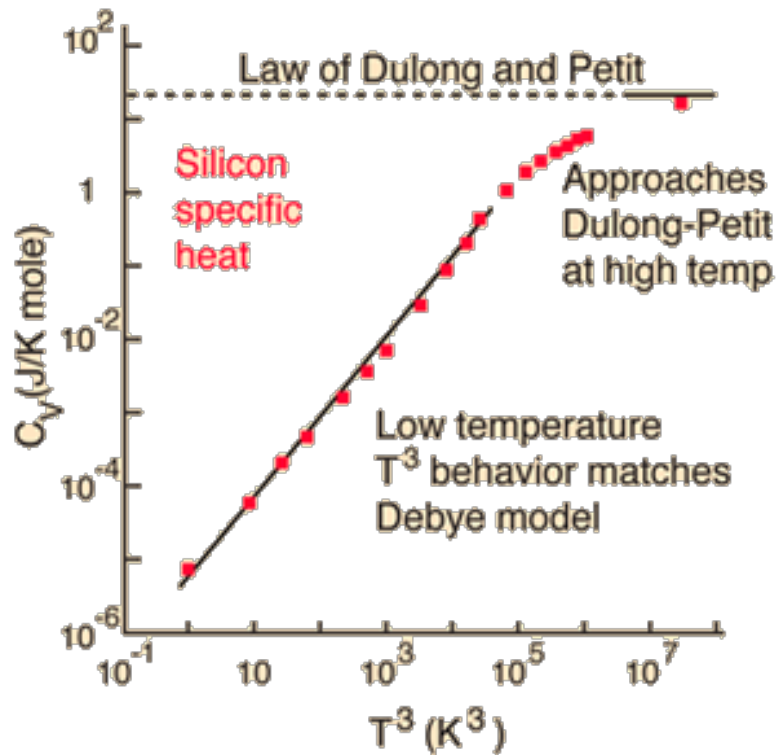
Fermi-Dirac distribution



Electronic specific heat

$$C_{\text{electrons}} = \frac{\pi^2 N_A k^2 T}{2E_F} \text{mole}^{-1}$$

Specific heat of solids



Electronic specific heat of metals

